24/01/2023 Morning



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Answers & Solutions

Time : 3 hrs.



M.M.: 300

JEE (Main)-2023 (Online) Phase-1

(Physics, Chemistry and Mathematics)

IMPORTANT INSTRUCTIONS:

- (1) The test is of **3 hours** duration.
- (2) The Test Booklet consists of 90 questions. The maximum marks are 300.
- (3) There are three parts in the question paper consisting of Physics, Chemistry and Mathematics having 30 questions in each part of equal weightage. Each part (subject) has two sections.
 - Section-A: This section contains 20 multiple choice questions which have only one correct answer. Each question carries 4 marks for correct answer and -1 mark for wrong answer.
 - (ii) Section-B: This section contains 10 questions. In Section-B, attempt any five questions out of 10. The answer to each of the questions is a numerical value. Each question carries 4 marks for correct answer and -1 mark for wrong answer. For Section-B, the answer should be rounded off to the nearest integer.



PHYSICS

SECTION - A

Multiple Choice Questions: This section contains 20 multiple choice questions. Each question has 4 choices (1), (2), (3) and (4), out of which **ONLY ONE** is correct.

Choose the correct answer:

1. A travelling wave is described by the equation

 $y(x, t) = [0.05 \sin(8x - 4t)]$ m

The velocity of the wave is : [all the quantities are in SI unit]

- (1) 8 ms⁻¹ (2) 0.5 ms⁻¹
- (3) 4 ms^{-1} (4) 2 ms^{-1}

Answer (2)

Sol. ::
$$y(x, t) = [0.05 \sin(8x - 4t)]$$
 m

Speed of wave = $\left| \frac{\text{Coefficient of } t}{\text{Coefficient of } x} \right|$ = $\frac{4}{8} = 0.5 \text{ ms}^{-1}$

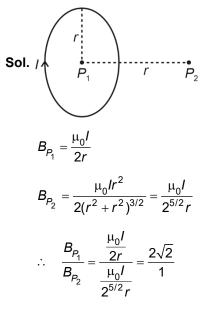
2. A circular loop of radius *r* is carrying current I A. The ratio of magnetic field at the center of circular loop and at a distance *r* from the center of the loop on its axis is:

(4) 1: $\sqrt{2}$

(1) $2\sqrt{2}:1$ (2) $1:3\sqrt{2}$

(3)
$$3\sqrt{2}$$
:2

Answer (1)



3. Consider the following radioactive decay process ${}^{218}_{84}A \xrightarrow{\alpha} A_1 \xrightarrow{\beta^-} A_2 \xrightarrow{\gamma} A_3 \xrightarrow{\alpha} A_4 \xrightarrow{\beta^+} A_5 \xrightarrow{\gamma} A_6$ The mass number and the atomic number of A₆ are given by:

(1) 210 and 80 (2) 210 and 82

(3) 210 and 84 (4) 211 and 80

Answer (1)

Sol. ${}^{218}_{84}A \xrightarrow{\alpha} {}^{214}_{82}A_1 \xrightarrow{\beta^-} {}^{214}_{83}A_2 \xrightarrow{\gamma} {}^{214}_{83}A_3 \xrightarrow{\alpha} {}^{210}_{81}A_4 \xrightarrow{\beta^+} {}^{210}_{80}A_5 \xrightarrow{\gamma} {}^{210}_{80}A_6$

Mass number = 210

Atomic number = 80

4. The weight of a body at the surface of earth is 18 N. The weight of the body at an altitude of 3200 km above the earth's surface is (given, radius of earth R_e = 6400 km):

(1) 8 N	(2)	19.6 N
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Answer (1)

$$mg_{\text{earth}} = 18$$

Also
$$mg_h = mg_{\text{earth}} \left(\frac{R}{R+h}\right)^2$$
$$= 18 \left(\frac{6400}{6400+3200}\right)^2$$

$$=18 \times \frac{4}{9} = 8$$

5. Given below are two statements : one is labelled as **Assertion A** and the other is labelled as **Reason R**.

Assertion A : Photodiodes are preferably operated in reverse bias condition for light intensity measurement.

Reason R : The current in the forward bias is more than the current in the reverse bias for a p - n junction diode.

In the light of the above statements, choose the *correct* answer from the options given below:

- (1) Both **A** and **R** are true and **R** is the correct explanation of **A**
- (2) **A** is false but **R** is true
- (3) \mathbf{A} is true but \mathbf{R} is false
- (4) Both **A** and **R** are true but **R** is **NOT** the correct explanation of **A**

Answer (4)

- **Sol.** Photodiodes are preferably operated in reverse bias condition for light intensity measurement because it increases the width of depletion layer, therefore both are correct but not the correct explanation.
- 6. Given below are two statements :

Statement I : If the Brewster's angle for the light propagating from air to glass is θ_B , then the Brewster's angle for the light propagating from

glass to air is $\frac{\pi}{2} - \theta_B$

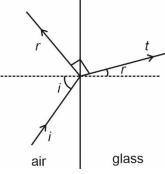
Statement II : The Brewster's angle for the light propagating from glass to air is $\tan^{-1}(\mu_g)$ where μ_g is the refractive index of glass.

In the light of the above statements, choose the **correct** answer from the options given below:

- (1) Both statement I and Statement II are true
- (2) Both statement I and statement II are false
- (3) Statement I is true but statement II is false
- (4) Statement I is false but statement II is true

Answer (3)

Sol. Case I :



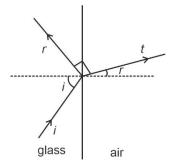
Transmitted is \perp to reflected. *i* + *r* = 90° Snell's law

 $\mu_a \sin i = \mu_g \sin r$

$$\tan i = \frac{\mu_g}{\mu_a}$$

$$i = \tan^{-1}\left(\frac{\mu_g}{\mu_a}\right) = \theta_B$$

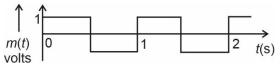
Case II :



 $i + r = 90^{\circ}$ as transmitted is \perp to reflected.

$$\tan i = \frac{\mu_a}{\mu_g} \Longrightarrow i = \tan^{-1} \frac{\mu_a}{\mu_g} = \frac{\pi}{2} - \theta_B$$

7. A modulating signal is a square wave, as shown in the figure.



If the carrier wave is given as $c(t) = 2 \sin (8\pi t)$ volts, the modulation index is:

(1) $\frac{1}{2}$ (2) $\frac{1}{4}$ (3) 1 (4) $\frac{1}{2}$

Answer (1)

Sol. A_m = Amplitude of modulating wave = 1 volt

 A_c = Amplitude of carrier wave = 2 volt

Modulation index =
$$\frac{A_m}{A_c} = \frac{1}{2} = \frac{1}{2}$$

8. As shown in the figure, a network of resistors is connected to a battery of 24V with an internal resistance of 3Ω . The currents through the resistors R_4 and R_5 are l_4 and l_5 respectively. The values of l_4 and l_5 are:

$$R_{1} = 2\Omega$$

$$R_{3} = 2\Omega$$

$$R_{4} = 20\Omega$$

$$R_{5} = 2\Omega$$

$$R_{5} = 5\Omega$$

$$R_{5} = 100$$





Sol.
$$R_{eq} = \frac{R_1 R_2}{R_1 + R_2} + R_3 + \frac{R_4 R_5}{R_5 + R_4} + R_6 + r$$

 $= (1 + 2 + 4 + 2 + 3)$
 $= 12 \Omega$
 $I = \frac{24}{12} = 2 A$
 $I_4 = \left(\frac{R_5}{R_4 + R_5}\right)I = \frac{5}{25} \times 2 = \frac{2}{5}A$
 $I_5 = \frac{R_4 I}{R_4 + R_5} = \frac{20}{25} \times I = \frac{8}{5}A$

9. Given below are two statements :

> **Statement I**: The temperature of a gas is -73°C. When the gas is heated to 527°C, the root mean square speed of the molecules is doubled.

> Statement II: The product of pressure and volume of an ideal gas will be equal to translational kinetic energy of the molecules.

In the light of the above statements, choose the **correct** answer from the options given below:

- (1) Both statement I and statement II are true
- (2) Statement I is false but statement II is true
- (3) Both statement I and statement II are false
- (4) Statement I is true but statement II is false

Answer (4)

Sol.	$T_{i} = 200 K$	$v_{\rm rms} \propto \sqrt{T}$
	$T_{f} = 800 \text{K}$	

$$\frac{V_i}{V_f} = \sqrt{\frac{T_i}{T_f}} = \sqrt{\frac{200}{800}} = \sqrt{\frac{1}{4}} = \frac{1}{2}$$
$$V_f = 2V_i$$

Translational K.E. = $\left(\frac{3}{2}PV\right)$

10. Two long straight wires *P* and *Q* carrying equal current 10A each were kept parallel to each other at 5 cm distance. Magnitude of magnetic force experienced by 10 cm length of wire P is F_1 . If distance between wires is halved and currents on them are doubled, force F_2 on 10 cm length of wire P will be:

(1) 10 <i>F</i> ₁	(2) 8 <i>F</i> ₁
(3) $\frac{F_1}{10}$	(4) $\frac{F_1}{8}$

Sol.
$$\frac{F_1}{I} = \frac{\mu_0 i_1}{2\pi r} i_2 = \frac{\mu_0 i_1 i_2}{2\pi r}$$

If i_1 and i_2 both are doubled, if $r_f = \left(\frac{r}{2}\right)$
 $\frac{F_f}{I} = \frac{\mu_0(2i_1)(2i_2)}{2\pi \left(\frac{r}{2}\right)} = \frac{8\mu_0 i_1 i_2}{2\pi r}$
 $F_f = 8F_1$

11. A conducting circular loop of radius $\frac{10}{\sqrt{\pi}}$ cm is placed perpendicular to a uniform magnetic field of 0.5 T. The magnetic field is decreased to zero in 0.5 s at a steady rate. The induced emf in the circular loop at 0.25 s is:

Answer (4)

Sol.

$$\varepsilon_{ind} = \left| -\frac{d\phi}{dt} \right|$$

$$= \left(\frac{dB}{dt} \right) A$$

$$= \left(\frac{0.5}{0.5} \right) \pi \left(\frac{1}{10\sqrt{\pi}} \right)^2$$

$$= 1 \times \frac{1}{100} V$$

$$= 0.01 V$$

$$= 10 \text{ mV}$$

12. 1 g of a liquid is converted to vapour at $3 \times 10^5 P_a$ pressure. If 10% of the heat supplied is used for increasing the volume by 1600 cm³ during this phase change, then the increase in internal energy in the process will be:

(1) 4800 J	(2) 432000 J
(3) 4.32 × 10 ⁸ J	(4) 4320 J

Answer (4)

. . . .

