## Memory Based Answers \& Solutions

Time : 3 hrs.

## JEE (Main)-2022 (Online) Phase-1

(Physics, Chemistry and Mathematics)

## IMPORTANT INSTRUCTIONS:

(1) The test is of $\mathbf{3}$ hours duration.
(2) The Test Booklet consists of 90 questions. The maximum marks are 300 .
(3) There are three parts in the question paper consisting of Physics, Chemistry and Mathematics having 30 questions in each part of equal weightage. Each part (subject) has two sections.
(i) Section-A: This section contains 20 multiple choice questions which have only one correct answer. Each question carries $\mathbf{4}$ marks for correct answer and -1 mark for wrong answer.
(ii) Section-B: This section contains 10 questions. In Section-B, attempt any five questions out of 10. The answer to each of the questions is a numerical value. Each question carries 4 marks for correct answer and $\mathbf{- 1}$ mark for wrong answer. For Section-B, the answer should be rounded off to the nearest integer.

## PHYSICS

## SECTION - A

Multiple Choice Questions: This section contains 20 multiple choice questions. Each question has 4 choices (1), (2), (3) and (4), out of which ONLY ONE is correct.

## Choose the correct answer:

1. Find radius ratio of proton, deuteron and alpha particle in same magnetic field having equal kinetic energy
(1) $1: 1: \sqrt{2}$
(2) $1: \sqrt{2}: 1$
(3) $1: 2: \sqrt{2}$
(4) $\sqrt{2}: 1: 2$

Answer (2)
Sol. $R=\frac{\sqrt{2 m K}}{q B}$
or $R \propto \frac{\sqrt{m}}{q}$
So, $R_{1}: R_{2}: R_{3}=\frac{\sqrt{m_{1}}}{q_{1}}: \frac{\sqrt{m_{2}}}{q_{2}}: \frac{\sqrt{m_{3}}}{q_{3}}$
$=1: \sqrt{2}: 1$
2. Two travelling wave in opposite direction create standing wave having equation. $y=10 \cos (\pi x) \sin \left(\frac{2 \pi t}{T}\right)$, where $x$ and $t$ are in $m$ and sec.
Amplitude at $x=\frac{4}{3} \mathrm{~m}$
(1) 5
(2) 10
(3) 12
(4) 11

Answer (1)
Sol. $y=10 \cos (\pi x) \sin \left(\frac{2 \pi t}{T}\right)$
$\therefore \quad \mathrm{Amp}=10 \cos (\pi x)$
at $x=\frac{4}{3} \mathrm{~m}$
Amp $=\left|10 \cos \left(\pi \times \frac{4}{3}\right)\right|$
$=10 \times\left(\frac{1}{2}\right)$
$=5$ units
3. A particle of mass 5 kg is projected vertically upward with velocity $20 \mathrm{~m} / \mathrm{s}$. If air resistance for motion is 10 N . Ratio of time of ascent to that of time of descent, is
(1) $\sqrt{2}: \sqrt{3}$
(2) $\sqrt{3}: \sqrt{2}$
(3) $2: 3$
(4) $3: 2$

Answer (1)
Sol. $a_{\text {upward }}=\left[-10+\left(-\frac{10}{5}\right)\right] \hat{j}=-12 \hat{j} \mathrm{~m} / \mathrm{s}^{2}$
$a_{\text {downwards }}=\left[-10+\frac{10}{5}\right] \hat{j}=-8 \hat{j} \mathrm{~m} / \mathrm{s}^{2}$
$d_{\text {upwards }}=d_{\text {downwards }}$
$\Rightarrow \frac{1}{2}(12) t_{\text {up }}^{2}=\frac{1}{2} 8\left(t_{\text {down }}\right)^{2}$
$\Rightarrow \frac{t_{\text {up }}}{t_{\text {down }}}=\frac{\sqrt{2}}{\sqrt{3}}$
4. A mass $m$ suspended with the help of string is moving in vertical circular motion. The tension in the string is
(1) Same through out
(2) Maximum at top
(3) Minimum at top
(4) Maximum at horizon level

## Answer (3)

Sol. $T_{\text {min }}=\frac{m v^{2}}{R}-m g$


At the top most position.
5. Parallel plate capacitor charged to 60 mC plate loses charge at $1.8 \times 10^{-8} \mathrm{C} / \mathrm{s}$. Magnitude of displacement current is
(1) $1.8 \times 10^{-8} \mathrm{C} / \mathrm{s}$
(2) $3.6 \times 10^{-8} \mathrm{C} / \mathrm{s}$
(3) $5 \times 10^{-8} \mathrm{C} / \mathrm{s}$
(4) $6 \times 10^{-8} \mathrm{C} / \mathrm{s}$

Answer (1)

Sol. In a parallel plate capacitor discharging

$$
\begin{aligned}
& I=I_{D} \\
& \Rightarrow \quad I_{D}=\frac{d q}{d t}
\end{aligned}
$$

$$
=1.8 \times 10^{-8} \mathrm{C} / \mathrm{s}
$$

$$
\Rightarrow I_{D}=1.8 \times 10^{-8} \mathrm{~A}
$$

6. A charged particle $(-q, m)$ revolves around a cylinder of charge density $\rho$ and radius $R$. The kinetic energy of particle is
(1) $\frac{\rho R^{2} q}{4 \varepsilon_{0}}$
(2) $\frac{\rho R^{2} q}{2 \varepsilon_{0}}$
(3) $\frac{\rho R^{2} q}{8 \varepsilon_{0}}$
(4) $\frac{\rho R^{2} q}{16 \varepsilon_{0}}$

## Answer (1)

Sol. $E=\frac{\lambda}{2 \pi \varepsilon_{0} \times r}$

$$
\begin{aligned}
& =\frac{\rho \times \pi R^{2} \times l}{I \times 2 \pi \varepsilon_{0} r} \\
& =\frac{\rho R^{2}}{2 \varepsilon_{0} r}
\end{aligned}
$$

$\therefore K E=\frac{1}{2} m v^{2}$

$$
=\frac{1}{2}\left(\frac{m v^{2}}{r}\right) \times r
$$

$$
=\frac{1}{2} \times\left(q \times \frac{\rho R^{2}}{2 \varepsilon_{0} r}\right) \times r
$$

$$
=\left(\frac{\rho R^{2} q}{4 \varepsilon_{0}}\right)
$$

7. If the ratio of intensities of two waves is $\frac{9}{4}$, then the ratio of maximum to minimum intensity, is
(1) $20: 1$
(2) $25: 1$
(3) $16: 1$
(4) $64: 1$

Answer (2)
Sol. $\frac{l_{1}}{l_{2}}=\frac{9}{4}$

$$
\begin{aligned}
& \Rightarrow \frac{A_{1}}{A_{2}}=\frac{3}{2} \text { or } \frac{A_{1}+A_{2}}{A_{2}-A_{2}}=\frac{5}{1} \\
& \Rightarrow \frac{I_{\text {max }}}{I_{\text {min }}}=\frac{\left(A_{1}+A_{2}\right)^{2}}{\left(A_{1}-A_{2}\right)^{2}}=\frac{25}{1}
\end{aligned}
$$

8. Potential energy of two atoms is $U=\frac{A}{r^{10}}-\frac{B}{r^{5}}$. Find $r$ for which potential energy is minimum.
(1) $2\left(\frac{A}{B}\right)^{\frac{1}{5}}$
(2) $\left(\frac{A}{B}\right)^{\frac{1}{5}}$
(3) $\left(\frac{2 A}{B}\right)^{\frac{1}{5}}$
(4) $\left(\frac{A}{2 B}\right)^{\frac{1}{5}}$

## Answer (3)

Sol. $U=\frac{A}{r^{10}}-\frac{B}{r^{5}}$
If $r$ is position for $U$ (potential energy) to be minimum then at $r$
$\left.\frac{d U}{d r}\right|_{\text {at } r}=0$
or $\frac{10 A}{r^{11}}=\frac{5 B}{r^{6}}$
$\Rightarrow r=\left(\frac{2 A}{B}\right)^{\frac{1}{5}}$
9.


