Date: 04/06/2023



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Time: 3 hrs. Answers & Solutions M

Max. Marks: 180

for

JEE (Advanced)-2023 (Paper-2)

PART-I: PHYSICS

SECTION 1 (Maximum Marks: 12)

- This section contains FOUR (04) questions.
- Each question has **FOUR** options (A), (B), (C) and (D). **ONLY ONE** of these four options is the correct answer.
- For each question, choose the option corresponding to the correct answer.
- Answer to each question will be evaluated according to the following marking scheme:

Full Marks : +3 If **ONLY** the correct option is chosen;

Zero Marks : 0 If none of the options is chosen (i.e. the question is unanswered);

Negative Marks : -1 In all other cases.

1. An electric dipole is formed by two charges +q and -q located in xy-plane at (0, 2) mm and (0, -2) mm, respectively, as shown in the figure. The electric potential at point P(100, 100) mm due to the dipole is V_0 . The charges +q and -q are then moved to the points (-1, 2) mm and (1, -2) mm, respectively. What is the value of electric potential at P due to the new dipole?

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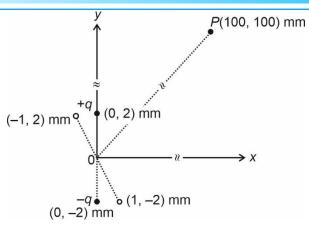






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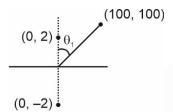




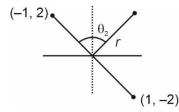
- (B) $\frac{V_0}{2}$
- (D) $\frac{3V_0}{4}$

Answer (B)

Sol.
$$V_1 \propto \frac{p_1 \cos \theta_1}{r_1^3}$$



$$V_2 \propto \frac{p_2 \cos \theta_2}{r_2^3}$$



$$\frac{V_2}{V_1} = \frac{p_2 \cos \theta_2}{p_1 \cos \theta_1}$$

$$\frac{V_2}{V_1} = \frac{q(-2\hat{i} + 4\hat{j}) \cdot (\hat{i} + \hat{j})}{q(0\hat{i} + 4\hat{j}) \cdot (\hat{i} + \hat{j})} = \frac{1}{2}$$

$$\Rightarrow V_2 = \frac{V_0}{2}$$

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- **2.** Young's modulus of elasticity Y is expressed in terms of three derived quantities, namely, the gravitational constant G, Planck's constant h and the speed of light c, as $Y = c^{\alpha}h^{\beta}G^{\gamma}$. Which of the following is the correct option?
 - (A) $\alpha = 7$, $\beta = -1$, $\gamma = -2$

(B) $\alpha = -7$, $\beta = -1$, $\gamma = -2$

(C) $\alpha = 7$, $\beta = -1$, $\gamma = 2$

(D) $\alpha = -7$, $\beta = 1$, $\gamma = -2$

Answer (A)

Sol. $Y = c^{\alpha} h^{\beta} G^{\gamma}$

$$[M^{1}L^{-1}T^{-2}] = [M^{0}L^{1}T^{-1}]^{\alpha} [M^{1}L^{2}T^{-1}]^{\beta} [M^{-1}L^{3}T^{-2}]^{\gamma}$$

$$1 = \beta - \gamma$$

$$-1 = \alpha + 2\beta + 3\gamma$$

$$-2 = -\alpha - \beta - 2\gamma$$

Solving

$$\alpha = 7, \beta = -1, \gamma = -2$$

3. A particle of mass m is moving in the xy-plane such that its velocity at a point (x, y) is given as $\vec{v} = \alpha(y\hat{x} + 2x\hat{y})$, where α is a non-zero constant. What is the force \vec{F} acting on the particle?