Sol.
$$\Delta Q = \Delta U + \Delta W$$

 $10 \ \Delta W = \Delta W + \Delta W$
 $\Delta U = 9 \ \Delta W$
 $= 9 \times (3 \times 10^5) \left[1600 \times \frac{1}{10^6} \right]$
 $= 4320 \ J$

 If two charges q1 and q2 are separated with distance 'd' and placed in a medium of dielectric constant K. What will be the equivalent distance between charges in air for the same electrostatic force?

 q_2

(1) $2d\sqrt{k}$ (2) $d\sqrt{k}$

(3)
$$1.5d\sqrt{k}$$
 (4) $k\sqrt{a}$

Answer (2)

Sol. 9₁

dielectric constant = K

$$F_{\text{medium}} = \frac{1}{4\pi(\text{K}\varepsilon_0)} \times \frac{q_1q_2}{d^2}$$

$$\therefore \quad F_{\text{air}} = F_{\text{medium}}$$

$$1 \quad q_1q_2 \qquad 1 \quad q$$

$$4\pi\varepsilon_0 (d_{air})^2 = 4\pi(K\varepsilon_0)$$

$$\therefore \quad \boxed{d_{air} = \sqrt{K}d}$$

14. Given below are two statements:

Statement I: An elevator can go up or down with uniform speed when its weight is balanced with the tension of its cable.

Statement II: Force exerted by the floor of an elevator on the foot of a person standing on it is more than his/her weight when the elevator goes down with increasing speed.

In the light of the above statements, choose the *correct* answer from the options given below:

- (1) Statement I is true but Statement II is false
- (2) Both Statement I and Statement II are true
- (3) Statement I is false but Statement II is true
- (4) Both statement I and Statement II are false

Answer (1)

Sol. Statement I is correct, because lift is moving with zero acceleration.

Statement II is incorrect as force exerted will be less than the weight.

15. In \vec{E} and \vec{K} represent electric field and propagation vectors of the EM waves in vacuum, then magnetic field vector is given by:

 $(\omega - angular frequency)$:

(1)
$$\frac{1}{\omega}(\overline{K} \times \overline{E})$$
 (2) $\omega(\overline{E} \times \overline{K})$

(3)
$$\overline{K} \times \overline{E}$$
 (4) $\omega(\overline{K} \times \overline{E})$

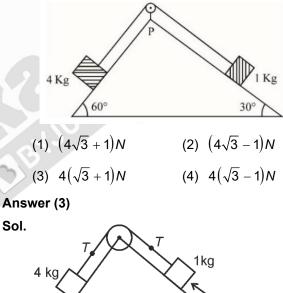
Answer (1)

Sol. $\vec{E} \Rightarrow$ electric field

 $\vec{K} \Rightarrow$ propagation vector

$$\vec{B} = \frac{1}{\omega} (\vec{K} \times \vec{E})$$

16. As per given figure, a weightless pulley P is attached on a double inclined frictionless surfaces. The tension in the string (massless) will be (if $g = 10 \text{ m/s}^2$)



Let's consider *T* tension in string and *a* acceleration of blocks.

FBD of 4kg







$$\frac{4g\sqrt{3}}{2} - T = 4a \qquad \dots (1)$$

$$T - \alpha \sin 30^\circ = a$$

$$T-\frac{g}{2}=a$$

From (1) and (2)

$$T = 4(1+\sqrt{3})N$$

17. Match List I with List II

	LIST I		LIST II
Α.	Planck's constant (h)	I.	[M ¹ L ² T ⁻²]
В.	Stopping potential (V _s)	II.	[M ¹ L ¹ T ⁻¹]
C.	Work function (ϕ)	III.	[M ¹ L ² T ⁻¹]
D.	Momentum (p)	IV.	[M ¹ L ² T ⁻³ A ⁻¹]

...(2)

Choose the correct answer from the options given below:

- (1) A-III, B-IV, C-I, D-II
- (2) A-I, B-III, C-IV, D-II
- (3) A-II, B-IV, C-III, D-I
- (4) A-III, B-I, C-II, D-IV

Answer (1)

Sol. $[h] = [ML^2T^{-1}] \rightarrow A-III$

 $[V_s] = [ML^2T^{-3}A^{-1}] \rightarrow B\text{-IV}$

 $[\phi] = [\mathsf{ML}^2\mathsf{T}^{-2}] \to \mathsf{C}\text{-}\mathsf{I}$

- $[p] = [MLT^{-1}] \rightarrow D\text{-II}$
- 18. From the photoelectric effect experiment, following observations are made. Identify which of these are correct.
 - A. The stopping potential depends only on the work function of the metal.
 - B. The saturation current increases as the intensity of incident light increases.
 - C. The maximum kinetic energy of a photo electron depends on the intensity of the incident light.
 - D. Photoelectric effect can be explained using wave theory of light.

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Choose the correct answer from the options given below:

- (1) A, B, D only
- (2) A, C, D only
- (3) B only
- (4) B, C only

Answer (3)

Sol. (A) From Einstein's equation

 $K_{\max} = eV_s = hv - \phi$

Form the stopping potential (Vs) depends on φ & v.

- (B) Saturation current is proportional to intensity, *i.e.*, number of incident photons.
- (C) K_{max} only depends on nature of photon and ϕ .
- (D) Einstein used particle behaviour of photon to explain photon electric effect.

Only B is correct.

- 19. The maximum vertical height to which a man can throw a ball is 136 m. The maximum horizontal distance upto which he can throw the same ball is:
 - (1) 272 m
 - (2) 68 m
 - (3) 192 m
 - (4) 136 m

Answer (1)

Sol. For vertical throw,

$$h = \frac{v^2}{2g}$$

v =

$$\sqrt{2gh} = \sqrt{2g \times 136}$$
 ...(1)

For max range, $\theta = 45^{\circ}$

$$R_{\max} = \frac{v^2}{g} \qquad \dots (2)$$

From (1) and (2)

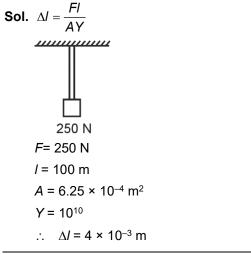
$$R_{\max} = \frac{v^2}{g} = \frac{2g \times 136}{g}$$

= 272 m

20. A 100 m long wire having cross-sectional area $6.25 \times 10^{-4} \text{ m}^2$ and Young's modulus is 10^{10} Nm^{-2} is subjected to a load of 250 N, then the elongation in the wire will be:

(1) 6.25 × 10 ^{−6} m	(2) 6.25 × 10⁻³ m
(3) 4 × 10 ^{−3} m	(4) 4 × 10 ^{−4} m

Answer (3)



SECTION - B

Numerical Value Type Questions: This section contains 10 questions. In Section B, attempt any five questions out of 10. The answer to each question is a **NUMERICAL VALUE.** For each question, enter the correct numerical value (in decimal notation, truncated/rounded-off to the second decimal place; e.g. 06.25, 07.00, -00.33, -00.30, 30.27, -27.30) using the mouse and the on-screen virtual numeric keypad in the place designated to enter the answer.

21. A hole is drilled in a metal sheet. At 27°C, the diameter of hole is 5 cm. When the sheet is heated to 177°C, the change in the diameter of hole is $d \times 10^{-3}$ cm. The value of *d* will be ______ if coefficient of linear expansion of the metal is 1.6×10^{-5} /°C.

Answer (12)

22.

Sol. $\Delta D = D\alpha \Delta t$ = 5 × 1.6 × 10⁻⁵ (177 – 27) = 0.012 cm = 12 × 10⁻³ cm so, d = 12

> A block of a mass 2 kg is attached with two identical springs of spring constant 20 N/m each. The block is placed on a frictionless surface and the ends of the springs are attached to rigid supports (see figure). When the mass is displaced from its equilibrium position, it executes a simple harmonic

> motion. The time period of oscillation is $\frac{\pi}{\sqrt{x}}$ in SI

unit. The value of x is _____.