Two point charge particles with equal charge $Q=10^{-5} \mathrm{C}$ are kept on a table top. The particles have mass $m=10 \mathrm{gm}$ each and coefficient of friction between particles and table is 0.25 . If particle are held at equilibrium so that $/$ is equal to its minimum value then / is
(1) 2 m
(2) 6 m
(3) 10 m
(4) 21 m

## Answer (2)

Sol. $\frac{k Q^{2}}{l^{2}}=\mu m g$

$$
\begin{aligned}
& I=\sqrt{\frac{9 \times 10^{9} \times 10^{-10}}{0.25 \times 10^{-2} \times 10}} \\
& =\frac{3}{0.5}=6 \mathrm{~m}
\end{aligned}
$$

10. For which combination of $2,4,6 \Omega$ resistances equivalent resistance is $\frac{22}{3} \Omega$
(1)

(2)

(3)

(4)


## Answer (1)

Sol. In $1^{\text {st }}$ combination $R_{e q}=\left(\frac{4 \times 2}{4+2}+6\right) \Omega=\frac{22}{3} \Omega$
In $2^{\text {nd }}$ combination $R_{e q}=\left(\frac{6 \times 2}{6+2}+4\right) \Omega=\frac{11}{2} \Omega$
In $3^{\text {rd }}$ combination $R_{\text {eq }}=\left(\frac{6 \times 4}{6+4}+2\right) \Omega=\frac{22}{5} \Omega$
In $4^{\text {th }}$ combination $R_{e q}=(2+4+6) \Omega=12 \Omega$
11. 1.5 kg hammer strikes the nail of mass 100 gm with speed of $60 \mathrm{~m} / \mathrm{sec}$. If heat given to nail is $1 / 4^{\text {th }}$ of energy of hammer, find change in temperature of nail ( $S_{\text {nail }}=2250 \mathrm{~J} / \mathrm{kg}^{\circ} \mathrm{C}$ )
(1) $1.5^{\circ} \mathrm{C}$
(2) $2^{\circ} \mathrm{C}$
(3) $2.5^{\circ} \mathrm{C}$
(4) $3^{\circ} \mathrm{C}$

## Answer (4)

Sol. $E=\frac{1}{2} \times(1.5) \times(60)^{2}$
$\therefore \quad Q=\frac{1}{4} \times E$
$\Rightarrow \quad m \times s \times \Delta T=\frac{1}{4} \times \frac{1}{2} \times 1.5 \times(3600)$
$\Rightarrow \quad 0.1 \times s \times \Delta T=900 \times 1.5 \times \frac{1}{2}$
$\Rightarrow \Delta T=\frac{450 \times 1.5}{0.1 \times s}=\frac{6750}{s}$

$$
=\frac{6750}{2250}
$$

$$
=3^{\circ} \mathrm{C}
$$

12. A bulb of power rating 200 W has efficiency of $3.5 \%$. What is the peak value of magnetic field at a distance of 4 m ?
(1) $7.5 \times 10^{-8} \mathrm{~T}$
(2) $1.1 \times 10^{-8} \mathrm{~T}$
(3) $1.5 \times 10^{-8} \mathrm{~T}$
(4) $1.7 \times 10^{-8} \mathrm{~T}$

Answer (4)
Sol. $P_{\text {eff }}=200 \times\left(\frac{3.5}{100}\right)=7 \mathrm{~W}$

$$
\begin{aligned}
& I=\frac{P_{\text {eff }}}{4 \pi r^{2}}=\frac{B_{0}^{2}}{2 \mu_{0}} \times C \\
& \begin{aligned}
\Rightarrow B_{0} & =\sqrt{2 \mu_{0} \times \frac{P_{\text {eff }}}{4 \pi r^{2}} \times \frac{1}{C}} \\
& =\sqrt{\frac{2 \times 10^{-7} \times 7}{4^{2} \times 3 \times 10^{8}}} \\
& =1.7 \times 10^{-8}
\end{aligned}
\end{aligned}
$$

13. Two massless springs with spring constants $2 K$ and $9 K$ having 50 g and 100 g attached at free end, both have same $V_{\text {max. }}$. Then find ratio of amplitude of vibrations.
(1) $2: 3$
(2) $1: 2$
(3) $3: 2$
(4) $2: 1$

Answer (3)
Sol.

$\omega_{1}$

$\omega_{2}$
$\because \omega_{1} A_{1}=\omega_{2} A_{2}$
$\Rightarrow \frac{A_{1}}{A_{2}}=\frac{\omega_{2}}{\omega_{1}}$
$=\sqrt{\frac{9 K}{100} \times \frac{50}{2 K}}$
$=\sqrt{\frac{9}{4}}=\frac{3}{2}$
14. Statement A: An AC current can be created with 0 reactance.

Statement B: An AC circuit without power is not possible.
(1) Only statement $A$ is true
(2) Only statement $B$ is true
(3) Statement $A$ is true and $B$ is false
(4) Statement $A$ is false and $B$ is true

## Answer (3)

Sol. An ideal circuit with inductor and capacitor with $X_{L}=X_{C}$ can have zero reactance
$\Rightarrow$ Statement $A$ is true
In an AC circuit if only reactance part of impedance is there that is resistance is not available then average power consumed by the circuit can be zero.
$\Rightarrow$ Statement $B$ is false
15. It is given that a flywheel starting from rest covers 5 rad in first second, then the angle covered in next second is (consider constant angular acceleration)
(1) 10 rad
(2) 15 rad
(3) 20 rad
(4) 25 rad

Answer (2)
Sol. $\theta_{1 \text { st }}=5=\frac{1}{2} \alpha(1)^{2}$

$$
\begin{aligned}
\theta_{2 \mathrm{nd}} & =\theta_{2}-\theta_{1 \mathrm{st}} \\
& =\frac{1}{2} \alpha(2)^{2}-\frac{1}{2} \alpha(1)^{2} \\
& =\frac{1}{2} \alpha(3)=15 \mathrm{rad}
\end{aligned}
$$

16. A coil of 1000 turns and area $1 \mathrm{~m}^{2}$ is rotated about its diameter with constant angular velocity of 1 rotation/sec in a uniform and perpendicular magnetic field of 0.07 tesla. The maximum emf induced in the coil approximately is
(1) 200 volts
(2) 400 volts
(3) 220 volts
(4) 440 volts

Answer (4)
Sol. $\phi=N B A \cos \theta$
So, $\operatorname{Emf}=\frac{-d \phi}{d t}=N B A \omega \sin \theta$
So $E m f_{\max }=N A B \omega$
$=1000 \times 1 \times 0.07 \times 2 \pi$
$=440$ volts.
17. Ratio of masses initially in radioactive substance is $2: 1$. Total mass of substance is $10^{-2} \mathrm{~kg}$. Half lives of $A$ and $B$ is 4 sec and 8 sec respectively. Find the ratio of decayed mass of $A$ to that of $B$ after 16 sec .
(1) $\frac{11}{4}$
(2) $\frac{8}{3}$
(3) $\frac{5}{2}$
(4) $\frac{9}{5}$

Answer (3)
Sol. $\frac{N_{01}}{N_{02}}=\frac{2}{1}$
No. of half lives of $A=\frac{16}{4}=4$
No. of half lives of $B=\frac{16}{8}=2$
$N_{1}=\frac{2}{2^{4}}=\frac{1}{8}$
$N_{2}=\frac{1}{2^{2}}=\frac{1}{4}$
$\therefore$ ratio of decayed $=\frac{2-\frac{1}{8}}{1-\frac{1}{4}}=\frac{15 \times 4}{8 \times 3}$
18.
19.
20.