(A)
$$\vec{F} = 2m\alpha^2(x\hat{x} + y\hat{y})$$

(B)
$$\vec{F} = m\alpha^2 (y\hat{x} + 2x\hat{y})$$

(C)
$$\vec{F} = 2m\alpha^2(y\hat{x} + x\hat{y})$$

(D)
$$\vec{F} = m\alpha^2(x\hat{x} + 2y\hat{y})$$

Answer (A)

Sol.
$$F = m \frac{d\vec{v}}{dt}$$

$$\vec{v} = \alpha (y\hat{x} + 2x\hat{y})$$

$$\frac{d\vec{v}}{dt} = \alpha \left(\frac{dy}{dt} \, \hat{x} + 2 \frac{dx}{dt} \, \hat{y} \right)$$

$$= \alpha \left(v_y \hat{x} + 2 v_x \hat{y} \right)$$

$$=\alpha\big[2x\alpha\hat{x}+2\alpha y\hat{y}\big]$$

$$= 2\alpha^2 \left[x\hat{x} + y\hat{y} \right]$$

$$\vec{F} = 2m\alpha^2 \left[x\hat{x} + y\hat{y} \right]$$

- **4.** An ideal gas is in thermodynamic equilibrium. The number of degrees of freedom of a molecule of the gas is *n*. The internal energy of one mole of the gas is *U*_n and the speed of sound in the gas is *v*_n. At a fixed temperature and pressure, which of the following is the correct option?
 - (A) $v_3 < v_6$ and $U_3 > U_6$

(B) $v_5 > v_3$ and $U_3 > U_5$

(C) $v_5 > v_7$ and $U_5 < U_7$

(D) $v_6 < v_7$ and $U_6 < U_7$

Answer (C)

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Sol.
$$U_n = \frac{1 \times n \times RT}{2} = \frac{nRT}{2}$$

$$v_n = \sqrt{\frac{\gamma RT}{M}} = \sqrt{\frac{\left(1 + \frac{2}{n}\right)RT}{M}}$$

$$\Rightarrow U_7 > U_5 \text{ and } U_7 > U_6$$

and $v_5 > v_7$

SECTION 2 (Maximum Marks: 12)

- This section contains THREE (03) questions.
- Each question has **FOUR** options (A), (B), (C) and (D). **ONE OR MORE THAN ONE** of these four option(s) is(are) correct answer(s).
- For each question, choose the option(s) corresponding to (all) the correct answer(s).
- Answer to each question will be evaluated according to the following marking scheme:

Full Marks : +4 ONLY if (all) the correct option(s) is(are) chosen;

Partial Marks : +3 If all the four options are correct but **ONLY** three options are chosen;

Partial Marks : +2 If three or more options are correct but ONLY two options are chosen, both of which

are correct;

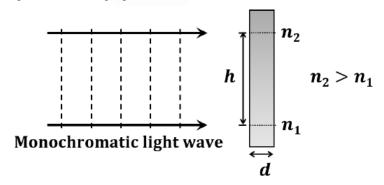
Partial Marks : +1 If two or more options are correct but ONLY one option is chosen and it is a correct

option;

Zero Marks : 0 If unanswered;

Negative Marks : -2 In all other cases.

5. A monochromatic light wave is incident normally on a glass slab of thickness d, as shown in the figure. The refractive index of the slab increases linearly from n_1 to n_2 over the height h. Which of the following statement(s) is(are) true about the light wave emerging out of the slab?



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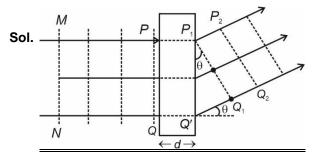


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- (A) It will deflect up by an angle $\tan^{-1} \left[\frac{\left(n_2^2 n_1^2 \right) d}{2h} \right]$
- (B) It will deflect up by an angle $\tan^{-1} \left[\frac{(n_2 n_1)d}{h} \right]$
- (C) It will not deflect
- (D) The deflection angle depends only on $(n_2 n_1)$ and not on the individual values of n_1 and n_2

Answer (D)



Dashed line MN and PQ are incident wave front.

Just before entering the slab PQ is maxima.

 P_1 and Q_1 are the points on other face of slab.

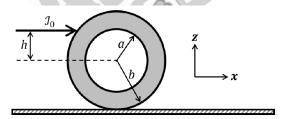
So, P_1 will be at lead of n_2 d w.r.t. P

And Q' will be at lead of n_1d w.r.t. Q

Right of the medium is homogeneous

So to have same path lead as P_1 . Q' must travel to Q_1 , so $n_2d = n_1d + h \sin\theta$

6. An annular disk of mass M, inner radius a and outer radius b is placed on a horizontal surface with coefficient of friction μ , as shown in the figure. At some time, an impulse $J_0\hat{x}$ is applied at a height h above the center of the disk. If $h = h_m$ then the disk rolls without slipping along the x-axis. Which of the following statement(s) is(are) correct?