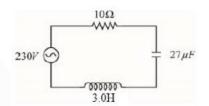
Answer (5)

Sol. Both the springs are in parallel so net spring constant is $K_{net} = K_1 + K_2 = 40 \text{ N/m}$

So
$$T = 2\pi \sqrt{\frac{m}{K_{\text{net}}}}$$

= $2\pi \sqrt{\frac{2}{40}}$
= $\frac{\pi}{\sqrt{5}}$
 $x = 5$

23. In the circuit shown in the figure, the ratio of the quality factor and the band width is ______s.



Answer (10)

Sol. Bandwidth
$$\Delta \omega = \frac{R}{r}$$

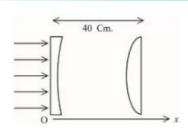
Quality factor
$$Q = \frac{1}{R} \sqrt{\frac{L}{C}}$$

$$So \frac{Q}{\Delta \omega} = \frac{\frac{1}{R}\sqrt{\frac{L}{C}}}{\frac{R}{L}}$$
$$= \frac{\frac{L^2}{R^2\sqrt{C}}}{\frac{3^2}{R^2\sqrt{C}}}$$
$$= \frac{3^{\frac{3}{2}}}{10^2(27 \times 10^{-6})^{\frac{1}{2}}}$$
$$= \frac{3\sqrt{3}}{100(3\sqrt{3} \times 10^{-3})}$$
$$= 10$$

24. As shown in the figure, a combination of a thin plano concave lens and a thin plano convex lens is used to image an object placed at infinity. The radius of curvature of both the lenses is 30 cm and refraction index of the material for both the lenses is 1.75. Both the lenses are placed at distance of 40 cm from each other. Due to the combination, the image of the object is formed at distance $x = _$ cm, from concave lens.







Answer (120)

Sol.
$$\frac{1}{f_{\text{concave}}} = (1.75 - 1) \left(\frac{1}{\infty} - \frac{1}{+30} \right) = -\frac{0.75}{30}$$

 $f_{\rm concave}$ = -40 cm

$$\frac{1}{f_{\text{convex}}} = (1.75 - 1) \left(\frac{1}{30} - \frac{1}{\infty}\right) = \frac{0.75}{30}$$

 $f_{\rm convex} = 40 \ {\rm cm}$

Let the first image is formed at v_1 so

$$\frac{1}{v_1} - \frac{1}{\infty} = \frac{1}{f_{\text{concave}}} = -\frac{1}{40}$$

$$\Rightarrow v_1 = -40 \text{ cm}$$

for second image

$$\frac{1}{x-40} - \frac{1}{-80} = \frac{1}{40}$$

⇒ x = 120 cm

25. A spherical body of mass 2 kg starting from rest acquires a kinetic energy of 10000 J at the end of 5th second. The force acted on the body is ______N.

Answer (40)

Sol. Let the force be *F* so acceleration $a = \frac{F}{m}$

So displacement
$$S = \frac{1}{2}at^2 = \frac{Ft^2}{2m}$$

So work done $W = F.S = \frac{F^2 t^2}{2m}$

From work energy Theorem

$$\Delta KE = W$$
$$W = \frac{F^2 t^2}{2m} = 10000$$
$$F = \sqrt{\frac{10000 \times 2 \times 2}{5^2}}$$

F = 40 N

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26. A stream of a positively charged particles having

$$\frac{q}{m} = 2 \times 10^{11} \frac{C}{kg}$$
 and velocity $\vec{v}_0 = 3 \times 10^7 \hat{i} \, m/s$ is

deflected by an electric field $1.8\hat{j}$ kV/m. The electric field exists in a region of 10 cm along *x* direction. Due to the electric field, the deflection of the charge particles in the *y* direction is _____ mm.

Answer (2)

Sol.
Sol.

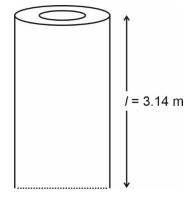
$$\begin{array}{c} \stackrel{\circ \longrightarrow V_0}{\uparrow} \quad \stackrel{\circ}{\longrightarrow} \quad$$

27. A hollow cylindrical conductor has length of 3.14 m, while its inner and outer diameters are 4 mm and 8 mm respectively. The resistance of the conductor is $n \times 10^{-3} \Omega$.

If the resistivity of the material is $2.4 \times 10^{-8} \Omega m$. The value of *n* is _____.

Answer (2)







$$R = \frac{\rho I}{A} = \frac{2.4 \times 10^{-8} \times 3.14}{\pi (16 - 4) \times 10^{-6}}$$
$$= \frac{2.4}{12} \times 10^{-2}$$
$$= \frac{24}{12} \times 10^{-3}$$
$$= 2 \times 10^{-3} \Omega$$

value of *n* is 2.

28. Assume that protons and neutrons have equal masses. Mass of a nucleon is 1.6×10^{-27} kg and radius of nucleus is 1.5×10^{-15} A^{1/3} m. The approximate ratio of the nuclear density and water density is $n \times 10^{13}$. The value of *n* is _____.

Answer (11)

Sol. Radius = $1.5 \times 10^{-15} A^{1/3}$

Volume $=\frac{4\pi}{3}r^3$

Mass of nucleus = $(1.6 \times 10^{-27}) A \text{ kg}$

Density of nucleus = $\frac{1.6 \times 10^{-27} \times A}{\frac{4}{3} \times \pi \times \left(1.5 \times 10^{-15} A^{\frac{1}{3}}\right)^3}$

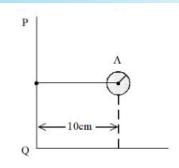
$$=\frac{1.6\times3\times8\times10^{18}}{4\pi\times27}$$

$$=\frac{32}{9\pi}\times10^{17}$$

Density of water = 1000 kg/m³

$$\frac{\text{Density of nucleus}}{\text{Density of water}} = \frac{\frac{32}{9\pi} \times 10^{17}}{1000}$$
$$= \frac{320}{9\pi} \times 10^{13}$$
$$= 11.32 \times 10^{13}$$
$$\text{Value of } n = 11$$

29. Solid sphere A is rotating about an axis PQ. If the radius of the sphere is 5 cm then its radius of gyration about PQ will be \sqrt{x} cm. The value of x is _____.



Sol.

Solid sphere ←10 cm $I = I_{\rm cm} + Md^2$ $\Rightarrow MK^2 = \frac{2}{5}MR^2 + Md^2$ $K = \sqrt{\frac{2}{5}R^2 + d^2}$ $K = \sqrt{\frac{2}{5} \times 5^2 + 10^2} \text{ cm}$ $K = \sqrt{110}$ cm Value of x = 110. $a\hat{i} + b\hat{j} + \hat{k}$ and $2\hat{i} - 3\hat{j} + 4\hat{k}$ 30. Vectors are perpendicular to each other when 3a + 2b = 7, the ratio of *a* to *b* is $\frac{x}{2}$. The value of *x* is _____. Answer (1) **Sol.** $(a\hat{i} + b\hat{j} + \hat{k}) \cdot (2\hat{i} - 3\hat{j} + 4\hat{k}) = 0$ 2a - 3b + 4 = 0 $(2a - 3b = -4 \times) 7$... (1) also, $(3a + 2b = 7 \times) 4$... (2) adding (1) and (2) 14a + 12a - 21b + 8b = 026a - 13b = 0 $\frac{a}{b} = \frac{13}{26} = \frac{1}{2}$ Value of x = 1.



CHEMISTRY

SECTION - A

Multiple Choice Questions: This section contains 20 multiple choice questions. Each question has 4 choices (1), (2), (3) and (4), out of which **ONLY ONE** is correct.

Choose the correct answer :

31. Given below are two statements:

Statement I : Noradrenaline is a neurotransmitter.

Statement II: Low level of noradrenaline is not the cause of depression in human.

In the light of the above statements, choose the correct answer from the options given below.

- (1) Statement I is correct but Statement II is incorrect
- (2) Statement I is incorrect but Statement II is correct
- (3) Both statement I and Statement II are incorrect
- (4) Both statement I and Statement II are correct

Answer (1)

- Sol. Noradrenaline is a neurotransmitter.
 - Low level of noradrenaline is a cause for depression in human.
- 32. Reaction of BeO with ammonia and hydrogen fluoride gives A which on thermal decomposition gives BeF₂ and NH₄F. What is 'A'?

(1) H ₃ NBeF ₃ (2	2) (NH4)BeF3
---	--------------

(3) $(NH_4)Be_2F_5$ (4) $(NH_4)_2BeF_4$

Answer (4)

Sol.

$$\mathsf{BeO} + \mathsf{NH}_3 + \mathsf{HF} \longrightarrow (\mathsf{NH}_4)_2 \underset{(A)}{\mathsf{BeF}_4} \xrightarrow{\Delta} \mathsf{BeF}_2 + \mathsf{NH}_4\mathsf{F}$$

Compound A is (NH₄)₂BeF₄

 Statement I : For colloidal particles, the values of colligative properties are of small order as compared to values shown by true solutions at same concentration.

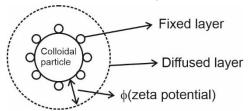
Statement II : For colloidal particles, the potential difference between the fixed layer and the diffused layer of same charges is called the electrokinetic potential or zeta potential.

In the light of the above statements, choose the correct answer from the options given below

- (1) Both statement I and Statement II are true
- (2) Both statement I and Statement II are false
- (3) Statement I is true but Statement II is false
- (4) Statement I is false but Statement II is true

Answer (3)

Sol. For colloidal particles value of colligative properties is less as compared to true solutions at same concentration as number of particles are less.



But fixed layer and diffused layer have opposite charges.