## SECTION - B

Numerical Value Type Questions: This section contains 10 questions. In Section-B, attempt any five questions out of 10 . The answer to each question is a NUMERICAL VALUE. For each question, enter the correct numerical value (in decimal notation, truncated/rounded-off to the second decimal place; e.g. $06.25,07.00,-00.33,-00.30,30.27,-27.30$ ) using the mouse and the on-screen virtual numeric keypad in the place designated to enter the answer.
21. If distance between sun and earth is R. How many years will the earth taken to complete one revolution if distance is $3 R$.

## Answer (5.20)

$3 \sqrt{3}$ Years
Sol. $\because T^{2} \propto R^{3}$

$$
\begin{aligned}
& \Rightarrow \quad\left(\frac{T_{2}}{T_{1}}\right)^{2}=\left(\frac{R_{2}}{R_{1}}\right)^{3} \\
& \Rightarrow \frac{T_{2}}{1}=\left(\frac{3}{1}\right)^{\frac{3}{2}}=3 \sqrt{3} \\
& \Rightarrow \quad T_{2}=3 \sqrt{3} \text { years } \simeq 5.20 \text { years }
\end{aligned}
$$

22. A projectile is fired at an angle $45^{\circ}$ with horizontal. After 2 seconds the projectile has velocity $20 \mathrm{~m} / \mathrm{s}$. Find range

## Answer (80 m)

Sol.

$\left(u \cos 45^{\circ}\right)^{2}+\left(u \sin 45^{\circ}-20\right)^{2}=(20)^{2}$
$\Rightarrow u=20 \sqrt{2} \mathrm{~m} / \mathrm{s}$
$\therefore \quad R=\frac{u^{2} \times \sin (2 \theta)}{g}$
$=\frac{20 \times 20 \times 2}{10}$
$=80 \mathrm{~m}$
23. A carnot engine absorbs heat 500 kcal at $727^{\circ} \mathrm{C}$ and rejects it at $127^{\circ} \mathrm{C}$. Find work done (in kcal) by carnot engine.

## Answer (300)

Sol. $\eta=\left(1-\frac{400}{1000}\right)=0.6$
$\therefore W=500 \times 0.6$

$$
=300 \mathrm{kcal}
$$

24. If the energy of capacitor increases by $44 \%$ then find percentage change in charge stored in capacitor.

## Answer (20)

Sol. $U=\frac{1}{2} \frac{q^{2}}{C}$

$$
U_{i}=\frac{q_{i}^{2}}{2 C} \& U_{f}=\frac{q_{f}^{2}}{2 C}
$$

$U_{f}=1.44 U_{i}$
$\frac{q_{f}^{2}}{2 C}=1.44 \frac{q_{i}^{2}}{2 C}$
$\Rightarrow q_{f}=1.2 q_{i}$
So $\frac{q_{f}-q_{i}}{q_{i}}=0.2$
So \% change in $q=20 \%$
25. In the circuit shown, current through Zener diode (in $m A$ ) is


Answer (0.125)

Sol.

$i_{L}=\frac{5}{10000}=0.5 \mathrm{~mA}$
$i_{R}=\frac{5}{8000}=0.625 \mathrm{~mA}$
$\therefore i_{z}=i_{R}-i_{L}=0.625-0.5$
$=0.125 \mathrm{~mA}$
26. $Q$ is the heat supplied to a system containing monoatomic gas. During the process if work done by the gas is $\frac{Q}{4}$ then molar specific heat during the process is $x R$. Value of $x$ is equal to $\qquad$ .

## Answer (2.0)

Sol. $\Delta U=Q-\frac{Q}{4}$
$\Rightarrow n C_{v} \Delta T=\frac{3 Q}{4}$
$\Rightarrow n \times\left(\frac{3 R}{2}\right) \Delta T=\frac{3 Q}{4}$
$\Rightarrow \frac{Q}{n \Delta T}=\frac{3}{2} R \times \frac{4}{3}$
$\therefore C=\frac{Q}{n \Delta T}=2 R$
$\therefore x=2$
27.
28.
29.
30.

## CHEMISTRY

## SECTION - A

Multiple Choice Questions: This section contains 20 multiple choice questions. Each question has 4 choices (1), (2), (3) and (4), out of which ONLY ONE is correct.

## Choose the correct answer :

1. $\mathrm{H}_{2}$ gas is produced in the preparation of
(1) $\mathrm{Na}_{2} \mathrm{CO}_{3}$
(2) NaOH
(3) Na metal
(4) $\mathrm{NaHCO}_{3}$

## Answer (2)

Sol. $\mathrm{H}_{2}$ gas is produced in the preparation of NaOH by electrolysis of aq. NaCl .

$$
\mathrm{NaCl} \longrightarrow \mathrm{Na}^{+}+\mathrm{Cl}^{-}
$$

$$
\mathrm{H}_{2} \mathrm{O} \rightleftharpoons \mathrm{H}^{+}+\mathrm{OH}^{-}
$$

Cathode : $2 \mathrm{H}^{+}+2 \mathrm{e}^{-} \longrightarrow \mathrm{H}_{2}$
Anode : $2 \mathrm{Cl}^{-} \longrightarrow \mathrm{Cl}_{2}+2 \mathrm{e}^{-}$
2. Among the following which one has highest melting point?
$\mathrm{Ag}, \mathrm{Hg}, \mathrm{Ga}, \mathrm{Ba}$
(1) Hg
(2) Ga
(3) Ag
(4) Ba

Answer (3)
Sol. Hg is liquid at room temperature. Ga has very low melting point around $30^{\circ} \mathrm{C}$.
Melting point of Ba is $727^{\circ} \mathrm{C}$ and melting point of Hg is around $960^{\circ} \mathrm{C}$.
Hence, highest melting point among given metals is of Ag .
3. The CFSE is maximum for
(1) $\left[\mathrm{Mo}\left(\mathrm{H}_{2} \mathrm{O}\right)_{6}\right]^{3+}$
(2) $\left[\mathrm{Cr}\left(\mathrm{H}_{2} \mathrm{O}\right)_{6}\right]^{3+}$
(3) $\left[\mathrm{Mn}\left(\mathrm{H}_{2} \mathrm{O}\right)_{6}\right]^{3+}$
(4) $\left[\mathrm{Fe}\left(\mathrm{H}_{2} \mathrm{O}\right) 6_{6}\right]^{3+}$

Answer (1)