- (A) For $\mu \neq 0$ and $a \rightarrow 0$, $h_m = b/2$
- (B) For $\mu \neq 0$ and $a \rightarrow b$, $h_m = b$
- (C) For $h = h_m$, the initial angular velocity does **not** depend on the inner radius a
- (D) For $\mu = 0$ and h = 0, the wheel always slides without rolling

Answer (A, B, C, D)

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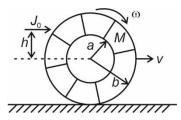
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Sol. $\vec{J} = \overrightarrow{\Delta p}$

$$\Rightarrow J_0 = MV$$
 ...(i)

About



$$\overrightarrow{A}.\overrightarrow{J} = \overrightarrow{\Delta L}$$

$$\Rightarrow J_0 \times h = \frac{M}{2} \left(a^2 + b^2 \right) \omega \qquad ...(A)$$

If
$$h = h_m \Rightarrow \omega = \frac{v}{b}$$

$$\Rightarrow J_0 \times h_m = \left(\frac{m}{2}\right) \left(a^2 + b^2\right) \frac{v}{b} \qquad \dots (ii)$$

Equation (i) ÷ Equation (ii)

$$\Rightarrow h_m = \frac{a^2 + b^2}{2b}$$

If
$$a \to 0 \Rightarrow h_m = \frac{b}{2}$$

If
$$a \rightarrow b \Rightarrow h_m = b$$

If $h = h_m \Rightarrow \omega$ is independent of a (equation A)

If r = 0, $h = 0 \Rightarrow$ always sliding

7. The electric field associated with an electromagnetic wave propagating in a dielectric medium is given by $\vec{E} = 30(2\hat{x} + \hat{y})\sin\left[2\pi\left(5\times10^{14}t - \frac{10^7}{3}z\right)\right] \text{V m}^{-1}.$ Which of the following option(s) is(are) correct? [Given: The speed of light in vacuum, $c = 3 \times 10^8 \text{ ms}^{-1}$]

(A)
$$B_x = -2 \times 10^{-7} \sin \left[2\pi \left(5 \times 10^{14} t - \frac{10^7}{3} z \right) \right] \text{Wb m}^{-2}$$

(B)
$$B_y = 2 \times 10^{-7} \sin \left[2\pi \left(5 \times 10^{14} t - \frac{10^7}{3} z \right) \right] \text{Wb m}^{-2}$$

- (C) The wave is polarized in the xy-plane with polarization angle 30° with respect to the x-axis
- (D) The refractive index of the medium is 2

Answer (A, D)

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Sol. Speed of light in medium is $V = \frac{W}{k}$

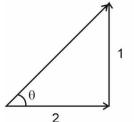
$$V = \frac{3 \times 5 \times 10^{14}}{10^7}$$

$$V = 1.5 \times 10^8$$

Refractive index
$$\mu = \frac{C}{V} = \frac{3 \times 10^8}{1.5 \times 10^8} = 2$$

$$\mu = 2$$

Given
$$\vec{E} = 30(2\hat{x} + \hat{y})\sin\left(2\pi\left(5 \times 10^{14}t - \frac{10^7}{3}2\right)\right)^{1/m}$$



1
$$\tan\theta = \frac{1}{2}$$

$$B_0 = \frac{E_0}{V} = \frac{30\sqrt{5}}{1.5 \times 10^8}$$

Direction of \vec{B}_0 is $(\vec{V} \times \vec{E})$

 $\vec{V} \times \vec{E}$

$$= \hat{k} \times \frac{\left(2\hat{i} + \hat{j}\right)}{\sqrt{5}}$$

$$\left(\frac{-\hat{i}+2\hat{j}}{\sqrt{5}}\right) \text{ put value } \vec{B}_0 = \frac{30\sqrt{5}}{1.5\times10^8} \times \left(\frac{-\hat{i}+2\hat{j}}{\sqrt{5}}\right)$$

$$B_x = -2 \times 10^7$$

Option A, D correct

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SECTION 3 (Maximum Marks: 24)

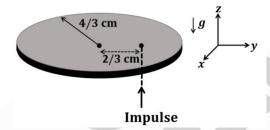
- This section contains SIX (06) questions.
- The answer to each question is a **NON-NEGATIVE INTEGER**.
- For each question, enter the correct integer corresponding to the answer using the mouse and the on-screen virtual numeric keypad in the place designated to enter the answer.
- Answer to each question will be evaluated according to the following marking scheme:

Full Marks : +4 If **ONLY** the correct integer is entered;

Zero Marks : 0 In all other cases.