- 34. In the depression of freezing point experiment
 - A. Vapour pressure of the solution is less than that of pure solvent
 - B. Vapour pressure of the solution is more than that of pure solvent
 - C. Only solute molecules solidify at the freezing point
 - D. Only solvent molecules solidify at the freezing point

Choose the most appropriate answer from the options given below:

- (1) A and D only (2) B and C only
- (3) A only (4) A and C only

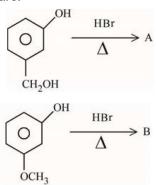
Answer (1)

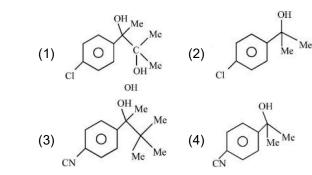
Sol. A and D are correct

as (vp)_{solution} < (vp)_{solvent}

and only solvent particles undergoes solidification.

35. 'A' and 'B' formed in the following set of reactions are:





Answer (1)

Sol.

OH

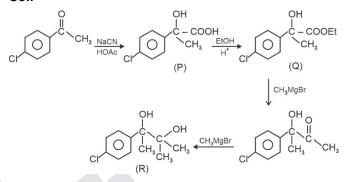
OH

OH

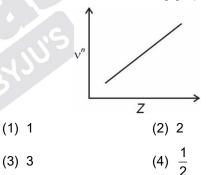
Br

Br

Br



It is observed that characteristic X-ray spectra of 38. elements show regularity. When frequency to the power "n" i.e. v^n of X-rays emitted is plotted against atomic number "Z", following graph is obtained.



Answer (4)

Sol.
$$hv = \Delta E = 13.6 \times Z^2 \left(\frac{1}{n_1^2} - \frac{1}{n_2^2}\right)$$

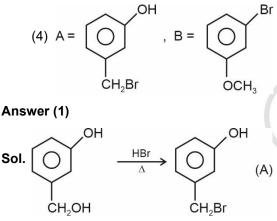
 $\Rightarrow v \propto Z^2$
 $\Rightarrow (v)^{1/2} \propto Z \left(n = \frac{1}{2}\right)$

39. The magnetic moment of a transition metal compound has been calculated to be 3.87 B.M. The metal ion is

(1) Ti ²⁺	(2) Mn ²⁺
----------------------	----------------------

(4) V²⁺ (3) Cr²⁺





OH

OH

CH₂Br

ĊH₂Br

ĊH₂OH

Br

B =

B =

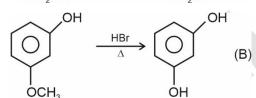
B =

(1) A =

(2)

(3) A =

A =



36. The primary and secondary valencies of cobalt respectively in [Co(NH₃)₅CI]Cl₂ are :

(1) 3 and 5 (2) 2 and 6
----------------	-----------

(3) 2 and 8 (4) 3 and 6

Answer (4)

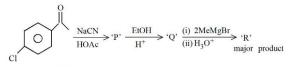
Sol.

Sol. [Co(NH₃)₅Cl]Cl₂

Oxidation no. = primary valencies = 3

Co-ordination no. = secondary valencies = 6

37. 'R' formed in the following sequence of reactions is





.(r₂)

n = 3

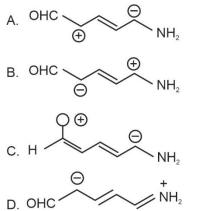
 $V_{23} = 4s^2 3d^3$

$$V^{2+} = 4s^0 3d^3 (n = 3)$$

- 40. Which of the following is true about freons?
 - (1) These are chemicals causing skin cancer
 - (2) These are radicals of chlorine and chlorine monoxide
 - (3) All radicals are called freons
 - (4) These are chlorofluorocarbon compounds

Answer (4)

- Sol. Freons are chlorofluorocarbons
- 41. Increasing order of stability of the resonance structures is:



Choose the correct answer from the options given below:

- (2) D, C, B, A (1) C, D, A, B
- (3) D, C, A, B (4) C, D, B, A

Answer (Not given in the options)

Sol. Correct stabilising order is

C < A < B < D

(This guestion should be given bonus)

42. Assertion A: Hydrolysis of an alkyl chloride is a slow reaction but in the presence of Nal, the rate of the hydrolysis increases.

Reason R: I- is a good nucleophile as well as a good leaving group.

In the light of the above statements, choose the correct answer from the options given below.

- (1) Both A and R are true and R is the correct explanation of A
- (2) A is false but R is true
- (3) A is true but R is false
- (4) Both A and R are true but R is NOT the correct explanation of A

Sol. CH₃ – CI
$$\xrightarrow{H_2O}$$
 CH₃OH (slow process)....(r₁)
CH₃ – CI $\xrightarrow{\text{Nal}}$ CH₃ – I $\xrightarrow{H_2O}$ CH₃ – OH ...(r₂)
(rbeing a Good) CH₃ – OH ...(r₂)

as I⁻ is a good nucleophile as well as good leaving group.

leaving group

43. Match List I with List II

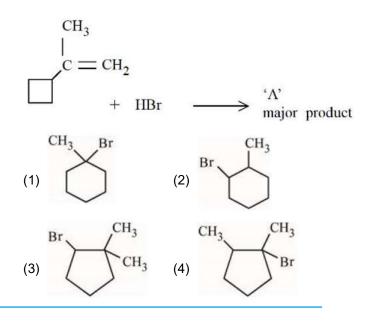
	LIST I		LIST II
Α.	Chlorophyll	I.	Na ₂ CO ₃
В.	Soda ash	11.	CaSO ₄
C.	Dentistry, Ornamental work	III.	Mg ²⁺
D.	Used in white washing	IV.	Ca(OH)₂

Choose the correct answer from the options given below:

- (1) A-II, B-I, C-III, D-IV
- (2) A-III, B-IV, C-I, D-II
- (3) A-II, B-III, C-IV, D-I
- (4) A-III, B-I, C-II, D-IV

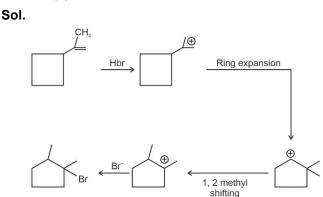
Answer (4)

- Sol. Chlorophyll contains Mg²⁺ lons (A III)
 - Soda ash is Na₂Co₃ (B - I)
 - Dentistry; ornamental work CaSO₄ (C II)
 - Used in white washing Ca(OH)₂ (D - IV)
- 44. In the following given reaction, 'A' is





Answer (4)



45. Match List I with List II

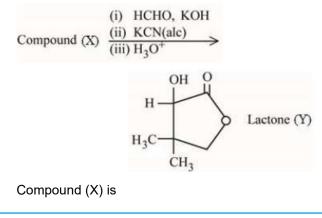
	LIST I	LIST II	
А.	Reverberatory furnace	Ι.	Pig Iron
В.	Electrolytic cell	11.	Aluminum
C.	Blast furnace	III.	Silicon
D.	Zone refining furnace	IV.	Copper

Choose the correct answer from the options given below:

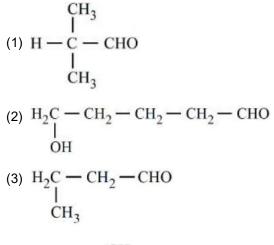
- (1) A-I, B-IV, C-II, D-III (2) A-I, B-III, C-II, D-IV
- (3) A-IV, B-II, C-I, D-III (4) A-III, B-IV, C-I, D-II

Answer (3)

- **Sol.** (A) Reverberatory furnance is used for extraction of copper.
 - (B) Electrolytic cell is used for obtaining highly reactive metals like aluminium.
 - (C) Blast furnace is used for extraction of Iron.
 - (D) Zone refining furnace is used for silicon.
- 46. Compound (X) undergoes following sequence of reactions to give the Lactone (Y).



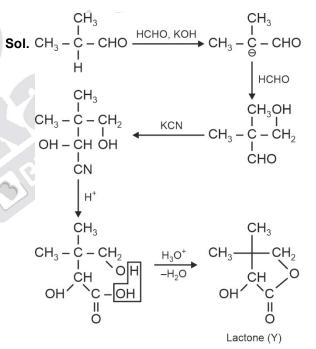
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(4)
$$HOH_2C - C - CHO$$

 $| CH_3$

Answer (1)



- 47. Which of the phosphorus oxoacid can create silver mirror from AgNO₃ solution?
 - (1) H₄P₂O₅
 - (2) (HPO₃)_n
 - (3) H₄P₂O₇
 - (4) H₄P₂O₆

Answer (1)

Sol. $H_4P_2O_5$ can act as a reducing agent due to (P - H) bond.