Sol. Magnitude of CFSE value will be maximum for $\left[\mathrm{Mo}\left(\mathrm{H}_{2} \mathrm{O}\right)_{6}\right]^{3+}$ as Mo belongs to $4 d$ series of transition elements and has a $\mathrm{d}^{3}$ configurarion.
4. Stability of carbocation

(A)

(B)

(1) $A>B>C$
(2) $B>C>A$
(3) $A>C>B$
(4) $C>A>B$

Answer (1)
Sol. Stability of carbocation $\propto+\mathrm{R}$ effect of the substituent attached to the carbocation So the correct order is

5. Which gas does not do global warming effect?
(1) $\mathrm{CH}_{4}$
(2) $\mathrm{N}_{2}$
(3) $\mathrm{O}_{3}$
(4) $\mathrm{H}_{2} \mathrm{O}$

## Answer (2)

Sol. Methane, ozone and water vapour show greenhouse effect, hence cause global warming.
$\mathrm{N}_{2}$ does not cause global warming.
6. Which of the following use in fire extinguisher?
(1) Baking Soda
(2) Washing Soda
(3) Caustic Soda
(4) Soda ash

Answer (1)
Sol. Among the following compounds.
Sodium hydrogen carbonate, $\mathrm{NaHCO}_{3}$, Baking soda is used in fire extinguisher.
7. Name of given structure

(1) Ranitidine
(2) Cimetidine
(3) Histamine
(4) Terfenadine

## Answer (2)

Sol. The name of the given below structure is
Cimetidine

8. Which metal burns with green flame, blue from centre?
(1) $\mathrm{Fe}^{2+}$
(2) $\mathrm{Cu}^{2+}$
(3) $\mathrm{Fe}^{3+}$
(4) $\mathrm{Al}^{3+}$

## Answer (2)

Sol. When a platinum wire dipped in salt paste $\left(\mathrm{Cu}^{2+}\right)$ and conc. HCl is taken near the mouth of flame having oxidising zone, the colour imparted by $\mathrm{Cu}^{2+}$ is green with blue centre.
9. $\mathrm{PCl}_{5}$ exists but $\mathrm{NCl}_{5}$ does not, which of the following statement correctly explain the above?
(1) N -does not have vacant $d$-orbital
(2) P-does not have $2 d$ orbitals
(3) Back-bonding in $\mathrm{NCl}_{5}$ is not possible
(4) N -atom is more E.N. atom, so does not form 5 bonds

## Answer (1)

Sol. $\mathrm{NCl}_{5}$ does not exist as N does not have vacant $d$-orbitals, whereas $\mathrm{PCl}_{5}$ exists as P has 3 d orbitals available for bonding
10. If $\mathrm{K}=$ Kinetic Energy of H -atom
$\mathrm{P}=$ Potential Energy of H -atom
$\mathrm{T}=$ Total Energy of H -atom
As total energy increase then which of the following option is correct?
(1) $K$ and $P$ increase but $T$ decrease
(2) $P$ and $T$ increase but $K$ decrease
(3) All increase
(4) All decrease

Answer (2)
Sol. Potential Energy $=\frac{-\mathrm{Ke}^{2}}{r}$
Total Energy $=\frac{-\mathrm{Ke}^{2}}{2 r}$
Kinetic Energy $=\frac{\mathrm{Ke}^{2}}{2 r}$
As n increases, distance from nucleus increase and both potential energy and total energy increases.

KE decreases.
11. Number of peptide linkage in given sequence
ALA - VAL - GLY - LYS
(1) 5
(2) 4
(3) 3
(4) 6

Answer (3)
Sol. Number of peptide linkages in the given below sequence is 3

ALA - VAL - GLY - LYS
12. Which of the following option represents correct decreasing order of bond order of the following compounds?
$\mathrm{C}_{2}^{2-}, \mathrm{N}_{2}^{2-}, \mathrm{O}_{2}^{2-}$
(1) $\mathrm{C}_{2}^{2-}>\mathrm{N}_{2}^{2-}>\mathrm{O}_{2}^{2-}$
(2) $\mathrm{O}_{2}^{2-}>\mathrm{N}_{2}^{2-}>\mathrm{C}_{2}^{2-}$
(3) $\mathrm{O}_{2}^{2-}>\mathrm{C}_{2}^{2-}>\mathrm{N}_{2}^{2-}$
(4) $\mathrm{N}_{2}^{2-}>\mathrm{O}_{2}^{2-}>\mathrm{C}_{2}^{2-}$

## Answer (1)

Sol. Bond order of $\mathrm{C}_{2}^{2-}=3$
Bond order of $\mathrm{N}_{2}^{2-}=2$
Bond order of $\mathrm{O}_{2}^{2-}=1$
Correct order of bond order will be, $\mathrm{C}_{2}^{2-}>\mathrm{N}_{2}^{2-}>\mathrm{O}_{2}^{2-}$
13. Statement-1 : $\pi$-bond makes the compound unstable

Statement-2 : Bond strength of C=C (double bond) is more than $\mathrm{C}-\mathrm{C}$ (single bond)
(1) SI and S2 both are correct and S2 is correct explanation of S 1
(2) S1 and S2 both are correct and S2 is not the correct explanation of S1
(3) S1 is false and S 2 is correct
(4) S1 is correct and S2 is false

Answer (3)
Sol. Presence of multiple bond provides stability as it leads to decrease in energy.
$\mathrm{C}=\mathrm{C}$ consist of both $\sigma$ and $\pi$ overlap thus strength of $\mathrm{C}=\mathrm{C}$ is more than $\mathrm{C}-\mathrm{C}$.
14. Which of the following polymer is not a condensation polymer?
(1) Buna-N
(2) Silicone
(3) Nylon-6,6
(4) Dacron

Answer (1)
Sol. Buna-N is an addition copolymer of 1,3-butadiene and acrylonitrile

15. Which of the following sequence of reagent can perform the below conversion?
$\mathrm{CH}_{3}-\mathrm{CH}_{2}-\mathrm{CH}_{2}-\mathrm{OH} \longrightarrow \mathrm{CH}_{3}-\mathrm{CH}_{2}-\mathrm{CH}_{2}-\mathrm{CH}_{2}-\mathrm{NH}_{2}$
(1) $\mathrm{SOCl}_{2}, \mathrm{KCN}, \mathrm{H}_{2} / \mathrm{Pd}$
(2) $\mathrm{SOCl}_{2}, \mathrm{AgCN}, \mathrm{H}_{2} / \mathrm{Pd}$
(3) $\mathrm{PCl} 5, \mathrm{AgCN}, \mathrm{H}_{2} / \mathrm{Pd}$
(4) Red P/HI, KCN, $\mathrm{H}_{2} / \mathrm{Pd}$

Answer (1)

Sol. $\mathrm{CH}_{3}-\mathrm{CH}_{2}-\mathrm{CH}_{2}-\mathrm{OH} \xrightarrow{\mathrm{SOCl}_{2}}$


$\mathrm{CH}_{3}-\mathrm{CH}_{2}-\mathrm{CH}_{2}-\mathrm{CH}_{2}-\mathrm{NH}_{2}$
16. Find the cell constant of a given cell.