8. A thin circular coin of mass 5 gm and radius 4/3 cm is initially in a horizontal *xy*-plane. The coin is tossed vertically up (+*z* direction) by applying an impulse of $\sqrt{\frac{\pi}{2} \times 10^{-2}}$ N-s at a distance 2/3 cm from its center. The coin spins about its diameter and moves along the +*z* direction. By the time the coin reaches back to its initial position, it completes *n* rotations. The value of *n* is _____.

[Given: The acceleration due to gravity $g = 10 \text{ ms}^{-2}$]



Answer (30)

Sol. By impulse – momentum theorem :

$$J = MV_{CM}$$

$$\Rightarrow V_{CM} = \frac{J}{M} = \frac{\sqrt{\frac{\pi}{2}}}{100 \times \frac{5}{1000}} = \sqrt{2\pi} \text{ m/s}$$

$$\Rightarrow$$
 Total time of journey $=\frac{2}{g} \times \sqrt{2\pi}$

$$\Rightarrow \Delta t = \frac{\sqrt{2\pi}}{5} s$$

Also, by angular impulse - momentum theorem :

$$J \times \frac{R}{2} = \left\lceil \frac{MR^2}{4} \right\rceil W$$

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$$\Rightarrow W = \frac{J \times \frac{R}{2}}{\frac{MR^2}{4}} = \frac{J}{MR} \times 2$$

$$=\frac{\frac{\sqrt{\frac{\pi}{2}}}{100} \times 2}{\frac{5}{1000} \times \frac{4}{3} \times \frac{1}{100}}$$

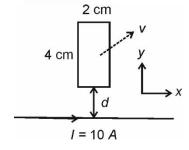
 $2 \times 75\sqrt{2\pi} \text{ rad/s}$

 \Rightarrow Number of rotations $\frac{w.\Delta t}{2\pi}$

$$\frac{2\times75\sqrt{2\pi}\times\frac{\sqrt{2\pi}}{5}}{2\pi}=30$$

- \Rightarrow n = 30
- 9. A rectangular conducting loop of length 4 cm and width 2 cm is in the xy-plane, as shown in the figure. It is being moved away from a thin and long conducting wire along the direction $\frac{\sqrt{3}}{2}\hat{x} + \frac{1}{2}\hat{y}$ with a constant speed v. The wire is carrying a steady current I = 10 A in the positive x-direction. A current of $10 \, \mu$ A flows through the loop when it is at a distance d = 4 cm from the wire. If the resistance of the loop is $0.1 \, \Omega$, then the value of v is _____ ms⁻¹.

[Given: The permeability of free space $\mu_0 = 4\pi \times 10^{-7} \text{ NA}^{-2}$]



Answer (4)

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Sol. The two sides perpendicular to the wire would contribute net zero emf.

For parallel sides:

$$\vec{E} = \vec{B} \times \vec{v}$$

$$=\frac{\mu_0 I}{2\pi x} \times V$$

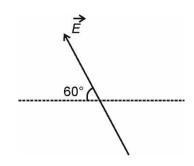
 \Rightarrow Net emf = $(E_1 \cos 60^\circ - E_2 \cos 60^\circ) \times \text{width}$

$$=\frac{1}{2} \times \frac{2}{100} \times \frac{\mu_0 I v}{2\pi} \left[\frac{1}{\frac{4}{100}} - \frac{1}{\frac{8}{100}} \right]$$

$$= \frac{1}{100} \times 10^{-7} \times 2 \times 10 \times v \times 100 \times \frac{1}{8}$$

$$=2.5v\times10^{-7}=i\times R$$

$$\Rightarrow v = \frac{10 \times 10^{-6} \times 0.1}{2.5 \times 10^{-7}} = 4 \text{ m/s}$$



10. A string of length 1 m and mass 2×10^{-5} kg is under tension T. When the string vibrates, two successive harmonics are found to occur at frequencies 750 Hz and 1000 Hz. The value of tension T is _____newton.