- 48. Decreasing order of the hydrogen bonding in following forms of water is correctly represented by
 - A. Liquid water
 - B. Ice

 $H_4P_2O_5$

C. Impure water

Choose the correct answer from the options given below:

- (1) A > B > C
- (2) B > A > C
- (3) A = B > C
- (4) C > B > A

Answer (2)

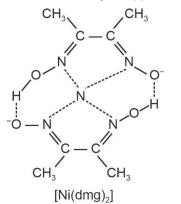
Sol. Extent of hydrogen bonding :

Ice > liquid water > impure water

- In ice, 4 molecules of H₂O are connected to H₂O molecule.
- Impure water will have less hydrogen bonding.
- An ammoniacal metal salt solution gives a brilliant red precipitate on addition of dimethylglyoxime. The metal ion is
 - (1) Ni²⁺
 - (2) Cu²⁺
 - (3) Fe²⁺
 - (4) Co²⁺

Answer (1)

Sol. Ni²⁺ forms cherry red ppt with dmg



50. Order of covalent bond :

- A. KF > KI; LiF > KF
- B. KF < KI; LiF > KF
- C. SnCl4 > SnCl2; CuCl > NaCl
- D. LiF > KF; CuCl < NaCl
- E. KF < KI; CuCl > NaCl

Choose the correct answer from the options given below:

- (1) B, C only (2) C, E only
- (3) B, C, E only (4) A, B only

Answer (3)

Sol. B is correct	KF < KI; LiF > KF
C is correct	$\underset{+4}{SnCl_4} > \underset{+2}{SnCl_2}; CuCl > NaCl$
E is correct	KF < KI; CuCl > NaCl

SECTION - B

Numerical Value Type Questions: This section contains 10 questions. In Section B, attempt any five questions out of 10. The answer to each question is a **NUMERICAL VALUE.** For each question, enter the correct numerical value (in decimal notation, truncated/rounded-off to the second decimal place; e.g. 06.25, 07.00, -00.33, -00.30, 30.27, -27.30) using the mouse and the on-screen virtual numeric keypad in the place designated to enter the answer.

51. The d-electronic configuration of $[CoCl_4]^{2-}$ in tetrahedral crystal field is $e^m t_2^n$. Sum of "m" and "number of unpaired electrons" is _____.

Answer (7)

Sol. [CoCl₄]²⁻

m = 4

 $Co^{2+} = 4s^0 3d^7$

 $\begin{array}{cccc} 1 & 1 & 1_t \\ 1 & 1_{e_g} \\ = & e_g^4 t_{2g^3} \end{array}$

Number of unpaired electrons = 3

52. The dissociation constant of acetic acid is $x \times 10^{-5}$. When 25 mL of 0.2 M CH₃COONa solution is mixed with 25 mL of 0.02 M CH₃COOH solution, the pH of the resultant solution is found to be equal to 5. The value of *x* is _____.

Answer (10)





Sol. pH = pK_a + log
$$\left(\frac{25 \times 0.2}{25 \times 0.02}\right)$$

5 = pK_a + log10
pK_a = 4 \Rightarrow K_a = 10⁻⁴ = 10 × 10⁻⁵
x = 10

53. Uracil is a base present in RNA with the following structure. % of N in uracil is _____.



Given :

Molar mass $N = 14 \text{ g mol}^{-1}$ $O = 16 \text{ g mol}^{-1}$ $C = 12 \text{ g mol}^{-1}$ $H = 1 \text{ g mol}^{-1}$

Answer (25)

Sol. Uracil is C₄H₄N₂O₂

% by mass of N =
$$\frac{14 \times 2}{112} \times 100$$

= 25%

- 54. The number of correct statement/s from the following is _____.
 - A. Larger the activation energy, smaller is the value of the rate constant.
 - B. The higher is the activation energy, higher is the value of the temperature coefficient.
 - C. At lower temperatures, increase in temperature causes more change in the value of k than at higher temperature.
 - D. A plot of ln k vs $\frac{1}{T}$ is a straight line with slope equal to $-\frac{E_a}{R}$.

Answer (3)

Sol. (A)
$$k = Ae^{\frac{E_a}{RT}} (E_a \uparrow k \downarrow)$$

(B) $\ln k = \ln A - \frac{E_a}{RT}$
 $\frac{1}{k} \frac{dk}{dT} = \frac{+E_a}{RT^2}$

 $E_a \uparrow$ temp. coefficient \uparrow

(C) K Large change
Small change
(D)
$$\ln k = \ln A - \frac{E_a}{RT}$$

Slope of $\ln k$ vs $\frac{1}{T}$ is $\left(\frac{-E_a}{R}\right)$
At 298 K, a 1 litre solution containing

55. At 298 K, a 1 litre solution containing 10 mmol of $Cr_2O_7^{2-}$ and 100 mmol of Cr^{3+} shows a pH of 3.0.

Given : $Cr_2O_7^{2-} \rightarrow Cr^{3+}$; E° = 1.330V

and
$$\frac{2.303 \text{ RT}}{\text{F}} = 0.059 \text{ V}$$

The potential for the half cell reaction is $x \times 10^{-3}$ V. The value of *x* is _____.

Answer (917)

Sol.
$$6e^{-} + 14H^{+} + Cr_2O_7^{2-} \rightarrow 2Cr^{3+} + 7H_2O$$

 $E_{cell} = 1.330 - \frac{0.059}{6} \log \frac{10^{-2}}{(10^{-2})(10^{-42})}$
 $= 1.330 - \frac{0.059}{6}(42)$
 $= 1.330 - 0.413$
 $= 0.917 = 917 \times 10^{-3}$
 $x = 917$

56. 5 g of NaOH was dissolved in deionized water to prepare a 450 mL stock solution. What volume (in mL) of this solution would be required to prepare 500 mL of 0.1 M solution?_____

Given : Molar Mass of Na, O and H is 23, 16 and 1 g mol⁻¹ respectively

Answer (180)

Sol. Molarity of solution = $\frac{5}{(40)} \frac{(1000)}{(450)}$ $\Rightarrow M \times V = 500 \times .1$ $\Rightarrow \frac{5}{450} \times \frac{1000}{450} \times V = 500 \times .1$ $\boxed{V = 180 \text{ mL}}$

57. For independent processes at 300 K

Process	∆H/kJ mol ⁻¹	∆S/J K ⁻¹
А	-25	-80
В	-22	40
С	25	-50
D	22	20

The number of non-spontaneous processes from the following is_____

Answer (2)

Sol. – C is non – spon tan eous as $(\Delta H > 0, \Delta S < 0)$

D is Non - spontaneous

For D,

- $\Delta G = \Delta H T \Delta S$
 - = 22,000 (300) (20)
 - = (22,000 6000) > 0

Non-spontaneous as $(\Delta G > 0)$

- 58. When Fe_{0.93}O is heated in presence of oxygen, it converts to Fe₂O₃. The number of correct statement/s from the following is_____
 - A. The equivalent weight of Fe_{0.93}O is <u>Molecular weight</u> 0.79
 - B. The number of moles of Fe²⁺ and Fe³⁺ in 1 mole of Fe_{0.93}O is 0.79 and 0.14 respectively.
 - C. Fe_{0.93}O is metal deficient with lattice comprising of cubic closed packed arrangement of O²⁻ ions.
 - D. The % composition of Fe²⁺ and Fe³⁺ in Fe_{0.93}O is 85% and 15% respectively.

Answer (4)

Sol. A: $Fe_{0.93}O \longrightarrow Fe_2O_3$

$$\mathsf{nF} = \left(3 - \frac{200}{93}\right) \times .93$$

= .79

eq. wt = $\frac{mw}{.79}$

$$Fe_x^{+2} Fe_{(93-x)}^{+3}O_{100}$$

$$2x + 3(93 - x) = 200$$

x = 79
% of Fe²⁺ = $\frac{79}{93} \times 100 = 85\%$

% of Fe³⁺ =
$$\frac{14}{93} \times 100 = 15\%$$

= Fe_{0.93}O is metal deficient compound

A, B, C, D are correct.

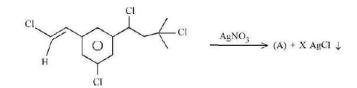
59. If wavelength of the first line of the Paschen series of hydrogen atom is 720 nm, then the wavelength of the second line of this series is _____ nm. (Nearest integer)

Answer (492)

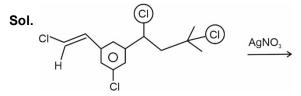
Sol.
$$\frac{1}{720} = \mathbb{R} \times \left(\frac{1}{9} - \frac{1}{16}\right)$$
$$\Rightarrow \mathbb{R} = \frac{9 \times 16}{720 \times 7}$$
$$\frac{1}{\lambda'} = \frac{9 \times 16}{720 \times 7} \times \left(\frac{1}{9} - \frac{1}{25}\right)$$
$$\lambda' = 492.18 \text{ nm}$$

 $\lambda' = 492 \text{ nm}$ (nearest integer)

60. Number of moles of AgCl formed in the following reaction is_____



Answer (2)



Circled Cl will get precipitated

MATHEMATICS

SECTION - A

Multiple Choice Questions: This section contains 20 multiple choice questions. Each question has 4 choices (1), (2), (3) and (4), out of which ONLY ONE is correct.