Given: Resistance $20 \Omega$, molar conductivity $=0.154 \times 10^{-3} \mathrm{Scm}^{2} \mathrm{~mol}^{-1}$

Concentration $=0.1 \mathrm{M}$
(1) $3.08 \times 10^{-7} \mathrm{~cm}^{-1}$
(2) $30.8 \times 10^{-7} \mathrm{~cm}^{-1}$
(3) $0.308 \times 10^{-9} \mathrm{~cm}^{-1}$
(4) $4.08 \times 10^{-5} \mathrm{~cm}^{-1}$

## Answer (1)

Sol. $\Lambda_{\mathrm{M}}=\frac{1000 \mathrm{~K}}{\mathrm{M}}$
$\Lambda_{\mathrm{M}}=0.154 \times 10^{-3} \mathrm{Scm}^{2} \mathrm{~mol}^{-1}$
Molarity $(M)=0.1 \mathrm{M}$
$K=\frac{0.154 \times 10^{-3} \times 0.1}{1000}=0.154 \times 10^{-7} \mathrm{Scm}^{-1}$
and $K=\frac{X}{R}$
$R=20 \Omega$
$x=0.154 \times 20 \times 10^{-7}$
$x=3.08 \times 10^{-7} \mathrm{~cm}^{-1}$
17. Which of the ore does not have sulphide in them?
(1) Baryte
(2) Copper pyrites
(3) Galena
(4) Zinc blende

Answer (1)
Sol. The formula of ores is,

$$
\begin{aligned}
& \text { Copper pyrite }=\mathrm{CuFeS}_{2} \\
& \text { Zinc blende }=\mathrm{ZnS} \\
& \text { Galena }=\mathrm{PbS} \\
& \text { Baryte }=\mathrm{BaSO}_{4}
\end{aligned}
$$

18. Which one is not the enamel of teeth?
(1) $\mathrm{Ca}^{2+}$
(2) $\mathrm{P}^{+3}$
(3) $\mathrm{F}^{\ominus}$
(4) $\mathrm{P}^{+5}$

Answer (2)
Sol. The $\mathrm{F}^{-}$ions make the enamel on teeth much harder by converting hydroxyapatite, $\left[3\left(\mathrm{Ca}_{3}\left(\mathrm{PO}_{4}\right)_{2} \cdot \mathrm{Ca}(\mathrm{OH})_{2}\right]\right.$, the enamel on the surface into much harder fluorapatite, $\left[3\left(\mathrm{Ca}_{3}\left(\mathrm{PO}_{4}\right)_{2}\right] \cdot \mathrm{CaF}_{2}\right]$.

So, $\mathrm{P}^{+3}$ is not present in enamel.
19. How many half-life are required to complete $90 \%$ of the reaction? [Given the reaction is first order]
(1) 3.32
(2) 2.07
(3) 1.44
(4) 4.02

Answer (1)
Sol. $\left(\frac{\ln 2}{t_{1 / 2}}\right)^{t_{90 \%}}=\ln \left(\frac{100}{10}\right)=\ln (10)$

$$
\begin{aligned}
& t_{90 \%} \times \frac{0.693}{t_{1 / 2}}=2.303 \\
& t_{90 \%}=\frac{t_{1 / 2}}{0.3}=t_{1 / 2} \times \frac{10}{3}=3.33 t_{1 / 2}
\end{aligned}
$$

20. 

## SECTION - B

Numerical Value Type Questions: This section contains 10 questions. In Section B, attempt any five questions out of 10 . The answer to each question is a NUMERICAL VALUE. For each question, enter the correct numerical value (in decimal notation, truncated/rounded-off to the second decimal place; e.g. $06.25,07.00,-00.33,-00.30,30.27,-27.30$ ) using the mouse and the on-screen virtual numeric keypad in the place designated to enter the answer.
21. Energy of one particle with wavelength is 300 nm . The energy of Na number of particle is $\mathrm{x} \times 10^{5} \mathrm{~J}$. Find $x$.

## Answer (4)

Sol. Energy of Na particles will be

$$
\begin{aligned}
& =\frac{6.626 \times 10^{-34} \times 3 \times 10^{8} \times 6.023 \times 10^{23}}{300 \times 10^{-9}} \\
& =0.397 \times 10^{6} \\
& \approx 4 \times 10^{5} \mathrm{~J} \\
& x=4 \text { (in nearest integer) }
\end{aligned}
$$

22. Volume occupied by a 3 g of gas A at 300 K is same as volume occupied by 0.2 g of $\mathrm{H}_{2}$ at 200 K . Find the molar mass of $A$.

## Answer (45)

Sol. $\mathrm{n}_{\mathrm{A}} \mathrm{T}_{\mathrm{A}}=\mathrm{n}_{\mathrm{H}_{2}} \mathrm{~T}_{\mathrm{H}_{2}}$ \{Assume constant pressure\}
$\frac{3}{M_{A}}(300)=\frac{0.2}{2}(200)$
$M_{A}=\frac{900}{0.1 \times 200}$
$=\frac{9000}{200}$
$=45 \mathrm{~g} / \mathrm{mol}$
23. For the equilibrium $A(g) \rightleftharpoons B(g), \Delta H$ is $-40 \mathrm{~kJ} / \mathrm{mole}$. If the ratio of the activation energy of the forward $\left(E_{a_{1}}\right)$ and backward $\left(E_{a_{b}}\right)$ is 2/3, then find the value of $E_{a}$

## Answer (80)

Sol. $\mathrm{A}(\mathrm{g}) \rightleftharpoons \mathrm{B}(\mathrm{g})$
Since, $\Delta H=E_{a_{f}}-E_{a_{b}}$
and, $\frac{E_{a_{f}}}{E_{a_{b}}}=\frac{2}{3}$

$$
E_{a_{\mathrm{f}}}=\frac{2 \mathrm{E}_{\mathrm{a}_{\mathrm{b}}}}{3}
$$

Putting in (i),

$$
\begin{aligned}
& -40=\frac{-\mathrm{E}_{\mathrm{a}_{\mathrm{b}}}}{3} \\
& \mathrm{E}_{\mathrm{a}_{\mathrm{b}}}=120 \mathrm{~kJ} / \mathrm{mole}
\end{aligned}
$$

Hence, $\mathrm{E}_{\mathrm{a}_{\mathrm{f}}}=-40+120=80 \mathrm{~kJ} / \mathrm{mole}$
24.
25.
26.
27.
28.
29.
30.

## MATHEMATICS

## SECTION - A

Multiple Choice Questions: This section contains 20 multiple choice questions. Each question has 4 choices (1), (2), (3) and (4), out of which ONLY ONE is correct.

