26/06/2022
Morning

Corporate Office : Aakash Tower, 8, Pusa Road, New Delhi-110005 | Ph.: 011-47623456

## Answers \& Solutions

Time : 3 hrs.

## JEE (Main)-2022 (Online) Phase-1

## (Physics, Chemistry and Mathematics)

IMPORTANT INSTRUCTIONS:
(1) The test is of $\mathbf{3}$ hours duration.
(2) The Test Booklet consists of 90 questions. The maximum marks are 300 .
(3) There are three parts in the question paper consisting of Physics, Chemistry and Mathematics having 30 questions in each part of equal weightage. Each part (subject) has two sections.
(i) Section-A: This section contains 20 multiple choice questions which have only one correct answer. Each question carries $\mathbf{4}$ marks for correct answer and $\mathbf{- 1}$ mark for wrong answer.
(ii) Section-B: This section contains 10 questions. In Section-B, attempt any five questions out of 10. The answer to each of the questions is a numerical value. Each question carries 4 marks for correct answer and -1 mark for wrong answer. For Section-B, the answer should be rounded off to the nearest integer.

## PHYSICS

## SECTION - A

Multiple Choice Questions: This section contains 20 multiple choice questions. Each question has 4 choices (1), (2), (3) and (4), out of which ONLY ONE is correct.

## Choose the correct answer :

1. An expression for a dimensionless quantity $P$ is given by $P=\frac{\alpha}{\beta} \log _{e}\left(\frac{k t}{\beta x}\right)$; where $\alpha$ and $\beta$ are constants, $x$ is distance; $k$ is Boltzmann constant and $t$ is the temperature. Then the dimensions of $\alpha$ will be
(A) $\left[\mathrm{M}^{0} \mathrm{~L}^{-1} \mathrm{~T}^{0}\right]$
(B) $\left[\mathrm{ML}^{0} \mathrm{~T}^{-2}\right]$
(C) $\left[\mathrm{MLT}^{-2}\right]$
(D) $\left[\mathrm{ML}^{2} \mathrm{~T}^{-2}\right]$

## Answer (C)

Sol. $[\alpha]=[\beta]=\left[\frac{k t}{x}\right]$

$$
\begin{aligned}
& =\left[\frac{\mathrm{ML}^{2} \mathrm{~T}^{-2}}{\mathrm{~L}}\right] \\
& =\left[\mathrm{MLT}^{-2}\right]
\end{aligned}
$$

2. A person is standing in an elevator. In which situation, he experiences weight loss?
(A) When the elevator moves upward with constant acceleration
(B) When the elevator moves downward with constant acceleration
(C) When the elevator moves upward with uniform velocity
(D) When the elevator moves downward with uniform velocity

## Answer (B)

Sol. Apparent weight $=m(g-a)$
$\Rightarrow$ Weight loss in downward accelerated elevator
3. An object is thrown vertically upwards. At its maximum height, which of the following quantity becomes zero?
(A) Momentum
(B) Potential Energy
(C) Acceleration
(D) Force

## Answer (A)

Sol. At topmost position,
$v=0$
$\Rightarrow$ momentum $=0$
4. A ball is released from rest from point $P$ of a smooth semi-spherical vessel as shown in figure. The ratio of the centripetal force and normal reaction on the ball at point $Q$ is $A$ while angular position of point $Q$ is $\alpha$ with respect to point $P$. Which of the following graphs represent the correct relation between $A$ and $\alpha$ when ball goes from $Q$ to $R$ ?

(A)

(B)

(C)

(D)


## Answer (C)

Sol. $N=m g \sin \alpha+\frac{m v^{2}}{R}$

$$
\text { and, } v^{2}=2 g \times R \sin \alpha
$$

$\therefore \quad N=m g \sin \alpha+m \times(2 g \sin \alpha)$

$$
=3 m g \sin \alpha
$$

$\therefore \quad$ ratio, $A=\frac{\frac{m v^{2}}{R}}{N}$

$$
\begin{aligned}
& =\frac{2 m g \sin \alpha}{3 m g \sin \alpha} \\
& =\frac{2}{3}
\end{aligned}
$$


5. A thin circular ring of mass $M$ and radius $R$ is rotating with a constant angular velocity $2 \mathrm{rad} \mathrm{s}^{-1}$ in a horizontal plane about an axis vertical to its plane and passing through the center of the ring. If two objects each of mass $m$ be attached gently to the opposite ends of a diameter of ring, the ring will then rotate with an angular velocity (in rad s-1).
(A) $\frac{M}{(M+m)}$
(B) $\frac{(M+2 m)}{2 M}$
(C) $\frac{2 M}{(M+2 m)}$
(D) $\frac{2(M+2 m)}{M}$

## Answer (C)

Sol. $I_{1} \omega_{1}=I_{2} \omega_{2}$

$$
\begin{aligned}
& \Rightarrow \quad M R^{2} \times 2=\left(M R^{2}+2 m R^{2}\right) \omega_{2} \\
& \Rightarrow \quad \omega_{2}=\frac{2 M}{M+2 m}
\end{aligned}
$$

6. The variation of acceleration due to gravity ( $g$ ) with distance ( $r$ ) from the center of the earth is correctly represented by
(Given $R=$ radius of earth)
(A)

(B)

(C)

(D)


Answer (A)

Sol. For $r<R g=\frac{G m r}{R^{3}}=\operatorname{Cr}(C=$ Constant $)$
For $r>R g=\frac{G m}{r^{2}}=\frac{C^{\prime}}{r^{2}}\left(C^{\prime}=\right.$ Constant $)$
For the above equations the best suited graph is as given in option (A)
7. The efficiency of a Carnot's engine, working between steam point and ice point, will be
(A) $26.81 \%$
(B) $37.81 \%$
(C) $47.81 \%$
(D) $57.81 \%$

## Answer (A)

Sol. $\eta=1-\frac{T_{C}}{T_{H}}=\frac{T_{H}-T_{C}}{T_{H}}$
$=\frac{100}{373} \times 100 \%$
$=26.81 \%$
$\Rightarrow$ option (A)
8. Time period of a simple pendulum in a stationary lift is ' $T$. If the lift accelerates with $\frac{g}{6}$ vertically upwards then the time period will be
(Where $g=$ acceleration due to gravity)
(A) $\sqrt{\frac{6}{5}} T$
(B) $\sqrt{\frac{5}{6}} T$
(C) $\sqrt{\frac{6}{7}} T$
(D) $\sqrt{\frac{7}{6}} T$

## Answer (C)

Sol. $T^{\prime}=2 \pi \sqrt{\frac{l}{g_{\text {eff }}}}$

$$
\begin{aligned}
& T^{\prime}=2 \pi \sqrt{\frac{l}{g+\frac{g}{6}}}=2 \pi \sqrt{\frac{6 l}{7 g}} \\
& \Rightarrow T^{\prime}=\sqrt{\frac{6}{7}} T \\
& \Rightarrow \text { Option (C) }
\end{aligned}
$$

9. A thermally insulated vessel contains an ideal gas of molecular mass $M$ and ratio of specific heats 1.4. Vessel is moving with speed $v$ and is suddenly brought to rest. Assuming no heat is lost to the surrounding and vessel temperature of the gas increases by
( $R=$ universal gas constant)
(A) $\frac{M v^{2}}{7 R}$
(B) $\frac{M v^{2}}{5 R}$
(C) $2 \frac{M v^{2}}{7 R}$
(D) $7 \frac{M v^{2}}{5 R}$

## Answer (B)

Sol. $\frac{1}{2} m v^{2}=n \frac{5}{2} R \Delta T$

$$
\begin{aligned}
\Rightarrow \Delta T & =\frac{m v^{2}}{5 n R} \\
& =\frac{M v^{2}}{5 R}
\end{aligned}
$$

Option (B)
10. Two capacitors having capacitance $C_{1}$ and $C_{2}$ respectively are connected as shown in figure. Initially, capacitor $C_{1}$ is charged to a potential difference $V$ volt by a battery. The battery is then removed and the charged capacitor $C_{1}$ is now connected to uncharged capacitor $C_{2}$ by closing the switch $S$. The amount of charge on the capacitor $C_{2}$, after equilibrium, is

(A) $\frac{C_{1} C_{2}}{\left(C_{1}+C_{2}\right)} V$
(B) $\frac{\left(C_{1}+C_{2}\right)}{C_{1} C_{2}} V$
(C) $\left(C_{1}+C_{2}\right) V$
(D) $\left(C_{1}-C_{2}\right) V$

## Answer (A)

Sol. $V_{\text {common }}=\frac{C_{1} V}{C_{1}+C_{2}}$
$\Rightarrow$ Charge on capacitor $C_{2}$

$$
\begin{aligned}
& =C_{2} V_{\text {common }} \\
& =\frac{C_{1} C_{2} V}{C_{1}+C_{2}}
\end{aligned}
$$

$$
\Rightarrow \text { Option }(\mathrm{A})
$$

11. Given below two statements: One is labelled as Assertion (A) and other is labelled as Reason (R).
Assertion (A) : Non-polar materials do not have any permanent dipole moment.
Reason (R) : When a non-polar material is placed in an electric field, the centre of the positive charge distribution of it's individual atom or molecule coincides with the centre of the negative charge distribution.
In the light of above statements, choose the most appropriate answer from the options given below.
$(A)$ Both (A) and (R) are correct and (R) is the correct explanation of (A).
$(B)$ Both $(A)$ and (R) are correct and (R) is not the correct explanation of (A).
$(C)(A)$ is correct but $(R)$ is not correct.
(D) (A) is not correct but (R) is correct.