Choose the correct answer :

- 61. Let N denote the number that turns up when a fair die is rolled. If the probability that the system of equations
 - x + y + z = 1
 - 2x + Ny + 2z = 2
 - 3x + 3y + Nz = 3

has unique solution is $\frac{k}{6}$, then the sum of value of

- k and all possible values of N is
- (1) 18 (2) 20 (4) 19
- (3) 21

Answer (2)

Sol. For unique solution $\Delta \neq 0$

- 1 1 1 i.e. $|2 \ N \ 2| \neq 0$ 3 3 N
- \Rightarrow $(N^2 6) (2N 6) + (6 3N) \neq 0$
- $\Rightarrow N^2 5N + 6 \neq 0$
- \therefore $N \neq 2$ and $N \neq 3$
- Probability of not getting 2 or 3 in a throw of dice $=\frac{2}{3}$ •

As given $\frac{2}{3} = \frac{k}{6} \implies k = 4$

 \therefore Required value = 1 + 4 + 5 + 6 + 4 = 20

62. Let
$$\vec{u} = \hat{i} - \hat{j} - 2\hat{k}$$
, $\vec{v} = 2\hat{i} + \hat{j} - \hat{k}$, $\vec{v} \cdot \vec{w} = 2$ and
 $\vec{v} \times \vec{w} = \vec{u} + \lambda \vec{v}$. Then $\vec{u} \cdot \vec{w}$ is equal to
(1) 2 (2) 1

(4) $-\frac{2}{3}$ (3)

Answer (2)

Sol. Given $\vec{v} \times \vec{w} = \lambda \vec{v} + \vec{u}$...(i) Taking dot with \vec{v} we get

$$\left[\vec{v}\,\vec{v}\,\vec{w}\right] = \lambda \left|\vec{v}\right|^2 + \vec{u}\cdot\vec{v}$$

Substituting values we have

$$6\lambda + 3 = 0 \implies \lambda = -\frac{1}{2}$$

Equation (i) becomes •

$$\vec{v} \times \vec{w} = \vec{u} - \frac{\vec{v}}{2}$$
 ...(ii)

Taking dot with \vec{w} of (ii) we get

$$0 = \vec{u} \cdot \vec{w} - \frac{\vec{v} \cdot \vec{w}}{2}$$

$$\Rightarrow \quad \vec{u} \cdot \vec{w} = \frac{2}{2} = 1 \qquad (as \ \vec{v} \cdot \vec{w} = 2 \ given)$$

63. The distance of the point (-1, 9, -16) from the plane 2x + 3y - z = 5 measured parallel to the line

(4) 31

$$\frac{x+4}{3} = \frac{2-y}{4} = \frac{z-3}{12}$$
 is
(1) $20\sqrt{2}$ (2) $13\sqrt{2}$

Answer (3)

Sol.
$$\frac{x+4}{3} = \frac{y-2}{-4} = \frac{z-3}{12}$$

$$A = \begin{bmatrix} P(-1, 9, -16) \\ 23, -4, \sqrt{2^7} \end{bmatrix}$$

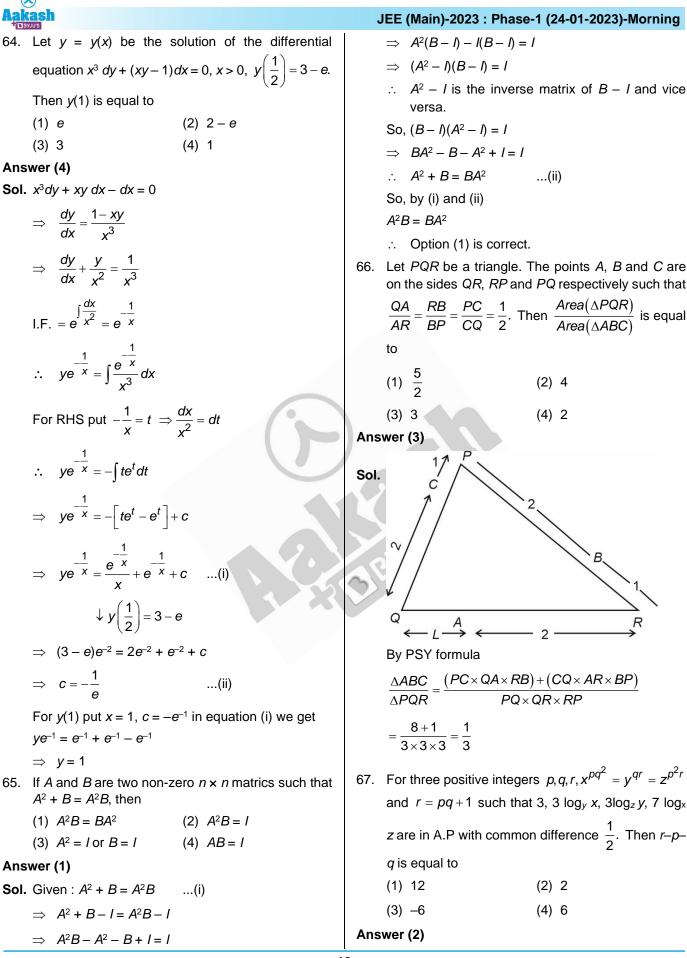
$$2x + 3y - z = 5$$

Equation of line AP

$$\frac{x+1}{3} = \frac{y-9}{-4} = \frac{z+16}{12}$$

z = 5, $\Rightarrow 6\lambda - 2 - 12\lambda + 27 - 12\lambda + 16 = 5 \Rightarrow \lambda = 2$ \Rightarrow Point A(5, 1, 8) $\Rightarrow AP^2 = 6^2 + 8^2 + 24^2 = 4(9 + 16 + 144) = 4 \times 169$ AP = 26

Option (3) is correct.



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Sol.
$$x^{pq^2} = y^{qr} = x^{p^2r}$$

 $\exists \log_y x = \frac{7}{2}, \exists \log_x y = 4, 7\log_y z = \frac{9}{2}$
 $\Rightarrow x = y^{\frac{7}{6}}, y = z^{\frac{3}{2}}, z = x^{\frac{1}{14}}$
 $y^{\frac{7}{6}} \tilde{p}^{q^2} = y^{qr} = y^{\frac{3}{4}} p^{2r}$
 $\Rightarrow \frac{7}{r} pq^{2} = qr = \frac{3}{4} p^{2r}$
 $\therefore 7pq = 6r, 4q = 3p^{2}$
 $r = pq + 1$
 $r = \frac{6r}{r} + 1 \Rightarrow r = 7$
 $pq = 6$
 $p(\frac{3p^{2}}{4}) = 6$
 $p = 2, q = 3$
 $r = pq - r - 5 = 2$
88. The relation
 $R = [(a,b): god(a,b) = 1, 2a \neq b, a, b \in \mathbb{Z}]$ is
(1) Reflexive but not symmetric
(2) Neither symmetric hor transitive
(3) Symmetric but not transitive
(4) Transitive but not transitive
(3) Symmetric but not transitive
(4) Transitive but not transitive
(5) Symmetric
If $god(a, b) = 1 \Rightarrow god(b, c) = 1$
 $\Rightarrow god(a, a) = a so (a, a) so $c R \Rightarrow$ not reflexive
If $god(a, b) = 1 \Rightarrow god(b, c) = 1$
 $\Rightarrow god(a, b) = 1 \Rightarrow god(b, c) = 1$
 $\Rightarrow god(a, b) = 1 \Rightarrow god(b, c) = 1$
 $\Rightarrow god(a, b) = 1 \Rightarrow god(b, c) = 1$
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 $\Rightarrow god(a, b) = 1 \Rightarrow god(b, c) = 1$
 $\Rightarrow god(a, b) = 1 \Rightarrow god(b, c) = 1$
 $\Rightarrow god(a, c) = 1 \Rightarrow (2) \frac{22}{3}$
 $(1) \frac{23}{3}$ $(2) \frac{27}{3}$
 $(2) \frac{25}{3}$ $(4) 9$
Answer (4)
Answer (4)
Answer (4)
Answer (4)$



Sol. $(1+x)^{22} = {}^{22}C_0 + {}^{22}C_1x + {}^{22}C_2x^2 + \dots + {}^{22}C_{21}x^{21} + {}^{22}C_{22}x^{22} - \dots (i)$ $(x+1)^{23} = {}^{23}C_0x^{23} + {}^{23}C_1x^{22} + {}^{23}C_2x^{21} + \dots + {}^{23}C_{21}x^2 + {}^{23}C_{22}x + {}^{23}C_{23} - \dots (ii)$

Multiplying (i) & (ii) and comparing coefficients of x^{23} on both sides

$${}^{45}C_{23} = {}^{22}C_0 \cdot {}^{23}C_0 + {}^{22}C_1 \cdot {}^{23}C_1 + {}^{22}C_2 \cdot {}^{23}C_2 + \dots + {}^{22}C_{22} \cdot {}^{23}C_{22}$$

$$\sum_{r=1}^{22} {}^{22}C_r \cdot {}^{23}C_r = {}^{45}C_{23}$$

- 72. Let a tangent to the curve $y^2 = 24x$ meet the curve xy = 2 at the points *A* and *B*. Then the mid points of such line segments *AB* lie on a parabola with the
 - (1) Directrix 4x = 3
 - (2) Length of latus rectum $\frac{3}{2}$
 - (3) Length of latus rectum 2
 - (4) Directrix 4x = -3

Answer (1)

r=0

Sol. $y^2 = 24x$, xy = 2

Let the equation of tangent to $y^2 = 24x$ is $ty = x + 6t^2$ $ty = x + 6t^2$ meet the curve xy = 2 at points A and B. Let mid-point of AB is P(h, k).

$$ty = \frac{2}{y} + 6t^{2}$$

$$ty^{2} - 6t^{2}y - 2 = 0$$

$$y_{1} + y_{2} = 6t$$

$$x^{2} + 6t^{2}x - 2t = 0$$

$$x_{1} + x_{2} = -6t^{2}$$

$$\Rightarrow \text{ Mid-point } P \text{ is } (-3t^{2}, 3t)$$

$$\Rightarrow h = -3t^{2}, k = 3t$$

$$\Rightarrow \left(\frac{h}{-3}\right) = \left(\frac{k}{3}\right)^{2}$$

$$\Rightarrow y^{2} = -3x$$

 \Rightarrow Length of L.R. = 3

Equation of directrix is $x = \frac{3}{4}$

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- 73. The equation $x^2 4x + [x] + 3 = x[x]$, where [x] denotes the greatest integer function, has
 - (1) Exactly two solutions in $(-\infty, \infty)$
 - (2) No solution
 - (3) A unique solution in $(-\infty, 1)$
 - (4) A unique solution in $(-\infty, \infty)$