Answer (5)

Sol. $I = 1 \text{ m}, m = 2 \times 10^{-5} \text{ kg}, T$: Tension in the string.

: Successive frequencies are being given

:. It is the case of both ends fixed.

Now.

$$f_{n+1} - f_n = 1000 - 750$$

$$\Rightarrow \frac{(n+1)}{2I}\sqrt{\frac{T}{\mu}} - \frac{n}{2I}\sqrt{\frac{T}{\mu}} = 250$$

$$\Rightarrow \frac{1}{2I}\sqrt{\frac{T}{\mu}} = 250$$

$$\Rightarrow \sqrt{\frac{T}{2 \times 10^{-5}}} = 250 \times 2 \times 1$$

$$\Rightarrow \frac{T}{2\times10^{-5}} = 25\times10^{-4}$$

$$\Rightarrow T = 50 \times 10^{-1}$$

$$T = 5 \text{ N}$$

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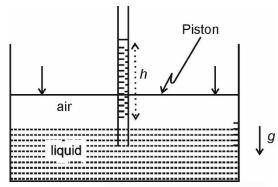
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11. An incompressible liquid is kept in a container having a weightless piston with a hole. A capillary tube of inner radius 0.1 mm is dipped vertically into the liquid through the airtight piston hole, as shown in the figure. The air in the container is isothermally compressed from its original volume V_0 to $\frac{100}{101}V_0$ with the movable piston. Considering air as an ideal gas, the height (h) of the liquid column in the capillary above the liquid level in cm

[Given: Surface tension of the liquid is 0.075 N m⁻¹, atmospheric pressure is 10⁵ N m⁻², acceleration due to gravity (g) is 10 m s⁻², density of the liquid is 10^3 kg m⁻³ and contact angle of capillary surface with the liquid is



Answer (25)

Sol. Let P_f be the air pressure

$$P_0 v_0 = P_f v_f$$

$$P_0 v_0 = P_f \left(\frac{100}{101} \right) v_0$$

$$P_c = 101 \times 10^3 \, \text{Pa}$$

$$P_f = 101 \times 10^3 \text{ Pa}$$
 (: $P_0 = 10^5 \text{ Nm}^{-2}$)

Now, consider the 4 points shown in diagram

$$P_d - P_C = \frac{2T}{R} \qquad (\because P_d = P_0)$$

$$(:: P_d = P_0)$$

$$P_c = P_0 - \frac{2T}{R}$$

Now,

$$P_a = P_b$$

(also,
$$P_a = P_f$$
)

$$P_f = \rho gh + P_c$$

$$101 \times 10^3 = (10^3 \times 10 \times h) + \left(10^5 - \frac{2 \times 0.075}{0.1 \times 10^{-3}}\right)$$

$$h = \frac{1}{4} \,\mathrm{m} = 25 \,\mathrm{cm}$$

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Piston

air





12. In a radioactive decay process, the activity is defined as $A = -\frac{dN}{dt}$, where N(t) is the number of radioactive nuclei at time (t). Two radioactive sources, S_1 and S_2 have same activity at time t = 0. At a later time, the activities of S_1 and S_2 are A_1 and A_2 , respectively. When S_1 and S_2 have just completed their 3^{rd} and 7^{th} half-lives, respectively, the ratio $\frac{A_1}{A_2}$ is ______.

Answer (16)

Sol.
$$A_1 = A_0 e^{-\lambda_1 t_1}$$

also
$$A_2 = A_0 e^{-\lambda_2 t_2}$$

at
$$t_1 = \frac{3\ln 2}{\lambda_1}$$
, $A_1 = A_0 e^{-\lambda_1 \frac{3\ln 2}{\lambda_1}}$

$$= A_0 e^{-3\ln 2}$$

Similarly, at

$$t_2 = \frac{7 \ln 2}{\lambda_2}, \ A_2 = A_0 e^{-\lambda_2 \frac{7 \ln 2}{\lambda_2}}$$

$$= A_0 e^{-7 \ln 2}$$

From (i) and (ii)

$$\frac{A_1}{A_2} = \frac{A_0 e^{-3\ln 2}}{A_0 e^{-7\ln 2}} = \frac{2^{-3}}{2^{-7}} = \frac{1}{2^{-4}} = 2^4 = 16$$

$$\therefore \qquad \boxed{\frac{A_1}{A_2} = 16}$$

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...(ii)