## Answer (C)

Sol. Non polar material does not have any permanent dipole moment and when placed in an electric field the positive and negative charges displace in opposite directions and result into an induced dipole moment as long as the field is applied.
12. The magnetic flux through a coil perpendicular to its plane is varying according to the relation $\phi=\left(5 t^{3}+4 t^{2}+2 t-5\right)$ Weber. If the resistance of the coil is 5 ohm, then the induced current through the coil at $t=2 \mathrm{~s}$ will be,
(A) 15.6 A
(B) 16.6 A
(C) 17.6 A
(D) 18.6 A

Answer (A)
Sol. Emf $=-\frac{d \phi}{d t}=-\left(15 t^{2}+8 t+2\right)$
So, $i=\frac{|E \mathrm{Ef}|}{R}=\frac{\left(15 t^{2}+8 t+2\right)}{5}$
at $t=2$
$i=15.6 \mathrm{~A}$
13. An aluminium wire is stretched to make its length, $0.4 \%$ larger. The percentage change in resistance is :
(A) $0.4 \%$
(B) $0.2 \%$
(C) $0.8 \%$
(D) $0.6 \%$

## Answer (C)

Sol. When the wire is stretched, volume remains constant. If length is increased by $0.4 \%$ area will decrease by $0.4 \%$ so

From $R=\rho \frac{I}{A}$
$\frac{d R}{R} \times 100=\frac{d l}{l} \times 100+\frac{d A}{A} \times 100$
$\% R=0.4+0.4=0.8 \%$
14. A proton and an alpha particle of the same velocity enter in a uniform magnetic field which is acting perpendicular to their direction of motion. The ratio of the radii of the circular paths described by the alpha particle and proton is :
(A) $1: 4$
(B) $4: 1$
(C) $2: 1$
(D) $1: 2$

## Answer (C)

Sol. $R=\frac{m v}{q B}$
$\frac{R_{\alpha}}{R_{p}}=\frac{m_{\alpha} / q_{\alpha}}{m_{p} / q_{p}}=2$
15. If electric field intensity of a uniform plane electro magnetic wave is given as
$E=-301.6 \sin (k z-\omega t) \hat{a}_{x}+452.4 \sin (k z-\omega t) \hat{a}_{y} \frac{V}{m}$.
Then, magnetic intensity ' H ' of this wave in $\mathrm{Am}^{-1}$ will be :
[Given : Speed of light in vacuum $c=3 \times 10^{8} \mathrm{~ms}^{-1}$, Permeability of vacuum $\mu_{0}=4 \pi \times 10^{-7} \mathrm{NA}^{-2}$ ]
(A) $+0.8 \sin (k z-\omega t) \hat{a}_{y}+0.8 \sin (k z-\omega t) \hat{a}_{x}$.
(B) $+1.0 \times 10^{-6} \sin (k z-\omega t) \hat{a}_{y}+1.5 \times 10^{-6}$

$$
(k z-\omega t) \hat{a}_{x}
$$

(C) $-0.8 \sin (k z-\omega t) \hat{a}_{y}-1.2 \sin (k z-\omega t) \hat{a}_{x}$
(D) $-1.0 \times 10^{-6} \sin (k z-\omega t) \hat{a}_{y}-1.5 \times 10^{-6}$

$$
\sin (k z-\omega t) \hat{a}_{x}
$$

## Answer (C)

Sol. We know
$\vec{B} \times \vec{C}=\vec{E}$
Taking cross product of $\vec{C}$ both the sides
$\vec{C} \times(\vec{B} \times \vec{C})=\vec{C} \times \vec{E}$
So $\vec{B}=\frac{\vec{C} \times \vec{E}}{C^{2}}$
$\vec{C}=C \hat{k}$
$\vec{E}=-301.6 \sin (k z-\omega t) \hat{a}_{x}+452.4 \sin (k z-\omega t) \hat{a}_{y}$
and $\vec{H}=\frac{\vec{B}}{\mu_{0}}$
On solving
$\vec{H}=-0.8 \sin (k z-\omega t) \vec{a}_{y}-1.2 \sin (k z-\omega t) \vec{a}_{x}$
16. In free space, an electromagnetic wave of 3 GHz frequency strikes over the edge of an object of size $\frac{\lambda}{100}$, where $\lambda$ is the wavelength of the wave in free space. The phenomenon, which happens there will be:
(A) Reflection
(B) Refraction
(C) Diffraction
(D) Scattering

Answer (D)
Sol. Since size is of the order of $\frac{\lambda}{100}$, hence scattering will take place.
17. An electron with speed $v$ and a photon with speed $c$ have the same de-Broglie wavelength. If the kinetic energy and momentum of electron are $E_{e}$ and $p_{\mathrm{e}}$ and that of photon are $E_{p h}$ and $p_{p h}$ respectively. Which of the following is correct?
(A) $\frac{E_{e}}{E_{p h}}=\frac{2 c}{v}$
(B) $\frac{E_{e}}{E_{p h}}=\frac{v}{2 c}$
(C) $\frac{p_{e}}{p_{p h}}=\frac{2 c}{v}$
(D) $\frac{p_{e}}{p_{p h}}=\frac{v}{2 c}$

## Answer (B)

Sol. $\lambda_{e}=\lambda_{p h} \Rightarrow \frac{h}{p_{e}}=\frac{h c}{E_{p h}}$

$$
\Rightarrow \quad E_{p h}=p_{e} \times c=2 E_{e} \frac{c}{v}
$$

$\Rightarrow \frac{E_{e}}{E_{p h}}=\frac{v}{2 c}$
18. How many alpha and beta particles are emitted when Uranium ${ }_{92} \mathrm{U}^{238}$ decays to lead ${ }_{82} \mathrm{~Pb}^{206}$ ?
(A) 3 alpha particles and 5 beta particles
(B) 6 alpha particles and 4 beta particles
(C) 4 alpha particles and 5 beta particles
(D) 8 alpha particles and 6 beta particles

## Answer (D)

Sol. ${ }_{92}^{238} \mathrm{U} \longrightarrow{ }_{82}^{206} \mathrm{~Pb}+x{ }_{2}^{4} \mathrm{He}+y\left({ }_{-1}^{0} e\right)$
$\Rightarrow 206+4 x=238$
and $82+2 x-y=92$.
$\Rightarrow x=8$ and $y=6$
19. The $I-V$ characteristics of a p-n junction diode in forward bias is shown in the figure. The ratio of dynamic resistance, corresponding to forward bias voltage of 2 V and 4 V respectively, is :

(A) $1: 2$
(B) $5: 1$
(C) $1: 40$
(D) $20: 1$

Answer (B)
Sol. Dynamic resistance $=\frac{d V}{d l}$

$$
\begin{aligned}
\Rightarrow & r_{1}=\frac{2.1-2}{10-5} \mathrm{k} \Omega \\
& \& r_{2}=\frac{4.2-4}{250-200} \mathrm{k} \Omega \\
\Rightarrow & r_{1}: r_{2}=5: 1
\end{aligned}
$$

20. Choose the correct statement for amplitude modulation :
(A) Amplitude of modulating signal is varied in accordance with the information signal.
(B) Amplitude of modulated signal is varied in accordance with the information signal.
(C) Amplitude of carrier signal is varied in accordance with the information signal.
(D) Amplitude of modulated signal is varied in accordance with the modulating signal.

## Answer (C)

Sol. In amplitude modulation, amplitude of carrier signal is varied according to the message signal.

## SECTION - B

Numerical Value Type Questions: This section contains 10 questions. In Section B, attempt any five questions out of 10 . The answer to each question is a NUMERICAL VALUE. For each question, enter the correct numerical value (in decimal notation, truncated/rounded-off to the second decimal place; e.g. $06.25,07.00,-00.33,-00.30,30.27,-27.30$ ) using the mouse and the on-screen virtual numeric keypad in the place designated to enter the answer.

1. A fighter jet is flying horizontally at a certain altitude with a speed of $200 \mathrm{~ms}^{-1}$. When it passes directly overhead an anti-aircraft gun, a bullet is fired from the gun, at an angle $\theta$ with the horizontal, to hit the jet. If the bullet speed is $400 \mathrm{~m} / \mathrm{s}$, the value of $\theta$ will be $\qquad$ $\stackrel{\circ}{\circ}$.
Answer (60)


To hit the jet
$400 \cos \theta=200$
$\Rightarrow \quad \cos \theta=\frac{1}{2}$
$\Rightarrow \theta=60^{\circ}$
2. A ball of mass 0.5 kg is dropped from the height of 10 m . The height, at which the magnitude of velocity becomes equal to the magnitude of acceleration due to gravity, is $\qquad$ m .
[Use $g=10 \mathrm{~m} / \mathrm{s}^{2}$ ]
Answer (5)

Sol. $g t=g$
$\Rightarrow t=1 \mathrm{sec}$
$\Delta h=\frac{1}{2} g t^{2}=\frac{1}{2} \times 5 \times 1^{2}=5 \mathrm{~m}$
$\therefore \quad h=H-\Delta h$

$$
=10-5
$$

$$
=5 \mathrm{~m}
$$

3. The elastic behaviour of material for linear stress and linear strain, is shown in the figure. The energy density for a linear strain of $5 \times 10^{-4}$ is $\qquad$ $\mathrm{kJ} / \mathrm{m}^{3}$. Assume that material is elastic upto the linear strain of $5 \times 10^{-4}$.


Answer (25)
Sol. $u_{d}=\frac{1}{2} \times Y \times(\text { strain })^{2}$
$=\frac{1}{2} \times\left(\frac{20}{10^{-10}}\right) \times\left(5 \times 10^{-4}\right)^{2}$
$=10^{11} \times 25 \times 10^{-8}$
$=25 \times 10^{3} \mathrm{~J} / \mathrm{m}^{3}$
$=25 \mathrm{~kJ} / \mathrm{m}^{3}$
4. The elongation of a wire on the surface of the earth is $10^{-4} \mathrm{~m}$. The same wire of same dimensions is elongated by $6 \times 10^{-5} \mathrm{~m}$ on another planet. The acceleration due to gravity on the planet will be $\qquad$ $\mathrm{ms}^{-2}$. (Take acceleration due to gravity on the surface of earth $=10 \mathrm{~ms}^{-2}$ )

## Answer (6)

Sol. $\Delta I=\frac{M^{\prime} g l}{2 \mathrm{~A} \mathrm{y}}$
$\Rightarrow \quad \Delta l \propto g$
$\Rightarrow \frac{g_{p}}{g_{e}}=\frac{\Delta I_{p}}{\Delta I_{e}}=\frac{6 \times 10^{-5}}{10 \times 10^{-5}}$
$\Rightarrow g_{p}=6 \mathrm{~m} / \mathrm{s}^{2}$ as $g_{e}=10 \mathrm{~m} / \mathrm{s}^{2}$
5. A $10 \Omega, 20 \mathrm{mH}$ coil carrying constant current is connected to a battery of 20 V through a switch. Now after switch is opened current becomes zero in $100 \mu \mathrm{~s}$. The average e.m.f. induced in the coil is
$\qquad$ V.

## Answer (400)

Sol. Initial flux through inductor $=\mathrm{LI}$

$$
\begin{aligned}
& \Rightarrow \quad \phi_{\mathrm{i}}=20 \times 10^{-3} \times \frac{20}{10} \\
& =4 \times 10^{-2} \text { weber }
\end{aligned}
$$

Final flux $=0$
$\Rightarrow$ average emf
$=\frac{\left|\phi_{i}-\phi_{f}\right|}{100 \mu \mathrm{~S}}$
$=\frac{4 \times 10^{-2}}{10^{-4}}=400 \mathrm{~V}$
6. A light ray is incident, at an incident angle $\theta_{1}$, on the system of two plane mirrors $M_{1}$ and $M_{2}$ having an inclination angle $75^{\circ}$ between them (as shown in figure). After reflecting from mirror $M_{1}$ it gets reflected back by the mirror $M_{2}$ with an angle of reflection $30^{\circ}$. The total deviation of the ray will be - degree.