Answer (4)

Sol.
$$x^2 - 4x + [x] + 3 = x[x]$$

 $(x - 1)(x - 3) = (x - 1)[x]$
 $(x - 1)(x - 3 - [x]) = 0$
 $x = 1 \text{ or } x - 3 - [x] = 0$
 $\{x\} = 3$
 $x = \phi$

- \Rightarrow a unique solution in (- ∞ , ∞)
- 74. The distance of a point (7, -3, -4) from the plane passing through the points (2, -3, 1), (-1, 1, -2) and (3, -4, 2) is
 - (1) 4√2
 - (2) 4
 - (3) 5√2
 - (4) 5

Answer (3)

Sol. A(2, -3, 1), B(-1, 1, -2), C(3, -4, 2)

$$\overrightarrow{AB} = -3\hat{i} + 4\hat{j} - 3\hat{k}$$
 $\overrightarrow{AC} = \hat{i} - \hat{j} + \hat{k}$

$$\vec{n} = \begin{vmatrix} \hat{i} & \hat{j} & \hat{k} \\ -3 & 4 & -3 \\ 1 & -1 & 1 \end{vmatrix} = \hat{i} - \hat{k}$$

Let equation of plane is $x - z + \lambda = 0$ passes through point $A(2, -3, 1) \Rightarrow \lambda = -1$

Equation of plane is x - z - 1 = 0

Distance of point (7, -3, -4) from the plane x - z - 1 = 0 is $5\sqrt{2}$

- 75. The compound statement $(\sim (P \land Q)) \lor ((\sim P) \land Q) \Rightarrow ((\sim P) \land (\sim Q))$ is equivalent to (1) $((\sim P) \lor Q) \land (\sim Q)$
 - (2) (~ Q) ∧ P
 - (3) (~ *P*)∨ *Q*

(4)
$$((\sim P) \lor Q) \land ((Q) \lor P)$$

Sol.
$$(\sim (P \land Q)) \lor ((\sim P) \land Q) \Rightarrow ((\sim P) \land (\sim Q))$$

 $(\sim P \lor \sim Q) \lor (\sim P \land Q) \Rightarrow (\sim P \land \sim Q)$
 $\Rightarrow (\sim P \lor \sim Q \lor \sim P) \land (\sim P \lor \sim Q \lor Q) \Rightarrow (\sim P \land \sim Q)$
 $\Rightarrow (\sim P \lor \sim Q) \land (T) \Rightarrow (\sim P \land \sim Q)$
 $\Rightarrow (\sim P \lor \sim Q) \land (T) \Rightarrow (\sim P \land \sim Q)$
 $\Rightarrow (\sim P \lor \sim Q) \Rightarrow (\sim P \land \sim Q)$
 $\Rightarrow (\sim P \lor \sim Q) \lor (\sim P \land \sim Q)$
 $\Rightarrow (\sim P \lor \sim Q) \lor (\sim P \land \sim Q)$

- 76. Let Ω be the sample space and $A \subseteq \Omega$ be an event. Given below are two statements:
 - (S1) : If P(A) = 0, then $A = \emptyset$
 - (S2) : If P(A) = 1, then $A = \Omega$

Then

- (1) Both (S1) and (S2) are false
- (2) Only (S1) is true
- (3) Only (S2) is true
- (4) Both (S1) and (S2) are true

Answer (4)

- Sol. Both statements are correct
- 77. Let α be a root of the equation $(a c)x^2 + (b a)x$ +(c-b) = 0 where a, b, c are distinct real numbers

 $\begin{bmatrix} \alpha^2 & \alpha & 1 \end{bmatrix}$ such that the matrix 1 1 is singular. Then, 1 b c а

the value of

$$\frac{(a-c)^2}{(b-a)(c-b)} + \frac{(b-a)^2}{(a-c)(c-b)} + \frac{(c-b)^2}{(a-c)(b-a)}$$
 is
(1) 12 (2) 6
(3) 9 (4) 3

Answer (4)

Sol.
$$\frac{(a-c)^{3} + (b-a)^{3} + (c-a)^{3}}{(a-c)(c-b)(b-a)}$$
$$= \frac{3(a-c)(b-a)(c-b)}{(a-c)(b-a)(c-b)}$$
$$= 3$$
78. Let $f(x) = \begin{cases} x^{2} \sin\left(\frac{1}{x}\right), & x \neq 0\\ 0, & x = 0 \end{cases}$ Then at $x = 0$

- (1) f is continuous but not differentiable
- (2) f and f both are continuous
- (3) f is continuous but not differentiable
- (4) *f* is continuous but *f* is not continuous

Answer (4)

Sol.
$$f(x) = \begin{cases} x^2 \sin \frac{1}{x}, & x \neq 0 \\ 0, & x = 0 \end{cases}$$

$$\lim_{x \to 0^+} f(x) = \lim_{x \to 0^-} f(x) = f(0) = 0$$

 \therefore f is continuous at x = 0

Now,
$$R.H.D = \lim_{h \to 0} \frac{f(h) - f(0)}{h} = \frac{h^2 \sin \frac{1}{h}}{h} = 0$$

at
$$x = 0$$

and LHD

$$= \lim_{h \to 0} \frac{f(-h) - f(0)}{-h} = \frac{-h^2 \sin\left(\frac{1}{h}\right)}{-h} = 0$$

$$\therefore$$
 RHD = LHD \therefore *f* is differentiable at *x* = 0

$$\therefore f'(x) = \begin{cases} 2x\sin\left(\frac{1}{x}\right) - \cos\left(\frac{1}{x}\right), & x \neq 0\\ 0, & x = 0 \end{cases}$$

 $\lim f(x)$ is oscillatory $x \rightarrow 0^{-}$

f is continuous but *f* is not at
$$x = 0$$

79.
$$\tan^{-1}\left(\frac{1+\sqrt{3}}{3+\sqrt{3}}\right) + \sec^{-1}\left(\sqrt{\frac{8+4\sqrt{3}}{6+3\sqrt{3}}}\right)$$
 is equal to:
(1) $\frac{\pi}{3}$ (2) $\frac{\pi}{6}$
(3) $\frac{\pi}{2}$ (4) $\frac{\pi}{4}$

4

Answer (1)

Sol.
$$\tan^{-1}\left(\frac{1+\sqrt{3}}{3+\sqrt{3}}\right) = \tan^{-1}\left(\frac{1}{\sqrt{3}}\right) = \frac{\pi}{6}$$

and $\sec^{-1}\left(\sqrt{\frac{4(2+\sqrt{3})}{3(2+\sqrt{3})}}\right) = \sec^{-1}\left(\frac{2}{\sqrt{3}}\right) = \frac{\pi}{6}$
 $\therefore \quad \tan^{-1}\left(\frac{1+\sqrt{3}}{3+\sqrt{3}}\right) + \sec^{-1}\left(\sqrt{\frac{8+4\sqrt{3}}{6+3\sqrt{3}}}\right) = \frac{\pi}{6} + \frac{\pi}{6} = \frac{\pi}{3}$



80. Let $p, q \in \mathbb{R}$ and $(1-\sqrt{3}i)^{200} = 2^{199} (p+iq), i = \sqrt{-1}$, then $p + q + q^2$ and $p - q + q^2$ are roots of the equation. (1) $x^2 - 4x + 1 = 0$ (2) $x^2 - 4x - 1 = 0$ (3) $x^2 + 4x - 1 = 0$ (4) $x^2 + 4x + 1 = 0$ Answer (1) Sol. Given $(1-\sqrt{3}i)^{200} = 2^{199} (p+iq) \dots (1)$

L.H.S =
$$2^{200} \left[\cos\left(\frac{5\pi}{3}\right) + i \sin\frac{5\pi}{3} \right]^{200}$$

= $2^{200} \left[\cos\frac{1000\pi}{3} + i \sin\frac{1000\pi}{3} \right]$
= $2^{200} \left[-\frac{1}{2} - \frac{\sqrt{3}}{2} i \right]$

So, by (1)

 $P = -1, \quad q = -\sqrt{3}$

$$\therefore \quad p+q+q^2 = -1 - \sqrt{3} + 3 = 2 - \sqrt{3} = \alpha$$

:. quadratic equation whose roots are α and β $x^2 - 4x + 1 = 0$

Option (1) is correct.

SECTION - B

Numerical Value Type Questions: This section contains 10 questions. In Section B, attempt any five questions out of 10. The answer to each question is a **NUMERICAL VALUE.** For each question, enter the correct numerical value (in decimal notation, truncated/rounded-off to the second decimal place; e.g. 06.25, 07.00, -00.33, -00.30, 30.27, -27.30) using the mouse and the on-screen virtual numeric keypad in the place designated to enter the answer.