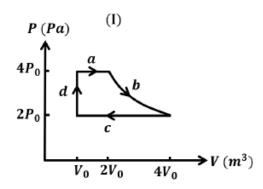


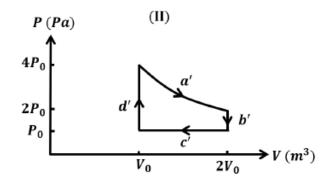
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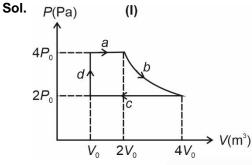
13. One mole of an ideal gas undergoes two different cyclic processes I and II, as shown in the P-V diagrams below. In cycle I, processes a, b, c and d are isobaric, isothermal, isobaric and isochoric, respectively. In cycle II, processes a', b', c' and d'are isothermal, isochoric, isobaric and isochoric, respectively. The total work done during cycle I is $W_{\rm I}$ and that during cycle II is $W_{\rm II}$. The ratio $W_{\rm I}/W_{\rm II}$ is

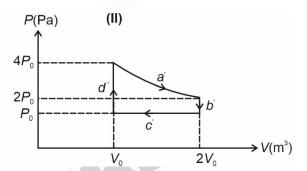




Answer (2)







$$W_{1} = W_{a} + W_{b} + W_{c} + W_{d}$$

$$= 4P_{0} (2V_{0} - V_{0}) + nRT \ln\left(\frac{4V_{0}}{2V_{0}}\right) + 2P_{0} (V_{0} - 4V_{0}) + 0$$

$$= 4P_{0}V_{0} + nR\left(\frac{8P_{0}V_{0}}{nR}\right) \ln 2 - 6P_{0}V_{0}$$

$$= 8P_{0}V_{0} \ln 2 - 2P_{0}V_{0}$$

$$= 8P_{0}V_{0} \ln 2 - 2P_{0}V_{0}$$

$$W_{II} = W_{a}^{'} + W_{b}^{'} + W_{c}^{'} + W_{d}^{'}$$

$$= nRT \ln \left(\frac{2V_{0}}{V_{0}} \right) + 0 + P_{0} \left(V_{0} - 2V_{0} \right) + 0$$

$$= nR \left(\frac{4P_{0}V_{0}}{nR} \right) \ln 2 - P_{0}V_{0}$$

$$= 4P_{0}V_{0} \ln 2 - P_{0}V_{0}$$

$$= \frac{W_{I}}{W_{II}} = \frac{8P_{0}V_{0} \ln 2 - 2P_{0}V_{0}}{4P_{0}V_{0} \ln 2 - P_{0}V_{0}} = 2$$

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SECTION 4 (Maximum Marks: 12)

- This section contains **TWO (02)** paragraphs.
- Based on each paragraph, there are **TWO (02)** questions.
- The answer to each question is a **NUMERICAL VALUE**.
- For each question, enter the correct numerical value of the answer using the mouse and the on-screen virtual numeric keypad in the place designated to enter the answer.
- If the numerical value has more than two decimal places, truncate/round-off the value to TWO decimal places.
- Answer to each question will be evaluated according to the following marking scheme:

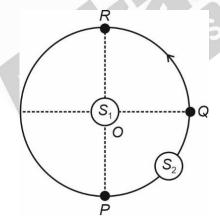
Full Marks : +3 If **ONLY** the correct numerical value is entered in the designated place;

Zero Marks: 0 In all other cases.

PARAGRAPH I

 S_1 and S_2 are two identical sound sources of frequency 656 Hz. The source S_1 is located at O and S_2 moves anticlockwise with a uniform speed $4\sqrt{2}$ ms⁻¹ on a circular path around O, as shown in the figure. There are three points P, Q and R on this path such that P and R are diametrically opposite while Q is equidistant from them. A sound detector is placed at point P. The source S_1 can move along direction OP.