## Answer (210)

Sol.


On first reflection angel of deviation is $90^{\circ}$ and on second reflection angle of deviation is $120^{\circ}$
so total deviation is $\delta=90^{\circ}+120^{\circ}=210^{\circ}$
7. In a vernier callipers, each cm on the main scale is divided into 20 equal parts. If tenth vernier scale division coincides with nineth main scale division. Then the value of vernier constant will be
$\qquad$ $\times 10^{-2} \mathrm{~mm}$.

## Answer (5)

Sol. $\mathrm{LC}=\frac{1 \mathrm{MSD}}{\mathrm{VSD}}=\frac{\frac{1}{20} \mathrm{~cm}}{10}$

$$
\begin{aligned}
& =\frac{1}{200} \mathrm{~cm} \\
& =5 \times 10^{-2} \mathrm{~mm}
\end{aligned}
$$

8. As per the given circuit, the value of current through the battery will be $\qquad$ A.


## Answer (1)

Sol. Because of diode $D_{2}$ current will not flow through it so new circuit diagram is

so $R_{\text {net }}=10 \Omega$
and $i=\frac{V}{R_{\text {net }}}=1 \mathrm{~A}$
9. A $110 \mathrm{~V}, 50 \mathrm{~Hz}, \mathrm{AC}$ source is connected in the circuit (as shown in figure). The current through the resistance $55 \Omega$, at resonance in the circuit, will be
$\qquad$ A.


## Answer (0)

Sol. At resonance $\left(\omega=\frac{1}{\sqrt{L C}}\right)$, impedance of the circuit is infinite.
$\Rightarrow$ Current through resistance $=0$.
10. An ideal fluid of density $800 \mathrm{k} \mathrm{gm}^{-3}$, flows smoothly through a bent pipe (as shown in figure) that tapers in cross-sectional area from a to $\frac{a}{2}$. The pressure difference between the wide and narrow sections of pipe is 4100 Pa . At wider section, the velocity of fluid is $\frac{\sqrt{x}}{6} \mathrm{~ms}^{-1}$ for $x=\longrightarrow$. (Given $g=10 \mathrm{~ms}^{-2}$ )


## Answer (363)

Sol. Applying Bernoulli's theorem:

$$
P_{1}+\rho g h+\frac{1}{2} \rho v^{2}=P_{2}+0+\frac{1}{2} \rho(2 v)^{2}
$$

Putting the values,
$4100=800\left\{\frac{3}{2} v^{2}-10\right\}$
$\Rightarrow v=\frac{\sqrt{363}}{6} \mathrm{~m} / \mathrm{s}$

## CHEMISTRY

## SECTION - A

Multiple Choice Questions: This section contains 20 multiple choice questions. Each question has 4 choices (1), (2), (3) and (4), out of which ONLY ONE is correct.

## Choose the correct answer :

1. A commercially sold conc. HCl is $35 \% \mathrm{HCl}$ by mass. If the density of this commercial acid is $1.46 \mathrm{~g} / \mathrm{mL}$, the molarity of this solution is :
(Atomic mass : $\mathrm{Cl}=35.5 \mathrm{amu}, \mathrm{H}=1 \mathrm{amu}$ )
(A) 10.2 M
(B) 12.5 M
(C) 14.0 M
(D) 18.2 M

## Answer (C)

Sol. Molarity $=\frac{35}{(36.5) \times \frac{100}{1.46}} \times 1000=14.0 \mathrm{M}$
2. An evacuated glass vessel weighs 40.0 g when empty, 135.0 g when filled with a liquid of density $0.95 \mathrm{~g} \mathrm{~mL}^{-1}$ and 40.5 g when filled with an ideal gas at 0.82 atm at 250 K . The molar mass of the gas in $\mathrm{g} \mathrm{mol}^{-1}$ is:
(Given : R $=0.082 \mathrm{~L} \mathrm{~atm} \mathrm{~K}^{-1} \mathrm{~mol}^{-1}$ )
(A) 35
(B) 50
(C) 75
(D) 175

## Answer (Bonus)

Sol. Weight of liquid $=135.0-40.0=95.0 \mathrm{~g}$
Volume of liquid $=\frac{95}{0.95}=100 \mathrm{ml}=0.1 \mathrm{~L}$
Weight of gas $=0.5 \mathrm{~g}$
Moles of gas $=\frac{0.5}{M}$
$P V=n R T$
$0.82 \times 0.1=\frac{0.5}{M} \times 0.082 \times 250$
$M=0.5 \times 250=125 \mathrm{~g} \mathrm{~mol}^{-1}$
3. If the radius of the $3^{\text {rd }}$ Bohr's orbit of hydrogen atom is $r_{3}$ and the radius of $4^{\text {th }}$ Bohr's orbit is $r_{4}$. Then :
(A) $r_{4}=\frac{9}{16} r_{3}$
(B) $r_{4}=\frac{16}{9} r_{3}$
(C) $r_{4}=\frac{3}{4} r_{3}$
(D) $r_{4}=\frac{4}{3} r_{3}$

## Answer (B)

Sol. $r_{n}=r_{0} \frac{n^{2}}{Z}$
$r_{4}=r_{0} \times \frac{4^{2}}{1}$
and $r_{3}=r_{0} \times \frac{3^{2}}{1}$
Therefore $\frac{r_{4}}{r_{3}}=\frac{4^{2}}{3^{2}} \Rightarrow r_{4}=\frac{16}{9} r_{3}$
4. Consider the ions/molecules

$$
\mathrm{O}_{2}^{+}, \mathrm{O}_{2}, \mathrm{O}_{2}^{-}, \mathrm{O}_{2}^{2-}
$$

For increasing bond order the correct option is:
(A) $\mathrm{O}_{2}^{2-}<\mathrm{O}_{2}^{-}<\mathrm{O}_{2}<\mathrm{O}_{2}^{+}$
(B) $\mathrm{O}_{2}^{-}<\mathrm{O}_{2}^{2-}<\mathrm{O}_{2}<\mathrm{O}_{2}^{+}$
(C) $\mathrm{O}_{2}^{-}<\mathrm{O}_{2}^{2-}<\mathrm{O}_{2}^{+}<\mathrm{O}_{2}$
(D) $\mathrm{O}_{2}^{-}<\mathrm{O}_{2}^{+}<\mathrm{O}_{2}^{2-}<\mathrm{O}_{2}$

## Answer (A)

Sol. Species
$\mathrm{O}_{2}^{+}$
Bond order
2.5
$\mathrm{O}_{2}$
$\mathrm{O}_{2}^{-}$
1.5
$\mathrm{O}_{2}^{-2}$
5. The $\left(\frac{\partial E}{\partial T}\right)_{P}$ of different types of half cells are as follows:
A
B
C
D
$1 \times 10^{-4} \quad 2 \times 10^{-4}$
$0.1 \times 10^{-4} \quad 0.2 \times 10^{-4}$
(Where $E$ is the electromotive force)

Which of the above half cells would be preferred to be used as reference electrode?
(A) A
(B) B
(C) C
(D) D

## Answer (C)

Sol. We know that

$$
\begin{aligned}
& \mathrm{E}=\mathrm{E}^{\circ}-\frac{2.303 \mathrm{RT}}{\mathrm{nF}} \log \left[\frac{1}{\text { (oxidized form) }}\right] \\
& \text { Slope }=\frac{-2.303 \mathrm{R}}{\mathrm{nF}} \log \left[\frac{1}{\text { (oxidized form) }}\right] \\
& \mathrm{T}
\end{aligned}
$$

Higher the value of slope $\left(\frac{\partial E}{\partial T}\right)_{P}$, more the deviation between $E$ and $E^{\circ}$. Thus, half cell with least value of $\left(\frac{\partial \mathrm{E}}{\partial \mathrm{T}}\right)_{\mathrm{P}}$ can be used as reference electrode.
6. Choose the correct stability order of group 13 elements in their +1 oxidation state.
(A) $\mathrm{Al}<\mathrm{Ga}<\mathrm{In}<\mathrm{TI}$
(B) $\mathrm{Tl}<\mathrm{In}<\mathrm{Ga}<\mathrm{Al}$
(C) $\mathrm{Al}<\mathrm{Ga}<\mathrm{Tl}<\mathrm{In}$
(D) $\mathrm{Al}<\mathrm{Tl}<\mathrm{Ga}<\mathrm{In}$

## Answer (A)

Sol. Due to inert pair effect, stability of +3 oxidation state decreases and that of +1 oxidation state increases for (down the group) group 13 elements.

So, the correct order of stability of group 13 elements in their +1 oxidation state is $\mathrm{Al}<\mathrm{Ga}<\mathrm{In}$ < TI.
7. Given below are two statements:

Statement I : According to the Ellingham diagram, any metal oxide with higher $\Delta G^{\circ}$ is more stable than the one with lower $\Delta \mathrm{G}^{\circ}$.

Statement II : The metal involved in the formation of oxide placed lower in the Ellingham diagram can reduce the oxide of a metal placed higher in the diagram.

In the light of the above statements, choose the most appropriate answer from the options given below:
(A) Both Statement I and Statement II are correct.
(B) Both Statement I and Statement II are incorrect.
(C) Statement I is correct but Statement II is incorrect.
(D) Statement I is incorrect but Statement II is correct.

## Answer (D)

Sol. Ellingham diagram is plot of $\Delta \mathrm{G}$ vs T .
The criterion for the feasibility of a thermal reduction is that at a given temperature Gibbs energy change of a reaction must be negative. The change in Gibbs energy, $\Delta G$ for any process at any specified temperature, is given by the equation

$$
\Delta \mathrm{G}=\Delta \mathrm{H}-\mathrm{T} \Delta \mathrm{~S}
$$

where $\Delta \mathrm{H}=$ enthalpy change and

$$
\Delta \mathrm{S}=\text { entropy change }
$$

According to the ellingham diagram, any metal oxide with higher $\Delta G^{\circ}$ has a tendency of getting reduced by the metal whose metal oxide has lower value of $\Delta G^{\circ}$.