## Choose the correct answer :

1. Find the area between the curves $y^{2}=2 x$ and $x+y=4$
(1) 10
(2) $\frac{11}{3}$
(3) 18
(4) $\frac{14}{3}$

Answer (3)


Solving the two curves, we get $(4-x)^{2}=2 x$
$\Rightarrow x^{2}-10 x+16=0$
$\Rightarrow x=2$ and $x=8$
Correspondingly $y=2$ and $y=-4$
Required Area
$=\int_{-4}^{2}\left[(4-y)-\frac{y^{2}}{2}\right] d y=4 y-\frac{y^{2}}{2}-\left.\frac{y^{3}}{6}\right|_{-4} ^{2}$
$=\left(8-2-\frac{4}{3}\right)-\left(-16-8+\frac{32}{3}\right)$
$=18$
2. For given probability distribution

| $x$ | 0 | 1 | 2 | 3 | 4 |
| :--- | :--- | :--- | :--- | :--- | :--- |
| $P(x)$ | $k$ | $2 k$ | $3 k$ | $4 k$ | $5 k$ |

Find $P\left(\frac{1<x<4}{x \leq 2}\right)$
(1) $\frac{1}{3}$
(2) $\frac{1}{4}$
(3) $\frac{2}{5}$
(4) $\frac{1}{2}$

## Answer (4)

Sol. $\because \quad \sum P(x)=1 \Rightarrow k+2 k+3 k+4 k+5 k=1$

$$
\begin{aligned}
& \Rightarrow k=\frac{1}{15} \\
& P\left(\frac{1<x<4}{x \leq 2}\right)=\frac{P(1<x<4 \cap x \leq 2)}{P(x \leq 2)} \\
& =\frac{P(x=2)}{P(x=0)+P(x=1)+P(x=2)} \\
& =\frac{3 k}{k+2 k+3 k}=\frac{1}{2}
\end{aligned}
$$

3. If $x>0, y>0$ and $x^{2} y^{3}=2^{15}$. Find the minimum value of $3 x+2 y$.
(1) $40\left(\frac{2}{3}\right)^{\frac{1}{5}}$
(2) $40\left(\frac{2}{3}\right)^{\frac{2}{5}}$
(3) $20\left(\frac{2}{3}\right)^{\frac{1}{5}}$
(4) $20\left(\frac{2}{3}\right)^{\frac{2}{5}}$

Answer (1)
Sol. $3 x+2 y=2\left(\frac{3 x}{2}\right)+3\left(\frac{2 y}{3}\right)$
as $x, y>0 \mathrm{AM}-\mathrm{GM}$ inequality to be applied

$$
\begin{aligned}
& \Rightarrow \quad \frac{2\left(\frac{3 x}{2}\right)+3\left(\frac{2 y}{3}\right)}{5} \geq\left(\left(\frac{3 x}{2}\right)^{2}\left(\frac{2 y}{3}\right)^{3}\right)^{\frac{1}{5}} \\
& \Rightarrow \quad \frac{3 x+2 y}{5} \geq\left(\frac{2}{3} \cdot 2^{15}\right)^{\frac{1}{3}}\left(\text { as } x^{2} y^{3}=2^{15}\right) \\
& \Rightarrow 3 x+2 y \geq 40\left(\frac{2}{3}\right)^{\frac{1}{5}}
\end{aligned}
$$

4. For hyperbola : $\frac{x^{2}}{a^{2}}-y^{2}=1$

Ellipse: $3 x^{2}+4 y^{2}=12$. If length of latus rectum is equal then find $12\left(e_{E}^{2}+e_{H}^{2}\right)$
(1) 36
(2) 48
(3) 42
(4) $\frac{41}{2}$

## Answer (3)

Sol. H: $\frac{x^{2}}{a^{2}}-\frac{y^{2}}{1}=1$, length of latus rectum $=\frac{2.1}{a}=\frac{2}{a}$
$\mathrm{E}: \frac{x^{2}}{4}-\frac{y^{2}}{3}=1$, length of latus rectum $=\frac{2(3)}{2}=3$
$\because \frac{2}{a}=3 \Rightarrow a=\frac{2}{3}$
$\because \quad e_{E}^{2}=1-\frac{3}{4}=\frac{1}{4}$
and $e_{H}^{2}=1+\frac{1}{a^{2}}=1+\frac{9}{4}=\frac{13}{4}$
So, $12\left(e_{E}^{2}+e_{H}^{2}\right)=12\left(\frac{1}{4}+\frac{13}{4}\right)=42$
5.
6. Find the sum of roots of the equation $\left(e^{2 x}-4\right)$ $\left(6 e^{2 x}-5 e^{x}+1\right)=0$
(1) $\ln 6$
(2) $\ln 3$
(3) $-\ln 3$
(4) In2

## Answer (3)

Sol. $\left(e^{2 x}-4\right)\left(6 e^{2 x}-5 e^{x}+1\right)=0$
$\Rightarrow\left(e^{x}-2\right)\left(e^{x}+2\right)\left(3 e^{x}-1\right)\left(2 e^{x}-1\right)=0$
$\Rightarrow \quad e^{x}=2, \frac{1}{3}, \frac{1}{2}$
$\therefore \quad x=\ln 2,-\ln 3,-\ln 2$
$\therefore$ sum of roots $=-\ln 3$
7. $\operatorname{lt}_{n \rightarrow \infty} \sum_{r=1}^{n} \frac{n^{2}}{\left(n^{2}+r^{2}\right)(n+r)}$ equals
(1) $\frac{1}{4} \ln 2+\frac{\pi}{8}$
(2) $\frac{1}{4} \ln 2-\frac{\pi}{8}$
(3) $-\frac{1}{4} \ln 2-\frac{\pi}{8}$
(4) $-\frac{1}{4} \ln 2+\frac{\pi}{8}$

Answer (1)

Sol. $\lim _{n \rightarrow \infty} \frac{1}{n} \sum_{r=1}^{n} \frac{1}{\left(1+\frac{r^{2}}{n^{2}}\right)\left(1+\frac{r}{n}\right)}$
$=\int_{0}^{1} \frac{d x}{\left(1+x^{2}\right)(1+x)}$
$\left\{\begin{array}{l}\text { let } \frac{1}{\left(1+x^{2}\right)(1+x)}=\frac{A}{1+x}+\frac{B x+C}{1+x^{2}} \\ \Rightarrow 1=A\left(1+x^{2}\right)+(B x+C)(1+x) \\ \text { So } A=\frac{1}{2}, B=-\frac{1}{2}, C=\frac{1}{2}\end{array}\right\}$
$=\frac{1}{2} \int_{0}^{1} \frac{d x}{1+x}-\frac{1}{2} \int_{0}^{1} \frac{x d x}{1+x^{2}}+\frac{1}{2} \int_{0}^{1} \frac{d x}{1+x^{2}}$
$=\left[\frac{1}{2} \ln (1+x)-\frac{1}{4} \ln \left(1+x^{2}\right)+\frac{1}{2} \tan ^{-1} x\right]_{0}^{1}$
$=\frac{1}{2} \ln 2-\frac{1}{4} \ln 2+\frac{\pi}{8}=\frac{1}{4} \ln 2+\frac{\pi}{8}$
8. Find the number of solutions of
$\cos \left(\frac{\pi}{3}-x\right) \cos \left(\frac{\pi}{3}+x\right)=\frac{\cos ^{2} 2 x}{4}$, in $(-2 \pi, 2 \pi)$
(1) 1
(2) 5
(3) 3
(4) 8

Answer (3)
Sol. $\cos ^{2} \frac{\pi}{3}-\sin ^{2} x=\frac{\cos ^{2} 2 x}{4}$
$\Rightarrow \quad 1-4 \sin ^{2} x=\left(1-2 \sin ^{2} x\right)^{2}$
$\Rightarrow 4 \sin ^{4} x=0 \Rightarrow \sin x=0$
$\Rightarrow \quad x=-\pi, 0, \pi$
9.
10. If $\log _{x+1}\left(x^{2}-x+6\right)^{2}=4$, find $x$.
(1) $\frac{4}{3}$
(2) $\frac{5}{3}$
(3) $\frac{1}{3}$
(4) $\frac{7}{3}$

Answer (2)

Sol. $\log _{(x+1)}\left(x^{2}-x+6\right)^{2}=4$
$\Rightarrow\left(x^{2}-x+6\right)^{2}=(x+1)^{4}$
$\Rightarrow\left(x^{2}-x+6\right)^{2}=\left(x^{2}+2 x+1\right)^{2}$
$\Rightarrow\left(x^{2}-x+6-x^{2}-2 x-1\right)\left(x^{2}-x+6+x^{2}+2 x+1\right)=0$
$\Rightarrow(-3 x+5)\left(2 x^{2}+x+7\right)=0$
So, $x=\frac{5}{3}$ is only possible real solution.
11. Let a triangle of maximum area be inscribed in the ellipse $\frac{x^{2}}{a^{2}}+\frac{y^{2}}{4}=1(a>2)$, such that one of its vertices coincides with one end of major axis, is $6 \sqrt{3}$ sq. units. Find its eccentricity.
(1) $\frac{1}{\sqrt{2}}$
(2) $\frac{\sqrt{3}}{2}$
(3) $\frac{1}{2}$
(4) $\frac{1}{3}$

## Answer (2)

Sol.