[Given: The speed of sound in air is 324 m s⁻¹]



14. When only S_2 is emitting sound and it is at Q, the frequency of sound measured by the detector in Hz is

_____.

Answer (648)

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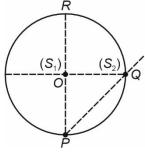




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Sol.



 $f_0 = 656 \text{ Hz}$

Velocity of sound

= 324 m/s.

Velocity of source away from detector

$$V_{\rm S} = 4\sqrt{2} \cos 45^{\circ} = 4 \, \text{m/s}$$

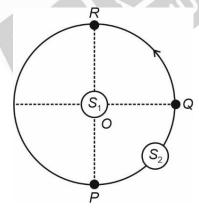
$$\therefore f = \left(\frac{v}{v + v_s}\right) f_0 = \left(\frac{324}{324 + 4}\right) 656$$

f = 648 Hz.

PARAGRAPH I

 S_1 and S_2 are two identical sound sources of frequency 656 Hz. The source S_1 is located at O and S_2 moves anti-clockwise with a uniform speed $4\sqrt{2}$ m s⁻¹ on a circular path around O, as shown in the figure. There are three points P, Q and R on this path such that P and R are diametrically opposite while Q is equidistant from them. A sound detector is placed at point P. The source S_1 can move along direction OP.

[Given: The speed of sound in air is 324 m s⁻¹]



15. Consider both sources emitting sound. When S_2 is at R and S_1 approaches the detector with a speed 4 m s⁻¹, the beat frequency measured by the detector is _____Hz.

Answer (8.20)

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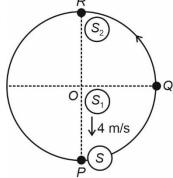




2022



Sol.



$$f_0 = 656 \text{ Hz}$$

$$v = 324 \text{ m/s}$$

Frequency heard due to movement of (S_1)

$$f_1 = \left(\frac{v}{v - u_s}\right) f_0$$

$$f_1 = \frac{324}{320} \times 656$$

And frequency heard due to movement of (S_2)

$$f_2 = 656 \text{ Hz}$$

$$\therefore \text{ Beat frequency } \Delta f = f_1 - f_2 = 656 \left(\frac{324}{320} - 1 \right)$$

$$\Delta f = 8.2$$

PARAGRAPH II

A cylindrical furnace has height (H) and diameter (D) both 1 m. It is maintained at temperature 360 K. The air gets heated inside the furnace at constant pressure P_a and its temperature becomes T = 360 K. The hot air with density ρ rises up a vertical chimney of diameter d = 0.1 m and height h = 9 m above the furnace and exits the chimney (see the figure). As a result, atmospheric air of density $\rho_a = 1.2$ kg m⁻³, pressure P_a and temperature $T_a = 300$ K enters the furnace. Assume air as an ideal gas, neglect the variations in ρ and T inside the chimney and the furnace. Also ignore the viscous effects.

[Given: The acceleration due to gravity $g = 10 \text{ m s}^{-2}$ and $\pi = 3.14$]

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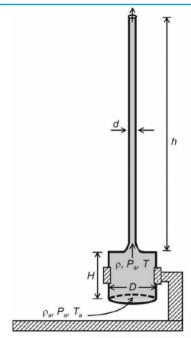






2020





- **16.** Considering the air flow to be streamline, the steady mass flow rate of air exiting the chimney is _____ gm s⁻¹. **Answer (47.10)**
- **Sol.** :: $PM = \rho RT$

And P inside furnace is constant

∴
$$\rho RT$$
 = constant

or
$$\rho T$$
 = constant

$$\therefore \rho_a T_a = \rho T$$

$$1.2(300) = \rho(360)$$

$$\rho = 1 \text{ kg/m}^3$$

Now, applying Bernoulli's theorem at the bottom and the top of chimney

$$P_a + \frac{1}{2}\rho(0)^2 + 0 = (P_a - \rho_a gh) + \frac{1}{2}\rho(v^2) + \rho gh$$

$$V = \sqrt{\frac{2(\rho_a - \rho)gh}{\rho}}$$

$$\Rightarrow$$
 v = 6 m/s

$$\therefore \frac{dm}{dt} \text{ at exit} = \rho v \left(\frac{\pi d^2}{4} \right)$$

$$=\frac{1\times 6\times 3.14\times (0.1)^2}{4}$$

$$= 0.0471 \text{ kg/s}$$

$$= 47.10 \text{ g/s}$$

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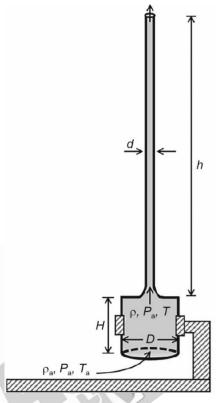
2020