Therefore, Statement I is incorrect but Statement II is correct.
8. Consider the following reaction:

$$
\left.2 \mathrm{HSO}_{4}^{-}(\mathrm{aq}) \xrightarrow[\text { (2) }) \text { Hydrolysis }\right]{\text { (1) }} 2 \mathrm{HSO}_{4}^{-}+2 \mathrm{H}^{+}+\mathbf{A}
$$

The dihedral angle in product $\mathbf{A}$ in its solid phase at 110 K is :
(A) $104^{\circ}$
(B) $111.5^{\circ}$
(C) $90.2^{\circ}$
(D) $111.0^{\circ}$

## Answer (C)

Sol. A should be $\mathrm{H}_{2} \mathrm{O}_{2}$
Structure of $\mathrm{H}_{2} \mathrm{O}_{2}$ is solid phase


Dihedral angle $=90.2^{\circ}$
9. The correct order of melting point is :
(A) $\mathrm{Be}>\mathrm{Mg}>\mathrm{Ca}>\mathrm{Sr}$
(B) $\mathrm{Sr}>\mathrm{Ca}>\mathrm{Mg}>\mathrm{Be}$
(C) $\mathrm{Be}>\mathrm{Ca}>\mathrm{Mg}>\mathrm{Sr}$
(D) $\mathrm{Be}>\mathrm{Ca}>\mathrm{Sr}>\mathrm{Mg}$

Answer (D)

## Sol. Element

Be
Mg
Ca
Sr
10. The correct order of melting points of hydrides of group 16 elements is:
(A) $\mathrm{H}_{2} \mathrm{~S}<\mathrm{H}_{2} \mathrm{Se}<\mathrm{H}_{2} \mathrm{Te}<\mathrm{H}_{2} \mathrm{O}$
(B) $\mathrm{H}_{2} \mathrm{O}<\mathrm{H}_{2} \mathrm{~S}<\mathrm{H}_{2} \mathrm{Se}<\mathrm{H}_{2} \mathrm{Te}$
(C) $\mathrm{H}_{2} \mathrm{~S}<\mathrm{H}_{2} \mathrm{Te}<\mathrm{H}_{2} \mathrm{Se}<\mathrm{H}_{2} \mathrm{O}$
(D) $\mathrm{H}_{2} \mathrm{Se}<\mathrm{H}_{2} \mathrm{~S}<\mathrm{H}_{2} \mathrm{Te}<\mathrm{H}_{2} \mathrm{O}$

## Answer (A)

| Sol. Hydride | M.P. |
| :--- | :--- |
| $\mathrm{H}_{2} \mathrm{O}$ | 273 K |
| $\mathrm{H}_{2} \mathrm{~S}$ | 188 K |
| $\mathrm{H}_{2} \mathrm{Se}$ | 208 K |
| $\mathrm{H}_{2} \mathrm{Te}$ | 222 K |

11. Consider the following reaction:
$\mathrm{A}+$ alkali $\rightarrow \mathrm{B}$ (Major Product)
If $B$ is an oxoacid of phosphorus with no $\mathrm{P}-\mathrm{H}$ bond, then $A$ is:
(A) White $\mathrm{P}_{4}$
(B) Red $\mathrm{P}_{4}$
(C) $\mathrm{P}_{2} \mathrm{O}_{3}$
(D) $\mathrm{H}_{3} \mathrm{PO}_{3}$

## Answer (B)

Sol. White $\mathrm{P}_{4}+$ alkali $\rightarrow \mathrm{H}_{3} \mathrm{PO}_{2}$
Red $\mathrm{P}_{4}+$ alkali $\rightarrow \mathrm{H}_{4} \mathrm{P}_{2} \mathrm{O}_{6}$

Structure of $\mathrm{H}_{4} \mathrm{P}_{2} \mathrm{O}_{6}=$


No P-H bond
12. Polar stratospheric clouds facilitate the formation of:
(A) $\mathrm{ClONO}_{2}$
(B) HOCl
(C) ClO
(D) $\mathrm{CH}_{4}$

## Answer (B)

Sol. In summer season nitrogen dioxide and methane react with chlorine monoxide and chlorine atoms forming, Chlorine sinks, preventing much ozone depletion, whereas in winter, special types of clouds called polar stratospheric clouds are formed over Antarctica. These polar stratospheric clouds provide surface on which chlorine nitrate formed gets hydrolysed to form hypochlorous acid.
$\mathrm{ClO}(\mathrm{g})+\mathrm{NO}_{2}(\mathrm{~g}) \rightarrow \mathrm{ClONO}_{2}(\mathrm{~g})$
$\mathrm{ClONO}_{2}(\mathrm{~g})+\mathrm{H}_{2} \mathrm{O}(\mathrm{g}) \rightarrow \mathrm{HOCl}(\mathrm{g})+\mathrm{HNO}_{3}(\mathrm{~g})$
13. Given below are two statements:

Statement I: In 'Lassaigne's Test', when both nitrogen and sulphur are present in an organic compound, sodium thiocyanate is formed.

Statement II: If both nitrogen and sulphur are present in an organic compound, then the excess of sodium used in sodium fusion will decompose the sodium thiocyanate formed to give NaCN and $\mathrm{Na}_{2} \mathrm{~S}$.

In the light of the above statements, choose the most appropriate answer from the options given below:
(A) Both Statement I and Statement II are correct.
(B) Both Statement I and Statement II are incorrect.
(C) Statement I is correct but Statement II is incorrect.
(D) Statement I is incorrect but Statement II is correct.

## Answer (A)

Sol. $\mathrm{NaSCN}+2 \mathrm{Na} \rightarrow \mathrm{NaCN}+\mathrm{Na}_{2} \mathrm{~S}$
14. $\left(\mathrm{C}_{7} \mathrm{H}_{5} \mathrm{O}_{2}\right)_{2} \xrightarrow{\mathrm{hv}}[\mathrm{X}] \rightarrow 2 \mathrm{C}_{6} \mathrm{H}_{5}+2 \mathrm{CO}_{2}$

Consider the above reaction and identify the intermediate ' $X$ '
(A)

(B)

(C)

(D)


## Answer (D)

Sol.

15.


Consider the above reaction sequence and identify the product $B$.
(A)

(B)

(C)

(D)


## Answer (A)

Sol.


16. Which will have the highest enol content?
(A)

(B)

(C)

(D)


Answer (C)

17. Among the following structures, which will show the most stable enamine formation?
(Where Me is $-\mathrm{CH}_{3}$ )
(A)

(B)

(C)

(D)


## Answer (C)

Sol. The most appropriate option is (C) as one group is far enough from -COOH group.
18. Which of the following sets are correct regarding polymer?
(A) Copolymer : Buna-S
(B) Condensation polymer : Nylon-6,6
(C) Fibres : Nylon-6,6
(D) Thermosetting polymer : Terylene
(E) Homopolymer : Buna-N

Choose the correct answer from given options below:
(A) (A), (B) and (C) are correct
(B) (B), (C) and (D) are correct
(C) (A), (C) and (E) are correct
(D) (A), (B) and (D) are correct

## Answer (A)

Sol. (A) Buna-S $\quad-\quad$ Copolymer
(B) Nylon-6,6 - Condensation polymer
(C) Nylon-6,6 - Fibre
(D) Terylene - Thermoplastic
(E) Buna-N - Copolymer
(A) A, B and C are correct.
19. A chemical which stimulates the secretion of pepsin is:
(A) Anti-histamine
(B) Cimetidine
(C) Histamine
(D) Zantac

## Answer (C)

Sol. Histamine stimulates the secretion of pepsin and hydrochloric acid in the stomach.


Histamine
20. Which statement is not true with respect to nitrate ion test?
(A) A dark brown ring is formed at the junction of two solutions.
(B) Ring is formed due to nitroferrous sulphate complex.
(C) The brown complex is $\left[\mathrm{Fe}\left(\mathrm{H}_{2} \mathrm{O}\right)_{5}(\mathrm{NO})\right] \mathrm{SO}_{4}$.
(D) Heating the nitrate salt with conc. $\mathrm{H}_{2} \mathrm{SO}_{4}$, light brown fumes are evolved.

## Answer (B)

Sol. Brown ring test

$$
\begin{gathered}
\mathrm{NO}_{3}^{-}+3 \mathrm{Fe}^{+2}+4 \mathrm{H}^{+} \rightarrow \mathrm{NO}+3 \mathrm{Fe}^{+3}+2 \mathrm{H}_{2} \mathrm{O} \\
{\left[\mathrm{Fe}\left(\mathrm{H}_{2} \mathrm{O}\right)_{6}\right]^{2+} \mathrm{NO} \rightarrow\left[\mathrm{Fe}\left(\mathrm{H}_{2} \mathrm{O}\right)_{5} \mathrm{NO}^{2+} \mathrm{H}_{2} \mathrm{O}\right.} \\
\text { Brown ring }
\end{gathered}
$$

## SECTION - B

Numerical Value Type Questions: This section contains 10 questions. In Section B, attempt any five questions out of 10 . The answer to each question is a NUMERICAL VALUE. For each question, enter the correct numerical value (in decimal notation, truncated/rounded-off to the second decimal place; e.g. $06.25,07.00,-00.33,-00.30,30.27,-27.30$ ) using the mouse and the on-screen virtual numeric keypad in the place designated to enter the answer.

1. For complete combustion of methanol
$\mathrm{CH}_{3} \mathrm{OH}(\mathrm{I})+\frac{3}{2} \mathrm{O}_{2}(\mathrm{~g}) \rightarrow \mathrm{CO}_{2}(\mathrm{~g})+2 \mathrm{H}_{2} \mathrm{O}(\mathrm{I})$
the amount of heat produced as measured by bomb calorimeter is $726 \mathrm{~kJ} \mathrm{~mol}^{-1}$ at $27^{\circ} \mathrm{C}$. The enthalpy of combustion for the reaction is $-\mathrm{x} \mathrm{kJ} \mathrm{mol}{ }^{-1}$, where x is $\qquad$ (Nearest integer)
(Given : $\mathrm{R}=8.3 \mathrm{JK}^{-1} \mathrm{~mol}^{-1}$ )

## Answer (727)

Sol. $\mathrm{CH}_{3} \mathrm{OH}(\mathrm{I})+\frac{3}{2} \mathrm{O}_{2}(\mathrm{~g}) \rightarrow \mathrm{CO}_{2}(\mathrm{~g})+2 \mathrm{H}_{2} \mathrm{O}(\mathrm{I})$
$\Delta H=\Delta U+\Delta n_{g} R T$
$=-726 \mathrm{~kJ}+\left(\frac{-1}{2}\right) \times 8.3 \times 300$
$\simeq-727 \mathrm{~kJ} \mathrm{~mol}^{-1}$
2. A 0.5 percent solution of potassium chloride was found to freeze at $-0.24^{\circ} \mathrm{C}$. The percentage dissociation of potassium chloride is $\qquad$ (Nearest integer)
(Molal depression constant for water is 1.80 K kg $\mathrm{mol}^{-1}$ and molar mass of KCl is $74.6 \mathrm{~g} \mathrm{~mol}^{-1}$ )