$\Delta=\frac{1}{2}(4 \sin \theta)(a-a \cos \theta)$
$\Rightarrow \quad \Delta=2 a(\sin \theta-\sin \theta \cos \theta)$
$\Rightarrow \quad \Delta=a(2 \sin \theta-\sin 2 \theta)$
$\Rightarrow \quad \Delta^{\prime}=a(2 \cos \theta-2 \cos 2 \theta)=0$
So, $\theta=\frac{2 \pi}{3}$
Hence $\Delta_{\max }=a\left(2 \cdot \frac{\sqrt{3}}{2}+\frac{\sqrt{3}}{2}\right)=\frac{3 \sqrt{3} a}{2}=6 \sqrt{3}$
So, $a=4$
Now, $e=\sqrt{1-\frac{4}{16}}=\frac{\sqrt{3}}{2}$
12.
13. $\int_{-\pi / 2}^{\pi / 2} \frac{d x}{\left(1+e^{x}\right)\left(\sin ^{6} x+\cos ^{6} x\right)}$ equals
(1) $\pi$
(2) $\frac{3 \pi}{4}$
(3) $\frac{5 \pi}{4}$
(4) $\frac{\pi}{3}$

## Answer (1)

Sol. Let $I=\int_{-\pi / 2}^{\pi / 2} \frac{d x}{\left(1+e^{x}\right)\left(\sin ^{6} x+\cos ^{6} x\right)} \ldots$ (i)

$$
\begin{equation*}
I=\int_{-\pi / 2}^{\pi / 2} \frac{d x}{\left(1+e^{-x}\right)\left(\sin ^{6} x+\cos ^{6} x\right)} \tag{ii}
\end{equation*}
$$

Using $\int_{a}^{b} f(x) d x=\int_{a}^{b} f(a+b-x) d x$
Adding eqns. (i) and (ii), we get

$$
\begin{aligned}
& 2 I=\int_{-\pi / 2}^{\pi / 2} \frac{d x}{\sin ^{6} x+\cos ^{6} x} \\
& \Rightarrow \quad I=\int_{0}^{\pi / 2} \frac{d x}{\sin ^{6} x+\cos ^{6} x}
\end{aligned}
$$

$$
\Rightarrow \quad I=\int_{0}^{\pi / 2} \frac{d x}{1-3 \sin ^{2} x \cos ^{2} x}=\int_{0}^{\pi / 2} \frac{4 d x}{4-3 \sin ^{2} 2 x}
$$

$$
\Rightarrow \quad I=2 \int_{0}^{\pi / 4} \frac{4 \sec ^{2} 2 x}{4 \sec ^{2} 2 x-3 \tan ^{2} 2 x} d x
$$

Let $\tan 2 x=t$
$\Rightarrow 2 \sec ^{2} 2 x d x=d t$
$\Rightarrow \quad I=\int_{0}^{\infty} \frac{4 d t}{4+t^{2}}=\left.2 \tan ^{-1}\left(\frac{t}{2}\right)\right|_{0} ^{\infty}=\pi$

14 Let $f(x)=\left\{\begin{array}{l}\frac{\sin \{x\}}{\{x\}}\end{array} \quad x \in(-2,-1)\right.$
14. Let $f(x)=$
$\max .([|x|], 2 x) \quad|x|<1$

$$
-1 \quad \text { otherwise }
$$

If $f(x)$ is not differentiable at $m$ points and not continuous at $n$ points, then $(m, n)$ equals (where \{ \} represents fractional part function and [ ] represents greatest integer function respectively).
(1) $(2,3)$
$(2)(3,3)$
(3) $(2,4)$
$(4)(4,3)$

## Answer (4)

Sol. $f(x)=\left\{\begin{array}{cl}\frac{\sin (x+2)}{x+2} & \text { if } x \in(-2,-1) \\ 0 & \text { if } x \in(-1,0) \\ 2 x & \text { if } x \in(0,1) \\ -1 & \text { otherwise }\end{array}\right.$
$f(x)$ is discontinuous at $=x=-2,-1,1$
$f(x)$ is not differentiable at $=x=-2,-1,0$ and 1
15. If $\Delta_{r}=\left|\begin{array}{ccc}2^{r-1} & \frac{(r+1)!}{1+\frac{1}{r}} & 4 r^{3}-2 n r \\ a & b & c \\ 2^{n}-1 & (n+1)!-1 & n^{3}(n+1)\end{array}\right|$
then, $\sum_{r=1}^{n} \Delta_{r}=$
(1) $a b c$
(2) $(n+3)$ !
(3) 0
(4) $a n!+b 2^{n}+c$

## Answer (3)

Sol. $\sum_{r=1}^{n} \Delta_{r}=\left|\begin{array}{ccc}\sum_{r=1}^{n} 2^{r-1} & \sum_{r=1}^{n}((r+1)!-r!) & \sum_{r=1}^{n}\left(4 r^{3}-2 n r\right) \\ a & b & c \\ 2^{n}-1 & (n+1)!-1 & n^{3}(n+1)\end{array}\right|$

$$
=\left|\begin{array}{ccc}
2^{n}-1 & (n+1)!-1 & n^{3}(n+1) \\
a & b & c \\
2^{n}-1 & (n+1)!-1 & n^{3}(n+1)
\end{array}\right|=0
$$

16. If slope of normal of a curve $y=y(x)$ is $\frac{x^{2}}{x y-x^{2} y^{2}-1}$, and it passes through $(1,1)$ then find $e y(e)$.
(1) $\frac{1+\tan 1}{1-\tan 1}$
(2) 1
(3) $\tan 1$
(4) $\frac{1-\tan 1}{1+\tan 1}$

## Answer (1)

Sol. $\because \quad-\frac{d x}{d y}=\frac{x^{2}}{x y-x^{2} y^{2}-1}$
$\Rightarrow-x y d x+x^{2} y^{2} d x+d x=x^{2} d y$
$\left(x^{2} y^{2}+1\right) d x=x(x d y+y d x)$
$\Rightarrow \quad \frac{d x}{x}=\frac{d(x y)}{1+(x y)^{2}}$
$\Rightarrow \operatorname{In} x=\tan ^{-1} x y+C \quad$ passes through $(1,1)$
So $C=-\frac{\pi}{4}$
$\Rightarrow \quad x y=\tan \left(\frac{\pi}{4}+\ln x\right)$
Put $x=e$, we get $e \cdot y(e)=\tan \left(\frac{\pi}{4}+1\right)$
$=\frac{1+\tan 1}{1-\tan 1}$
17. Curve $C$ has tangent at $P$ such that it meets $x$-axis at $Q$. $Y$-axis bisects $P Q$, then curve $C$ is a parabola passing through $(3,3)$ with
(1) focus $\left(\frac{4}{3}, 0\right)$
(2) focus $\left(0, \frac{3}{4}\right)$
(3) Latus Rectum $=3$
(4) Latus Rectum $=6$