81. Suppose
$$\sum_{r=0}^{2023} r^2 \, {}^{2023}C_r = 2023 \times \alpha \times 2^{2022}$$
. Then

the value of α is _____

Answer (1012)

Sol. ::
$$(1+x)^{2023} = \sum_{r=0}^{2023} {}^{2023}C_r x^r$$

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$$\Rightarrow (2023)(1+x)^{2022} = \sum_{r=0}^{2023} 2023 C_r r x^{r-1}$$

$$\Rightarrow (2023) x (1+x)^{2022} = \sum_{r=0}^{2023} r^{2023} C_r x^r$$

$$\Rightarrow (2023) \Big[x2022(1+x)^{2021} + (1+x)^{2022} \Big]$$

$$= \sum_{r=0}^{2023} r^{22023} C_r x^{r-1}$$

Put x = 1

$$\Rightarrow 2023 \Big[2022 \cdot 2^{2021} + 2^{2022} \Big] = \sum_{r=0}^{2023} r^{22023} C_r$$

$$\therefore \sum_{r=0}^{2023} r^{22023} C_r = 2023 \cdot 2^{2022} (1012)$$

$$\therefore \alpha = 1012$$

82. Let *C* be the largest circle centred at (2, 0) and inscribed in the ellipse $\frac{x^2}{36} + \frac{y^2}{16} = 1$.

If $(1, \alpha)$ lies on *C*, then $10\alpha^2$ is equal to ____ Answer (118.00)

Sol.
$$\frac{x^2}{36} + \frac{y^2}{16} = 1$$

 $r^2 = (x-2)^2 + y^2$
Solving simultaneously
 $-5x^2 + 36x + (9r^2 - 180) = 0$
 $D = 0$
 $r^2 = \frac{128}{10}$

Distance between (1, α) and (2, 0) should be r

$$1 + \alpha^2 = \frac{128}{10}$$
$$\alpha^2 = \frac{118}{10}$$
$$- 118.00$$

83. The number of 9 digit numbers, that can be formed using all the digits of the number 123412341 so that the even digits occupy only even places, is

Answer (60)

Sol. Given number 123412341

$$\overline{\land \land \land \land}$$

Total number
$$=$$
 $\frac{4!}{2!2!} \times \frac{5!}{3!2!} = 6 \times 10 = 60$

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84. Let a tangent to the curve $9x^2 + 16y^2 = 144$ intersect the coordinate axes at the points *A* and *B*. Then, the minimum length of the line segment *AB* is

Answer (07)

Sol. Given curve : $9x^2 + 16y^2 = 144$

$$\Rightarrow \frac{x^2}{16} + \frac{y^2}{9} = 1$$

Let $P(4\cos\theta, 3\sin\theta)$ be any point on it.

Now tangent at P

$$\frac{x\cos\theta}{4} + \frac{y\sin\theta}{3} = 1$$

$$\therefore A \equiv (4\sec\theta, 0) B \equiv (0, 3\csc\theta)$$

$$AB = \sqrt{16\sec^2\theta + 9\csc^2\theta}$$

$$= \sqrt{16 + 9 + 16\tan^2\theta + 9\cot^2\theta}$$

$$AB_{\min} = \sqrt{25 + 2 \times 12}$$

$$= 7$$

85. A boy needs to select five courses from 12 available courses, out of which 5 courses are language courses. If he can choose at most two language courses, then the number of ways he can choose five courses is _____

Answer (546)

Sol. Case 1 If no language course is selected.

$$= {}^{7}C_{5}$$

Case 2 If one language course is selected.

$${}^{7}C_{4} \cdot {}^{5}C_{1}$$

Case 3 If two language course is selected.

$$^{7}C_{3} \cdot {}^{5}C_{2} \cdot$$

Total =
$${}^{7}C_{5} + {}^{7}C_{4} \cdot {}^{5}C_{1} + {}^{7}C_{3} \cdot {}^{5}C_{2}$$

= 21 + 175 + 350
= 546

86. The value of 12
$$\int_{0}^{3} |x^2 - 3x + 2| dx$$
 is _____.

Answer (22)

Sol.
$$12\int_{0}^{3} |x^{2} - 3x + 2| dx$$

Let $I = \int_{0}^{3} |(x - 2)(x - 1) dx$

 $=\int_{-1}^{1} (x-1)(x-2)dx - \int_{-1}^{2} (x-1)(x-2)dx + \int_{-1}^{3} (x-1)(x-2)dx$ $= \left[\frac{x^{3}}{3} - \frac{3x^{2}}{2} + 2x\right]^{1} - \left[\frac{x^{3}}{3} - \frac{3x^{2}}{2} + 2x\right]^{2}$ $+\left[\frac{x^3}{3}-\frac{3x^2}{2}+2x\right]^3$ $= \left\lceil \frac{1}{3} - \frac{3}{2} + 2 \right\rceil - \left\{ \left(\frac{8}{3} - 6 + 4 \right) - \left(\frac{1}{3} - \frac{3}{2} + 2 \right) \right\}$ $+\left\{\left(9-\frac{27}{2}+6\right)-\left(\frac{8}{3}-6+4\right)\right\}$ $=\frac{11}{6}$ $12I = \frac{11}{6} \times 12$ = 22 87. The 4th term of GP is 500 and its common ratio is $\frac{1}{m}$, $m \in \mathbb{N}$. Let S_n denote the sum of the first *n* terms of this GP. If $S_6 > S_5 + 1$ and $S_7 < S_6 + \frac{1}{2}$, then the number of possible values of m is Answer (12) **Sol.** $T_{4} = 500$ $ar^3 = 500 \Rightarrow a = \frac{500}{r^3}$ Now. $S_6 > S_5 + 1$ $\frac{a(1-r^6)}{1-r} - \frac{a(1-r^5)}{1-r} > 1$ *ar*⁵ > 1 Now, $r = \frac{1}{m}$ and $a = \frac{500}{r^3}$ $\Rightarrow m^2 < 500$ \therefore $m > 0 \Rightarrow m \in (0, 10\sqrt{5})$...(i) $S_7 < S_6 + \frac{1}{2}$

$$\frac{a(1-r^{6})}{1-r} < \frac{a(1-r^{6})}{1-r} + \frac{1}{2}$$
$$ar^{6} < \frac{1}{2}$$

lakash JEE (Main)-2023 : Phase-1 (24-01-2023)-Morning $\lambda \in \mathbb{R}$ and let the equation \therefore $r = \frac{1}{m}$ and $a = \frac{500}{r^5}$ 89. Let E be $|x|^2 - 2|x| + |\lambda - 3| = 0$. Then the largest element in the set $S = \{x + \lambda : x \text{ is an integer solution of } E\}$ is $\frac{1}{m^3} < \frac{1}{1000}$ Answer (05) $\Rightarrow m \in (10, \infty)$...(ii) **Sol.** $D \ge 0 \Rightarrow 4-4|\lambda-3| \ge 0$ Possible values of *m* is {11, 12,22} $|\lambda - 3| \le 1$ $\therefore m \in N$ $-1 \leq \lambda - 3 \leq 1$ Total 12 values $2 \le \lambda \le 4$ 88. The shortest distance between the lines $|\mathbf{x}| = \frac{2 \pm \sqrt{4 - 4\left|\lambda - 3\right|}}{2}$ $\frac{x-2}{3} = \frac{y+1}{2} = \frac{z-6}{2}$ and $\frac{x-6}{3} = \frac{1-y}{2} = \frac{z+8}{0}$ is equal to $=1\pm\sqrt{1-|\lambda-3|}$ Answer (14) $x_{\text{largest}} = 1 + 1 = 2$, when $\lambda = 3$ **Sol.** $\vec{a}_1 = 2\hat{i} - \hat{j} + 6\hat{k}$ Largest element of S = 2 + 3 = 5 $\vec{a}_2 = 6\hat{i} + \hat{j} - 8\hat{k}$ 90. The value of $\frac{8}{\pi} \int_{0}^{\frac{1}{2}} \frac{(\cos x)^{2023}}{(\sin x)^{2023} + (\cos x)^{2023}} dx$ is $\vec{a} = 3\hat{i} + 2\hat{i} + 2\hat{k}$ $\vec{b} = 3\hat{i} - 2\hat{i} + 0\hat{k}$ $S.D = \frac{\left| \begin{bmatrix} \vec{a}_2 - \vec{a}_1 & \vec{a} & \vec{b} \end{bmatrix} \right|}{\left| \vec{a} \times \vec{b} \right|}$ Answer (02) **Sol.** $I = \frac{8}{\pi} \int_{0}^{\frac{1}{2}} \frac{(\cos x)^{2023}}{(\sin x)^{2023} + (\cos x)^{2023}} dx$...(i) $\begin{bmatrix} \vec{a}_2 - \vec{a}_1 & \vec{a} & \vec{b} \end{bmatrix} = \begin{vmatrix} 4 & 2 & -14 \\ 3 & 2 & 2 \\ 3 & 2 & 0 \end{vmatrix}$ $I = \frac{8}{\pi} \int_{-\infty}^{\frac{\pi}{2}} \frac{(\sin x)^{2023}}{(\sin x)^{2023} + (\cos x)^{2023}} dx$...(ii) $= 4 \times (4) - 2 (-6) - 14 (-12)$ = 16 + 12 + 168 = 196 $\therefore \int_{a}^{b} f(x) dx = \int_{a}^{b} f(a+b-x) dx$ $\vec{a} \times \vec{b} = \begin{vmatrix} \hat{i} & \hat{j} & \hat{k} \\ 3 & 2 & 2 \\ 2 & 2 & 0 \end{vmatrix} = 4\hat{i} + 6\hat{j} - 12\hat{k}$ (i) + (ii) $\Rightarrow 2I = \frac{8}{\pi} \int_{0}^{\frac{\pi}{2}} 1 dx = \frac{8}{\pi} \times \frac{\pi}{2} = 4$ $\left|\vec{a} \times \vec{b}\right| = \sqrt{16 + 36 + 144} = \sqrt{196} = 14$ $S.D = \frac{196}{14} = 14$ I = 2