## Answer (98)

Sol. $\Delta \mathrm{T}_{\mathrm{f}}=\mathrm{i} \mathrm{K}_{\mathrm{b}} \mathrm{m}$
$\mathrm{i}=\frac{0.24 \times 99.5 \times 74.6}{1.80 \times 0.5 \times 1000}$
$=1.98$
$\alpha=\frac{\mathrm{i}-1}{\mathrm{n}-1}=\frac{0.98}{1}=0.98$
3. 50 mL of $0.1 \mathrm{M} \mathrm{CH}_{3} \mathrm{COOH}$ is being titrated against 0.1 M NaOH . When 25 mL of NaOH has been added, the pH of the solution will be $\qquad$ $\times 10^{-2}$. (Nearest integer)
(Given : $\mathrm{pK}_{\mathrm{a}}\left(\mathrm{CH}_{3} \mathrm{COOH}\right)=4.76$ )

$$
\begin{aligned}
& \log 2=0.30 \\
& \log 3=0.48
\end{aligned}
$$

$$
\begin{aligned}
& \log 5=0.69 \\
& \log 7=0.84 \\
& \log 11=1.04
\end{aligned}
$$

## Answer (476)

Sol. $\quad \mathrm{CH}_{3} \mathrm{COOH}+\mathrm{NaOH} \rightarrow \mathrm{CH}_{3} \mathrm{COONa}+\mathrm{H}_{2} \mathrm{O}$

$$
\text { at initially } \underset{\text { m moles }}{50 \times 0.1} \underset{\text { m mole }}{25 \times 0.1}
$$

at time t 2.5 m moles $0 \quad 2.5 \mathrm{~m}$ mol

$$
\begin{aligned}
& \mathrm{pH}=\mathrm{pK}_{\mathrm{a}}+\log \left|\frac{(\text { salt })}{(\text { acid })}\right| \\
& \mathrm{pH}=4.76+\log \left|\frac{2.5}{2.5}\right| \\
& \mathrm{pH}=4.76
\end{aligned}
$$

4. A flask is filled with equal moles of $A$ and $B$. The half lives of $A$ and $B$ are 100 s and 50 s respectively and are independent of the initial concentration. The time required for the concentration of $A$ to be four times that of $B$ is $\qquad$ s.
(Given : $\ln 2=0.693$ )

## Answer (200)

Sol.

|  | $A$ | $+B$ |
| :--- | :--- | :---: |
| at initially | 1 | 1 |
|  | mole | mole |

after $100 \mathrm{~s} \quad \frac{1}{2} \quad \frac{1}{4}$
after 200 s $\quad \frac{1}{4} \quad \frac{1}{16}$
Ans. 200 seconds
5. 2.0 g of $\mathrm{H}_{2}$ gas is adsorbed on 2.5 g of platinum powder at 300 K and 1 bar pressure. The volume of the gas adsorbed per gram of the adsorbent is
$\qquad$ mL .
(Given : R $=0.083 \mathrm{~L}^{\text {bar K}}{ }^{-1} \mathrm{~mol}^{-1}$ )
Answer (9960)

Sol. $\mathrm{PV}=\mathrm{nRT}$
$V=\frac{2 \times 0.083 \times 300}{2 \times 1}=24.9$ litre
$\therefore \quad$ Volume of the gas adsorbed per gram of the

$$
\begin{aligned}
\text { adsorbent } & =\frac{24.9}{2.5}=9.96 \mathrm{~L} \\
& =9960 \mathrm{ml}
\end{aligned}
$$

6. The spin-only magnetic moment value of the most basic oxide of vanadium among $\mathrm{V}_{2} \mathrm{O}_{3}, \mathrm{~V}_{2} \mathrm{O}_{4}$ and $\mathrm{V}_{2} \mathrm{O}_{5}$ is $\qquad$ B.M. (Nearest integer)

## Answer (3)

Sol. The most basic oxide among $\mathrm{V}_{2} \mathrm{O}_{3}, \mathrm{~V}_{2} \mathrm{O}_{4}$ and $\mathrm{V}_{2} \mathrm{O}_{5}$ is $\mathrm{V}_{2} \mathrm{O}_{3}$
$\mathrm{V}_{2} \mathrm{O}_{3}=\mathrm{V}^{+3}\left(\mathrm{~d}^{2}\right)$
Magnetic moment $=\sqrt{2(2+2)}=\sqrt{8}$

$$
=2.83 \approx 3
$$

7. The spin-only magnetic moment value of an octahedral complex among $\mathrm{CoCl} 3.4 \mathrm{NH}_{3}$, $\mathrm{NiCl}_{2} \cdot 6 \mathrm{H}_{2} \mathrm{O}$ and $\mathrm{PtCl}_{4} \cdot 2 \mathrm{HCl}$, which upon reaction with excess of $\mathrm{AgNO}_{3}$ gives 2 moles of AgCl is
$\qquad$ B.M. (Nearest Integer)

## Answer (3)

Sol. $\mathrm{CoCl}_{3} \cdot 4 \mathrm{NH}_{3} \xrightarrow[\text { excess }]{\mathrm{AgNO}_{3}}\left[\mathrm{Co}\left(\mathrm{NH}_{3}\right)_{4} \cdot \mathrm{Cl}_{2}\right]+\mathrm{AgCl}$

$\mathrm{PtCl}_{4} \cdot 2 \mathrm{HCl} \longrightarrow\left[\mathrm{PtCl}_{6}\right]^{4-}+\mathrm{No} \mathrm{AgCl} \mathrm{ppt}$
$\left[\mathrm{Ni}^{+2}\left(\mathrm{H}_{2} \mathrm{O}\right)_{6}\right]=\mathrm{d}^{8}=\mathrm{t}^{6}{ }_{\mathrm{g}} \mathrm{e}_{\mathrm{g}}^{2}=2$ unpaired electrons
Magnetic moment $=\sqrt{2(2+2)}$

$$
2 \sqrt{8} \approx 3
$$

8. On complete combustion 0.30 g of an organic compound gave 0.20 g of carbon dioxide and 0.10 g of water. The percentage of carbon in the given organic compound is $\qquad$ .
(Nearest Integer)

## Answer (18)

Sol. $\mathrm{C} \%=\frac{12}{44} \times \frac{0.20}{0.30} \times 100$

$$
=\frac{200}{11}=18.18 \approx 18
$$

9. Compound 'P' on nitration with dil. $\mathrm{HNO}_{3}$ yields two isomers (A) and (B) show the intramolecular and intermolecular hydrogen bonding respectively. Compound ( P ) on reaction with conc. $\mathrm{HNO}_{3}$ yields a yellow compound ' C ', a strong acid. The number of oxygen atoms is present in compound ' $C$ ' $\qquad$ .

## Answer (7)

Sol.


The number of oxygen atoms $=7$
10. The number of oxygens present in a nucleotide formed from a base, that is present only in RNA is
$\qquad$ -

## Answer (9)

Sol. Nucleotide formed by Uracil, the base present in RNA, is


The number of oxygen $=9$

## MATHEMATICS

## SECTION - A

Multiple Choice Questions: This section contains 20 multiple choice questions. Each question has 4 choices (1), (2), (3) and (4), out of which ONLY ONE is correct.

## Choose the correct answer :

1. Let $f(x)=\frac{x-1}{x+1}, x \in \mathrm{R}-\{0,-1,1\}$. If $f^{n+1}(x)=$ $f\left(f^{n}(x)\right)$ for all $n \in \mathbf{N}$, then $f^{6}(6)+f^{7}(7)$ is equal to :
(A) $\frac{7}{6}$
(B) $-\frac{3}{2}$
(C) $\frac{7}{12}$
(D) $-\frac{11}{12}$

## Answer (B)

Sol. $f(x)=\frac{x-1}{x+1} \Rightarrow f(f(x))=\frac{\frac{x-1}{x+1}-1}{\frac{x-1}{x+1}+1}=-\frac{1}{x}$
$\Rightarrow f^{3}(x)=-\frac{x+1}{x-1} \Rightarrow f^{4}(x)=-\frac{\frac{x-1}{x+1}+1}{\frac{x-1}{x+1}-1}=x$
So, $f^{6}(6)+f^{7}(7)=f^{2}(6)+f^{3}(7)$
$=-\frac{1}{6}-\frac{7+1}{7-1}=-\frac{9}{6}=-\frac{3}{2}$
2. Let $A=\left\{z \in \mathbf{C}:\left|\frac{z+1}{z-1}\right|<1\right\}$
and $B=\left\{z \in \mathbf{C}: \arg \left(\frac{z-1}{z+1}\right)=\frac{2 \pi}{3}\right\}$.
Then $A \cap B$ is :
(A) A portion of a circle centred at $\left(0,-\frac{1}{\sqrt{3}}\right)$ that lies in the second and third quadrants only
(B) A portion of a circle centred at $\left(0,-\frac{1}{\sqrt{3}}\right)$ that lies in the second quadrant only
(C) An empty set
(D) A portion of a circle of radius $\frac{2}{\sqrt{3}}$ that lies in the third quadrant only
Answer (B)

Sol. $\left|\frac{z+1}{z-1}\right|<1 \Rightarrow|z+1|<|z-1| \Rightarrow \operatorname{Re}(z)<0$
and $\arg \left(\frac{z-1}{z+1}\right)=\frac{2 \pi}{3}$ is a part of circle as shown.