## Answer (3)

Sol. Let $(x, y)$ be point on curve tangent at $P(x, y)$

$Y-y=m(X-x)$
Where $m=\frac{d y}{d x}$
For $Q$
Put $y=0$
$\Rightarrow \quad X=x-\frac{y}{m}$
As $y$-axis bisects $P Q$,
$x+x-\frac{y}{m}=0$
$\Rightarrow \frac{d y}{y}=\frac{d x}{2 x}$
$\ln y=\frac{1}{2} \ln x+\ln C$
$\Rightarrow y^{2}=C x$
As it passes through $(3,3)$
$\Rightarrow C=3$
$\Rightarrow$ Parabola $y^{2}=3 x$
So length of Latus Rectum $=3$
18. If it is given that $x^{*} y=x^{2}+y^{3}$. Now if $\left(x^{*} 1\right)^{*} 1$ and $x^{*}\left(1^{*} 1\right)$ both are equal, then find the value of $2 \sin ^{-1}\left[\frac{x^{4}+x^{2}-2}{x^{4}+x^{2}+2}\right]$
(1) $\frac{\pi}{6}$
(2) $\frac{\pi}{2}$
(3) $\frac{2 \pi}{3}$
(4) $\frac{\pi}{3}$

Answer (4)

Sol. $\because \quad\left(x^{*} 1\right)^{*} 1=\left(x^{2}+1\right)^{*} 1=\left(x^{2}+1\right)^{2}+1=x^{4}+2 x^{2}+2$
And $x^{*}(1 * 1)=x^{*}(1+1)=x^{*} 2=x^{2}+8$
$\therefore \quad x^{4}+2 x^{2}+2=x^{2}+8$
So $x^{4}+x^{2}-6=0 \Rightarrow\left(x^{2}+3\right)\left(x^{2}-2\right)=0$
Hence $x^{2}=2$

$$
\begin{aligned}
& \text { Now, } 2 \sin ^{-1}\left(\frac{x^{4}+x^{2}-2}{x^{4}+x^{2}+2}\right)=2 \sin ^{-1}\left(\frac{4+2-2}{4+2+2}\right) \\
& =2 \sin ^{-1}\left(\frac{1}{2}\right)=\frac{\pi}{3}
\end{aligned}
$$

19. 
20. 

## SECTION - B

Numerical Value Type Questions: This section contains 10 questions. In Section B, attempt any five questions out of 10. The answer to each question is a NUMERICAL VALUE. For each question, enter the correct numerical value (in decimal notation, truncated/rounded-off to the second decimal place; e.g. $06.25,07.00,-00.33,-00.30,30.27,-27.30$ ) using the mouse and the on-screen virtual numeric keypad in the place designated to enter the answer.
21. If the circle $(x-h)^{2}+(y-k)^{2}=r^{2}$ touches $x$-axis at (1, 0), $k>0$ and $x+y=0$ intersect the circle at two points making. Chord of length 2 units. Find $h+k+r$.

## Answer (07)

Sol. Equation of circle touching $x$-axis at $(1,0)$ is
$(x-1)^{2}+y^{2}+\lambda y=0$, where $\lambda$ is parameter
$(x-1)^{2}+\left(y+\frac{\lambda}{2}\right)^{2}=\frac{\lambda^{2}}{4}$,
on comparing with $(x-h)^{2}+(y-k)^{2}=r^{2}$
we get, $h=1, k=-\frac{\lambda}{2}, r=\frac{\lambda}{2}$
distance of chord $x+y=0$ from centre $\left(1,-\frac{\lambda}{2}\right)$ is equal to
$\frac{\left|1-\frac{\lambda}{2}\right|}{\sqrt{2}}=\sqrt{\frac{\lambda^{2}}{4}-1}$
$\Rightarrow \frac{\lambda^{2}}{4}+\lambda-3=0$
$\Rightarrow \lambda=-6,2$ but 2 is not acceptable.
$\therefore \quad h=1, k=3, r=3$
$\therefore h+k+r=7$
22. If the system of equations $x+y+\alpha z=1$
$x+y+3 z=1$, has unique solution
$x+2 z=4$
if $\alpha \in\{1,2,3, \ldots \ldots . .100\}$ and
$\operatorname{HCF}(24, \alpha)=1$, find sum of values of $\alpha$.

## Answer (1633)

Sol. For unique solution
$\left|\begin{array}{lll}1 & 1 & \alpha \\ 1 & 1 & 3 \\ 1 & 0 & 2\end{array}\right| \neq 0 \Rightarrow \alpha \neq 3$
If $\operatorname{HCF}(24, \alpha)=1$ then $\alpha$ should not be multiple of 2 OR 3.

Required sum $=$

$$
\begin{aligned}
& (1+2+3+\ldots+100)-(2+4+6+\ldots+100) \\
& -(3+6+9+\ldots+99)+(6+12+18+\ldots+96) \\
= & \frac{100 \cdot 101}{2}-2 \cdot \frac{50 \cdot 51}{2}-\frac{3 \cdot 33 \cdot 34}{2}+6 \cdot \frac{16 \cdot 17}{2} \\
= & 1633
\end{aligned}
$$

23. Number of real solutions of the equation $x^{7}-x+3$ $=0$ is

## Answer (01)

Sol. $x^{7}=x-3$


We can observe that there exists only one point of intersection of these two curves, so number of real solution is 1.
24. Find the remainder when $1+3+3^{2}+$ $\qquad$ $3^{2021}$ is divided by 50.

## Answer (04)

Sol. $1+3+3^{2}+\ldots \ldots \ldots \ldots .+3^{2021}=\frac{3^{2022}-1}{2}$
$\because \quad \frac{3^{2022}-1}{2}=\frac{(10-1)^{1011}-1}{2}$
$=\frac{\left({ }^{1011} C_{0} \cdot 10^{1011}-{ }^{1011} C_{1} \cdot 10^{1010}+\ldots .+{ }^{1011} C_{1010} \cdot 10-1-1\right)}{2}$
$=50[$ Integer $]+1011.5-1$
$=50$ [Integer] +5054
So, remainder will be 4 .
25. Find the number of 7 digits numbers which are divisible by 11 and made up of $1,2,3,4,5,7,9$
Answer (576)
Sol. Sum of all digits $=31$
So clearly sum of digits at even places is 10 and sum of digits at odd places is 21 or vice versa.
Case I:5, 7, 9 are used at even places
No. of numbers $=\lfloor 3 . \mid 4=144$
Case II: 1, 2, 7 are used at even places
No. of numbers $=\lfloor 3.44=144$
Case III: 2, 3, 5 are used at even places
No. of numbers $=\lfloor 3 . \mid 4=144$
Case IV:1,4, 5 are used at even places
No. of numbers $=\lfloor 3 . \mid 4=144$
Total numbers $=576$
26.
27.
28.
29.
30.