3. Let $A$ be a $3 \times 3$ invertible matrix. If $|\operatorname{adj}(24 A)|=\mid a d j$ ( 3 adj $(2 A)) \mid$, then $|A|^{2}$ is equal to :
(A) $6^{6}$
(B) $2^{12}$
(C) $2^{6}$
(D) 1

Answer (C)
Sol. $|\operatorname{adj}(24 A)|=|\operatorname{adj}(3 \operatorname{adj}(2 A))|$

$$
\begin{aligned}
& \Rightarrow|24 A|^{2}=|3 \operatorname{adj}(2 A)|^{2} \\
& \Rightarrow\left(24^{3}\right)^{2} \cdot|A|^{2}=\left(3^{3}\right)^{2}|\operatorname{adj}(2 A)|^{2} \\
& \Rightarrow 24^{6} \cdot|A|^{2}=3^{6}|2 A|^{4} \\
& \Rightarrow 24^{6}|A|^{2}=3^{6} \cdot\left(2^{3}\right)^{4}|A|^{4} \\
& \Rightarrow|A|^{2}=\frac{24^{6}}{3^{6} \cdot 2^{12}}=\frac{2^{18} \cdot 3^{6}}{3^{6} \cdot 2^{12}}=2^{6}
\end{aligned}
$$

4. The ordered pair $(a, b)$, for which the system of linear equations
$3 x-2 y+z=b$
$5 x-8 y+9 z=3$
$2 x+y+a z=-1$
has no solution, is :
(A) $\left(3, \frac{1}{3}\right)$
(B) $\left(-3, \frac{1}{3}\right)$
(C) $\left(-3,-\frac{1}{3}\right)$
(D) $\left(3,-\frac{1}{3}\right)$

## Answer (C)

Sol. $\left|\begin{array}{ccc}3 & -2 & 1 \\ 5 & -8 & 9 \\ 2 & 1 & a\end{array}\right|=0 \Rightarrow-14 a-42=0 \Rightarrow a=-3$
Now 3(equation (1)) - (equation (2)) - 2(equation (3)) is

$$
\begin{aligned}
3(3 x-2 y+z-b)-(5 x-8 y+ & 9 z-3) \\
& -2(2 x+y+a z+1)=0 \\
\Rightarrow-3 b+3-2=0 \Rightarrow b= & \frac{1}{3}
\end{aligned}
$$

So for no solution $a=-3$ and $b \neq \frac{1}{3}$
5. The remainder when $(2021)^{2023}$ is divided by 7 is :
(A) 1
(B) 2
(C) 5
(D) 6

## Answer (C)

Sol. $2021 \equiv-2(\bmod 7)$

$$
\begin{aligned}
\Rightarrow(2021)^{2023} & \equiv-(2)^{2023}(\bmod 7) \\
& \equiv-2(8)^{674}(\bmod 7) \\
& \equiv-2(1)^{674}(\bmod 7) \\
& \equiv-2(\bmod 7) \\
& \equiv 5(\bmod 7)
\end{aligned}
$$

So when (2021) ${ }^{2023}$ is divided by 7 , remainder is 5 .
6. $\lim _{x \rightarrow \frac{1}{\sqrt{2}}} \frac{\sin \left(\cos ^{-1} x\right)-x}{1-\tan \left(\cos ^{-1} x\right)}$ is equal to :
(A) $\sqrt{2}$
(B) $-\sqrt{2}$
(C) $\frac{1}{\sqrt{2}}$
(D) $-\frac{1}{\sqrt{2}}$

## Answer (D)

Sol. $\lim _{x \rightarrow \frac{1}{\sqrt{2}}} \frac{\sin \left(\cos ^{-1} x\right)-x}{1-\tan \left(\cos ^{-1} x\right)} \quad$ let $\cos ^{-1} x=\frac{\pi}{4}+\theta$

$$
\begin{aligned}
& =\lim _{\theta \rightarrow 0} \frac{\sin \left(\frac{\pi}{4}+\theta\right)-\cos \left(\frac{\pi}{4}+\theta\right)}{1-\tan \left(\frac{\pi}{4}+\theta\right)} \\
& =\lim _{\theta \rightarrow 0} \frac{\sqrt{2} \sin \left(\frac{\pi}{4}+\theta-\frac{\pi}{4}\right)}{1-\frac{1+\tan \theta}{1-\tan \theta}}
\end{aligned}
$$

$$
=\lim _{\theta \rightarrow 0} \frac{\sqrt{2} \sin \theta}{-2 \tan \theta}(1-\tan \theta)=-\frac{1}{\sqrt{2}}
$$

7. $f, g: \mathbf{R} \rightarrow \mathbf{R}$ be two real valued functions defined as $f(x)=\left\{\begin{array}{cc}-|x+3|, & x<0 \\ e^{x}, & x \geq 0\end{array}\right.$ and $g(x)=\left\{\begin{array}{ll}x^{2}+k_{1} x, & x<0 \\ 4 x+k_{2}, & x \geq 0\end{array}\right.$, where $k_{1}$ and $k_{2}$ are real constants. If ( $g \circ f$ ) is differentiable at $x=0$, then $(g \circ f)(-4)+(g \circ f)(4)$ is equal to :
(A) $4\left(e^{4}+1\right)$
(B) $2\left(2 e^{4}+1\right)$
(C) $4 e^{4}$
(D) $2\left(2 e^{4}-1\right)$

## Answer (D)

Sol. $\because$ gof is differentiable at $x=0$
So R.H.D = L.H.D

$$
\begin{aligned}
& \frac{d}{d x}\left(4 e^{x}+k_{2}\right)=\frac{d}{d x}\left((-|x+3|)^{2}-k_{1}|x+3|\right) \\
\Rightarrow & 4=6-k_{1} \Rightarrow k_{1}=2
\end{aligned}
$$

Also $g\left(f\left(0^{+}\right)\right)=g\left(f\left(0^{-}\right)\right)$
$\Rightarrow 4+k_{2}=9-3 k_{1} \Rightarrow k_{2}=-1$
Now $g(f(-4))+g(f(4))$

$$
\begin{aligned}
=g(-1)+g\left(e^{4}\right) & =\left(1-k_{1}\right)+\left(4 e^{4}+k_{2}\right) \\
& =4 e^{4}-2 \\
& =2\left(2 e^{4}-1\right)
\end{aligned}
$$

8. The sum of the absolute minimum and the absolute maximum values of the function $f(x)=\left|3 x-x^{2}+2\right|$ $-x$ in the interval $[-1,2]$ is :
(A) $\frac{\sqrt{17}+3}{2}$
(B) $\frac{\sqrt{17}+5}{2}$
(C) 5
(D) $\frac{9-\sqrt{17}}{2}$

## Answer (A)

Sol. $f(x)=\left|x^{2}-3 x-2\right|-x \forall x \in[-1,2]$
$\Rightarrow f(x)=\left\{\begin{array}{l}x^{2}-4 x-2 \text { if }-1 \leq x<\frac{3-\sqrt{17}}{2} \\ -x^{2}+2 x+2 \text { if } \frac{3-\sqrt{17}}{2} \leq x \leq 2\end{array}\right.$

$f(x)_{\text {max }}=3$
$f(x)_{\min }=f\left(\frac{3-\sqrt{17}}{2}\right)$
$=\frac{\sqrt{17}-3}{2}$
9. Let $S$ be the set of all the natural numbers, for which the line $\frac{x}{a}+\frac{y}{b}=2$ is a tangent to the curve $\left(\frac{x}{a}\right)^{n}+\left(\frac{y}{b}\right)^{n}=2$ at the point $(a, b), a b \neq 0$. Then :
(A) $S=\phi$
(B) $n(S)=1$
(C) $S=\{2 k: k \in \mathbf{N}\}$
(D) $S=\mathbf{N}$

## Answer (D)

Sol. $\left(\frac{x}{a}\right)^{n}+\left(\frac{y}{b}\right)^{n}=2$
$\Rightarrow \frac{n}{a}\left(\frac{x}{a}\right)^{n-1}+\frac{n}{b}\left(\frac{y}{b}\right)^{n-1} \frac{d y}{d x}=0$
$\Rightarrow \frac{d y}{d x}=-\frac{b}{a}\left(\frac{b x}{a y}\right)^{n-1}$
$\Rightarrow \frac{d y}{d x}(a, b)=-\frac{b}{a}$
So line always touches the given curve.
10. The area bounded by the curve $y=\left|x^{2}-9\right|$ and the line $y=3$ is
(A) $4(2 \sqrt{3}+\sqrt{6}-4)$
(B) $4(4 \sqrt{3}+\sqrt{6}-4)$
(C) $8(4 \sqrt{3}+3 \sqrt{6}-9)$
(D) $8(4 \sqrt{3}+\sqrt{6}-9)$

## Answer (*)

Sol. $y=3$ and $y=\left|x^{2}-9\right|$
Intersect in first quadrant at $x=\sqrt{6}$ and $x=\sqrt{12}$


Required area

$$
\begin{aligned}
& =2\left[\frac{2}{3}(6 \times \sqrt{6})+\int_{\sqrt{6}}^{3}\left(3-\left(9-x^{2}\right)\right) d x+\int_{3}^{\sqrt{12}}\left(3-\left(x^{2}-9\right)\right) d x\right] \\
& =2\left[4 \sqrt{6}+\left.\left(\frac{x^{3}}{3}-6 x\right)\right|_{\sqrt{6}} ^{3}+\left.\left(12 x-\frac{x^{3}}{3}\right)\right|_{3} ^{\sqrt{12}}\right] \\
& =2[4 \sqrt{6}+(4 \sqrt{6}-9)+(8 \sqrt{12}-27)] \\
& =2[8 \sqrt{6}+16 \sqrt{3}-36]=8[2 \sqrt{6}+4 \sqrt{3}-9]
\end{aligned}
$$

11. Let $R$ be the point $(3,7)$ and let $P$ and $Q$ be two points on the line $x+y=5$ such that $P Q R$ is an equilateral triangle, Then the area of $\triangle P Q R$ is :
(A) $\frac{25}{4 \sqrt{3}}$
(B) $\frac{25 \sqrt{3}}{2}$
(C) $\frac{25}{\sqrt{3}}$
(D) $\frac{25}{2 \sqrt{3}}$

## Answer (D)

Sol.


Altitude of equilateral triangle,
$\frac{\sqrt{3} 1}{2}=\frac{5}{\sqrt{2}}$
$I=\frac{5 \sqrt{2}}{\sqrt{3}}$
Area of triangle $=\frac{\sqrt{3}}{4} I^{2}=\frac{\sqrt{3}}{4} \cdot \frac{50}{3}=\frac{25}{2 \sqrt{3}}$
12. Let $C$ be a circle passing through the points $A(2,-1)$ and $B(3,4)$. The line segment $A B$ is not a diameter of $C$. If $r$ is the radius of $C$ and its centre lies on the circle $(x-5)^{2}+(y-1)^{2}=\frac{13}{2}$, then $r^{2}$ is equal to :
(A) 32
(B) $\frac{65}{2}$
(C) $\frac{61}{2}$
(D) 30

## Answer (B)

Sol. Equation of perpendicular bisector of $A B$ is

$$
y-\frac{3}{2}=-\frac{1}{5}\left(x-\frac{5}{2}\right) \Rightarrow x+5 y=10
$$

Solving it with equation of given circle,

$$
\begin{aligned}
& (x-5)^{2}+\left(\frac{10-x}{5}-1\right)^{2}=\frac{13}{2} \\
\Rightarrow & (x-5)^{2}\left(1+\frac{1}{25}\right)=\frac{13}{2} \\
\Rightarrow & x-5= \pm \frac{5}{2} \Rightarrow x=\frac{5}{2} \text { or } \frac{15}{2}
\end{aligned}
$$

But $x \neq \frac{5}{2}$ because $A B$ is not the diameter.
So, centre will be $\left(\frac{15}{2}, \frac{1}{2}\right)$
Now $r^{2}=\left(\frac{15}{2}-2\right)^{2}+\left(\frac{1}{2}+1\right)^{2}$

$$
=\frac{65}{2}
$$

13. Let the normal at the point $P$ on the parabola $y^{2}=6 x$ pass through the point $(5,-8)$. If the tangent at $P$ to the parabola intersects its directrix at the point $Q$, then the ordinate of the point $Q$ is :
(A) -3
(B) $-\frac{9}{4}$
(C) $-\frac{5}{2}$
(D) -2

## Answer (B)

Sol. Let $P\left(a t^{2}, 2 a t\right)$ where $a=\frac{3}{2}$
$T: y t=x+a t^{2}$ So point $Q$ is $\left(-a, a t-\frac{a}{t}\right)$
$N: y=-t x+2 a t+a t^{3}$ passes through $(5,-8)$

$$
-8=-5 t+3 t+\frac{3}{2} t^{3}
$$

$\Rightarrow 3 t^{3}-4 t+16=0$
$\Rightarrow(t+2)\left(3 t^{2}-6 t+8\right)=0$
$\Rightarrow t=-2$
So ordinate of point $Q$ is $-\frac{9}{4}$.
14. If the two lines $I_{1}: \frac{x-2}{3}=\frac{y+1}{-2}, z=2$ and $I_{2}: \frac{x-1}{1}=\frac{2 y+3}{\alpha}=\frac{z+5}{2}$ are perpendicular, then an angle between the lines $1 / 2$ and $I_{3}: \frac{1-x}{3}=\frac{2 y-1}{-4}=\frac{z}{4}$ is :
(A) $\cos ^{-1}\left(\frac{29}{4}\right)$
(B) $\sec ^{-1}\left(\frac{29}{4}\right)$
(C) $\cos ^{-1}\left(\frac{2}{29}\right)$
(D) $\cos ^{-1}\left(\frac{2}{\sqrt{29}}\right)$

## Answer (B)

Sol. $\because \quad L_{1}$ and $L_{2}$ are perpendicular, so

$$
\begin{aligned}
& 3 \times 1+(-2)\left(\frac{\alpha}{2}\right)+0 \times 2=0 \\
\Rightarrow & \alpha=3
\end{aligned}
$$

Now angle between $l_{2}$ and $l_{3}$,

$$
\begin{aligned}
\cos \theta & =\frac{1(-3)+\frac{\alpha}{2}(-2)+2(4)}{\sqrt{1+\frac{\alpha^{2}}{4}+4 \sqrt{9+4+16}}} \\
\Rightarrow \cos \theta & =\frac{2}{\frac{29}{2}} \Rightarrow \theta=\cos ^{-1}\left(\frac{4}{29}\right)=\sec ^{-1}\left(\frac{29}{4}\right)
\end{aligned}
$$

15. Let the plane $2 x+3 y+z+20=0$ be rotated through a right angle about its line of intersection with the plane $x-3 y+5 z=8$. If the mirror image of the point $\left(2,-\frac{1}{2}, 2\right)$ in the rotated plane is $B(a, b, c)$, then :
(A) $\frac{a}{8}=\frac{b}{5}=\frac{c}{-4}$
(B) $\frac{a}{4}=\frac{b}{5}=\frac{c}{-2}$
(C) $\frac{a}{8}=\frac{b}{-5}=\frac{c}{4}$
(D) $\frac{a}{4}=\frac{b}{5}=\frac{c}{2}$

## Answer (A)

Sol. Consider the equation of plane,
$P:(2 x+3 y+z+20)+\lambda(x-3 y+5 z-8)=0$
$P:(2+\lambda) x+(3-3 \lambda) y+(1+5 \lambda) z+(20-8 \lambda)=0$
$\because$ Plane $P$ is perpendicular to $2 x+3 y+z+20=0$
So, $4+2 \lambda+9-9 \lambda+1+5 \lambda=0$

$$
\begin{aligned}
& \Rightarrow \lambda=7 \\
& P: 9 x-18 y+36 z-36=0
\end{aligned}
$$

Or $P: x-2 y+4 z=4$

If image of $\left(2,-\frac{1}{2}, 2\right)$ in plane $P$ is $(a, b, c)$ then

$$
\frac{a-2}{1}=\frac{b+\frac{1}{2}}{-2}=\frac{c-2}{4}
$$

and $\left(\frac{a+2}{2}\right)-2\left(\frac{b-\frac{1}{2}}{2}\right)+4\left(\frac{c+2}{2}\right)=4$
Clearly $a=\frac{4}{3}, b=\frac{5}{6}$ and $c=-\frac{2}{3}$
So, $a: b: c=8: 5:-4$
16. If $\vec{a} \cdot \vec{b}=1, \vec{b} \cdot \vec{c}=2$ and $\vec{c} \cdot \vec{a}=3$, then the value of
$[\vec{a} \times(\vec{b} \times \vec{c}), \vec{b} \times(\vec{c} \times \vec{a}), \vec{c} \times(\vec{b} \times \vec{a})]$ is :
(A) 0
(B) $-6 \vec{a} \cdot(\vec{b} \times \vec{c})$
(C) $12 \vec{c} \cdot(\vec{a} \times \vec{b})$
(D) $-12 \vec{b} \cdot(\vec{c} \times \vec{a})$

## Answer (A)

Sol. $\because \quad \vec{a} \times(\vec{b} \times \vec{c})=3 \vec{b}-\vec{c}=\vec{u}$

$$
\begin{aligned}
& \vec{b} \times(\vec{c} \times \vec{a})=\vec{c}-2 \vec{a}=\vec{v} \\
& \vec{c} \times(\vec{b} \times \vec{a})=3 \vec{b}-2 \vec{a}=\vec{w} \\
\therefore \quad & \vec{u}+\vec{v}=\vec{w}
\end{aligned}
$$

So vectors $\vec{u}, \vec{v}$ and $\vec{w}$ are coplanar, hence their Scalar triple product will be zero.
17. Let a biased coin be tossed 5 times. If the probability of getting 4 heads is equal to the probability of getting 5 heads, then the probability of getting atmost two heads is:
(A) $\frac{275}{6^{5}}$
(B) $\frac{36}{5^{4}}$
(C) $\frac{181}{5^{5}}$
(D) $\frac{46}{6^{4}}$

## Answer (D)

Sol. Let probability of getting head $=p$
So, ${ }^{5} C_{4} p^{4}(1-p)={ }^{5} C_{5} p^{5}$
$\Rightarrow p=5(1-p) \Rightarrow p=\frac{5}{6}$

Probability of getting atmost two heads =
${ }^{5} C_{0}(1-p)^{5}+{ }^{5} C_{1} p(1-p)^{4}+{ }^{5} C_{2} p^{2}(1-p)^{3}$
$=\frac{1+25+250}{6^{5}}$
$=\frac{276}{6^{5}}=\frac{46}{6^{4}}$
18. The mean of the numbers $a, b, 8,5,10$ is 6 and their variance is 6.8 . If $M$ is the mean deviation of the numbers about the mean, then 25 M is equal to:
(A) 60
(B) 55
(C) 50
(D) 45

Answer (A)
Sol. $\because \bar{x}=6=\frac{a+b+8+5+10}{5} \Rightarrow a+b=7 \ldots$ (i)
And $\sigma^{2}=\frac{a^{2}+b^{2}+8^{2}+5^{2}+10^{2}}{5}-6^{2}=6.8$
$\Rightarrow a^{2}+b^{2}=25$
From (i) and (ii) $(a, b)=(3,4)$ or $(4,3)$
Now mean deviation about mean

$$
\begin{aligned}
& M=\frac{1}{5}(3+2+2+1+4)=\frac{12}{5} \\
\Rightarrow & 25 M=60
\end{aligned}
$$

19. Let $f(x)=2 \cos ^{-1} x+4 \cot ^{-1} x-3 x^{2}-2 x+10, x \in[-1,1]$, If $[a, b]$ is the range of the function, $f$ then $4 a-b$ is equal to :
(A) 11
(B) $11-\pi$
(C) $11+\pi$
(D) $15-\pi$

## Answer (B)

Sol. $f(x)=2 \cos ^{-1} x+4 \cot ^{-1} x-3 x^{2}-2 x+10 \forall x \in[-1,1]$
$\Rightarrow f^{\prime}(x)=-\frac{2}{\sqrt{1-x^{2}}}-\frac{4}{1+x^{2}}-6 x-2<0 \quad \forall x \in[-1,1]$
So $f(x)$ is decreasing function and range of $f(x)$ is $[f(1), f(-1)]$, which is $[\pi+5,5 \pi+9]$

Now $4 a-b=4(\pi+5)-(5 \pi+9)$
$=11-\pi$
20. Let $\Delta, \nabla \in\{\wedge, v\}$ be such that $p \nabla q \Rightarrow((p \Delta q) \nabla r)$ is a tautology. Then $(p \nabla q) \Delta r$ is logically equivalent to :
(A) $(p \Delta r) \vee q$
(B) $(p \Delta r) \wedge q$
(C) $(p \wedge r) \Delta q$
(D) $(p \nabla r) \wedge q$

## Answer (A)

## Sol. Case-I

If $\nabla$ is same as $\wedge$
Then $(p \wedge q) \Rightarrow((p \Delta q) \wedge r)$ is equivalent to $\sim(p \wedge q) \vee$ $((p \Delta q) \wedge r)$ is equivalent to $(\sim(p \wedge q) \vee(p \Delta q)) \wedge(\sim(p \wedge$ q) $\vee r)$

Which cannot be a tautology
For both $\Delta($ i.e. $\vee$ or $\wedge)$

## Case-II

If $\nabla$ is same as $v$
Then $(p \vee q) \Rightarrow((p \Delta q) \vee r)$ is equivalent to
$\sim(p \vee q) \vee(p \Delta q) \vee r$ which can be a tautology if $\Delta$ is also same as $\vee$.
Hence both $\Delta$ and $\nabla$ are same as $v$.
Now $(p \nabla q) \Delta r$ is equivalent to $(p \vee q \vee r)$.

## SECTION - B

Numerical Value Type Questions: This section contains 10 questions. In Section B, attempt any five questions out of 10. The answer to each question is a NUMERICAL VALUE. For each question, enter the correct numerical value (in decimal notation, truncated/rounded-off to the second decimal place; e.g. $06.25,07.00,-00.33,-00.30,30.27,-27.30$ ) using the mouse and the on-screen virtual numeric keypad in the place designated to enter the answer.

1. The sum of the cubes of all the roots of the equation $x^{4}-3 x^{3}-2 x^{2}+3 x+1=0$ is $\qquad$ .

## Answer (36)

Sol. $x^{4}-3 x^{3}-x^{2}-x^{2}+3 x+1=0$

$$
\left(x^{2}-1\right)\left(x^{2}-3 x-1\right)=0
$$

Let the root of $x^{2}-3 x-1=0$ be $\alpha$ and $\beta$ and other two roots of given equation are 1 and -1
So sum of cubes of roots $=1^{3}+(-1)^{3}+\alpha^{3}+\beta^{3}$

$$
\begin{aligned}
& =(\alpha+\beta)^{3}-3 \alpha \beta(\alpha+\beta) \\
& =(3)^{3}-3(-1)(3) \\
& =36
\end{aligned}
$$

2. There are ten boys $B_{1}, B_{2}, \ldots, B_{10}$ and five girls $G_{1}$, $G_{2}, \ldots, G_{5}$ in a class. Then the number of ways of forming a group consisting of three boys and three girls, if both $B_{1}$ and $B_{2}$ together should not be the members of a group, is $\qquad$ -.

## Answer (1120)

Sol. Required number of ways = Total ways of selection - ways in which $B_{1}$ and $B_{2}$ are present together.

$$
\begin{aligned}
={ }^{10} C_{3} \cdot{ }^{5} C_{3}-{ }^{8} C_{1} \cdot{ }^{5} C_{3} & =10(120-8) \\
& =1120
\end{aligned}
$$

3. Let the common tangents to the curves $4\left(x^{2}+y^{2}\right)=9$ and $y^{2}=4 x$ intersect at the point $Q$. Let an ellipse, centered at the origin $O$, has lengths of semi-minor and semi-major axes equal to $O Q$ and 6 , respectively. If $e$ and $/$ respectively denote the eccentricity and the length of the latus rectum of this ellipse, then $\frac{l}{e^{2}}$ is equal to $\qquad$ .

## Answer (4)

Sol. Let $y=m x+c$ is the common tangent
So $c=\frac{1}{m}= \pm \frac{3}{2} \sqrt{1+m^{2}} \Rightarrow m^{2}=\frac{1}{3}$
So equation of common tangents will be $y= \pm \frac{1}{\sqrt{3}} x \pm \sqrt{3}$, which intersects at $Q(-3,0)$
Major axis and minor axis of ellipse are 12 and 6. So eccentricity
$e^{2}=1-\frac{1}{4}=\frac{3}{4}$ and length of latus rectum $=\frac{2 b^{2}}{a}=3$
Hence $\frac{\ell}{e^{2}}=\frac{3}{3 / 4}=4$
4. Let $f(x)=\max \{|x+1|,|x+2|, \ldots \ldots . .,|x+5|\}$. Then $\int_{-6}^{0} f(x) d x$ is equal to $\qquad$ .

## Answer (21)

Sol.


$$
\int_{-6}^{0} f(x) d x=2\left[\frac{1}{2}(2+5) 3\right]=21
$$

5. Let the solution curve $y=y(x)$ of the differential equation $\left(4+x^{2}\right) d y-2 x\left(x^{2}+3 y+4\right) d x=0$ pass through the origin. Then $y(2)$ is equal to $\qquad$ .

## Answer (12)

Sol. $\left(4+x^{2}\right) d y-2 x\left(x^{2}+3 y+4\right) d x=0$
$\Rightarrow \frac{d y}{d x}=\left(\frac{6 x}{x^{2}+4}\right) y+2 x$
$\Rightarrow \frac{d y}{d x}-\left(\frac{6 x}{x^{2}+4}\right) y=2 x$
I.F. $=e^{-3 \ln \left(x^{2}+4\right)}=\frac{1}{\left(x^{2}+4\right)^{3}}$

So $\frac{y}{\left(x^{2}+4\right)^{3}}=\int \frac{2 x}{\left(x^{2}+4\right)^{3}} d x+c$
$\Rightarrow \quad y=-\frac{1}{2}\left(x^{2}+4\right)+c\left(x^{2}+4\right)^{3}$
When $x=0, y=0$ gives $c=\frac{1}{32}$,
So, for $x=2, y=12$
6. If $\sin ^{2}\left(10^{\circ}\right) \sin \left(20^{\circ}\right) \sin \left(40^{\circ}\right) \sin \left(50^{\circ}\right) \sin \left(70^{\circ}\right)$ $=\alpha-\frac{1}{16} \sin \left(10^{\circ}\right)$, then $16+\alpha^{-1}$ is equal to
$\qquad$ .

## Answer (80)

Sol: $\left(\sin 10^{\circ} \cdot \sin 50^{\circ} \cdot \sin 70^{\circ}\right) \cdot\left(\sin 10^{\circ} \cdot \sin 20^{\circ} \cdot \sin 40^{\circ}\right)$

$$
\begin{aligned}
& =\left(\frac{1}{4} \sin 30^{\circ}\right) \cdot\left[\frac{1}{2} \sin 10^{\circ}\left(\cos 20^{\circ}-\cos 60^{\circ}\right)\right] \\
& =\frac{1}{16}\left[\sin 10^{\circ}\left(\cos 20^{\circ}-\frac{1}{2}\right)\right] \\
& =\frac{1}{32}\left[2 \sin 10^{\circ} \cdot \cos 20^{\circ}-\sin 10^{\circ}\right] \\
& =\frac{1}{32}\left[\sin 30^{\circ}-\sin 10^{\circ}-\sin 10^{\circ}\right] \\
& =\frac{1}{64}-\frac{1}{16} \sin 10^{\circ}
\end{aligned}
$$

Clearly $\alpha=\frac{1}{64}$
Hence $16+\alpha^{-1}=80$
7. Let $A=\{n \in \mathbf{N}:$ H.C.F. $(n, 45)=1\}$ and

Let $B=\{2 k: k \in\{1,2, \ldots, 100\}\}$. Then the sum of all the elements of $A \cap B$ is $\qquad$ .

## Answer (5264)

Sol: Sum of all elements of $A \cap B=2$ [Sum of natural numbers upto 100 which are neither divisible by 3 nor by 5 ]

$$
\begin{aligned}
& =2\left[\frac{100 \times 101}{2}-3\left(\frac{33 \times 34}{2}\right)-5\left(\frac{20 \times 21}{2}\right)+15\left(\frac{6 \times 7}{2}\right)\right] \\
& =10100-3366-2100+630 \\
& =5264
\end{aligned}
$$

8. The value of the integral
$\frac{48}{\pi^{4}} \int_{0}^{\pi}\left(\frac{3 \pi x^{2}}{2}-x^{3}\right) \frac{\sin x}{1+\cos ^{2} x} d x$ is equal to
$\qquad$ .

## Answer (6)

Sol: $I=\frac{48}{\pi^{4}} \int_{0}^{\pi}\left[\left(\frac{\pi}{2}-x\right)^{3}-\frac{3 \pi^{2}}{4}\left(\frac{\pi}{2}-x\right)+\frac{\pi^{3}}{4}\right] \frac{\sin x d x}{1+\cos ^{2} x}$
Using $\int_{a}^{b} f(x) d x=\int_{a}^{b} f(a+b-x) d x$ we get

$$
I=\frac{48}{\pi^{4}} \int_{0}^{\pi}\left[-\left(\frac{\pi}{2}-x\right)^{3}+\frac{3 \pi^{2}}{4}\left(\frac{\pi}{2}-x\right)+\frac{\pi^{3}}{4}\right] \frac{\sin x d x}{1+\cos ^{2} x}
$$

Adding these two equations, we get

$$
\begin{aligned}
& 2 I=\frac{48}{\pi^{4}} \int_{0}^{\pi} \frac{\pi^{3}}{2} \cdot \frac{\sin x d x}{1+\cos ^{2} x} \\
\Rightarrow \quad & I=\frac{12}{\pi}\left[-\tan ^{-1}(\cos x)\right]_{0}^{\pi}=\frac{12}{\pi} \cdot \frac{\pi}{2}=6
\end{aligned}
$$

9. Let $A=\sum_{i=1}^{10} \sum_{j=1}^{10} \min \{i, j\}$ and $B=\sum_{i=1}^{10} \sum_{j=1}^{10} \max \{i, j\}$.

Then $A+B$ is equal to $\qquad$ -.

## Answer (1100)

Sol: Each element of ordered pair $\{i, j\}$ is either present in $A$ or in $B$.

So, $A+B=$ Sum of all elements of all ordered pairs
$\{i, \beta\}$ for $1 \leq i \leq 10$ and $1 \leq j \leq 10$
$=20(1+2+3+\ldots+10)$
$=1100$
10. Let $S=(0,2 \pi)-\left\{\frac{\pi}{2}, \frac{3 \pi}{4}, \frac{3 \pi}{2}, \frac{7 \pi}{4}\right\}$. Let $y=y(x), x \in$ $S$, be the solution curve of the differential equation $\frac{d y}{d x}=\frac{1}{1+\sin 2 x}, y\left(\frac{\pi}{4}\right)=\frac{1}{2}$. If the sum of abscissas of all the points of intersection of the curve $y=y(x)$ with the curve $y=\sqrt{2} \sin x$ is $\frac{k \pi}{12}$, then $k$ is equal to $\qquad$ .

## Answer (42)

Sol: $\frac{d y}{d x}=\frac{1}{1+\sin 2 x}$

$$
\begin{aligned}
& \Rightarrow \quad d y=\frac{\sec ^{2} x d x}{(1+\tan x)^{2}} \\
& \Rightarrow \quad y=-\frac{1}{1+\tan x}+c
\end{aligned}
$$

When $x=\frac{\pi}{4}, \quad y=\frac{1}{2}$ gives $c=1$
So $y=\frac{\tan x}{1+\tan x} \Rightarrow y=\frac{\sin x}{\sin x+\cos x}$
Now, $y=\sqrt{2} \sin x \Rightarrow \sin x=0$
or $\sin x+\cos x=\frac{1}{\sqrt{2}}$
$\sin x=0$ gives $x=\pi$ only.
and $\sin x+\cos x=\frac{1}{\sqrt{2}} \Rightarrow \sin \left(x+\frac{\pi}{4}\right)=\frac{1}{2}$
So $x+\frac{\pi}{4}=\frac{5 \pi}{6}$ or $\frac{13 \pi}{6} \Rightarrow x=\frac{7 \pi}{12}$ or $\frac{23 \pi}{12}$
Sum of all solutions $=\pi+\frac{7 \pi}{12}+\frac{23 \pi}{12}=\frac{42 \pi}{12}$
Hence $k=42$.