PARAGRAPH II

A cylindrical furnace has height (H) and diameter (D) both 1 m. It is maintained at temperature 360 K. The air gets heated inside the furnace at constant pressure P_a and its temperature becomes T=360 K. The hot air with density ρ rises up a vertical chimney of diameter d=0.1 m and height h=9 m above the furnace and exits the chimney (see the figure). As a result, atmospheric air of density $\rho_a=1.2$ kg m⁻³, pressure P_a and temperature $T_a=300$ K enters the furnace. Assume air as an ideal gas, neglect the variations in ρ and T inside the chimney and the furnace. Also ignore the viscous effects.

[Given: The acceleration due to gravity $g = 10 \text{ m s}^{-2}$ and $\pi = 3.14$]



17. When the chimney is closed using a cap at the top, a pressure difference ΔP develops between the top and the bottom surfaces of the cap. If the changes in the temperature and density of the hot air, due to the stoppage of air flow, are negligible then the value of ΔP is ______ N m⁻².

Answer (30.00)

Sol.
$$P_{\text{out}} = P_a - \rho_a g(H + h)$$

$$P_{\text{in}} = P_a - \rho g h$$

$$\therefore \Delta P = (P_a - \rho g h) - (P_a - \rho_a g (H + h))$$

$$= (\rho_a - \rho)gh + \rho_a gH$$

$$= 0.2 \times 10 \times 9 + 1.2 \times 10 \times 1$$

 $= 30 \text{ Nm}^{-2}$

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2020



PART-II: CHEMISTRY

SECTION 1 (Maximum Marks: 12)

• This section contains FOUR (04) questions.

• Each question has **FOUR** options (A), (B), (C) and (D). **ONLY ONE** of these four options is the correct answer.

For each question, choose the option corresponding to the correct answer.

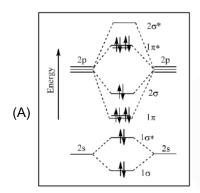
Answer to each question will be evaluated <u>according to the following marking scheme:</u>

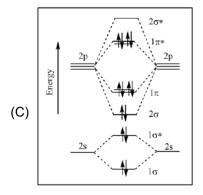
Full Marks : +3 If **ONLY** the correct option is chosen;

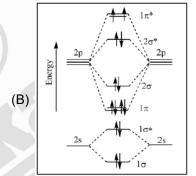
Zero Marks : 0 If none of the options is chosen (i.e. the question is unanswered);

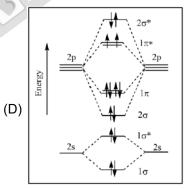
Negative Marks : -1 In all other cases.

1. The correct molecular orbital diagram for F2 molecule in the ground state is









Answer (C)

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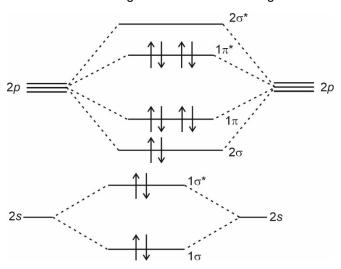








Sol. Molecular orbital diagram for F2 molecule is given below



Hence the correct option is (C).

- 2. Consider the following statements related to colloids.
 - (I) Lyophobic colloids are **not** formed by simple mixing of dispersed phase and dispersion medium.
 - (II) For emulsions, both the dispersed phase and the dispersion medium are liquid.
 - (III) Micelles are produced by dissolving a surfactant in any solvent at any temperature.
 - (IV) Tyndall effect can be observed from a colloidal solution with dispersed phase having the same refractive index as that of the dispersion medium.

The option with the correct set of statements is

- (A) (I) and (II)
- (B) (II) and (III)
- (C) (III) and (IV)
- (D) (II) and (IV)

Answer (A)

Sol. Lyophobic colloids are not formed by simple mixing of dispersed phase and dispersion medium. Their colloidal sols can be prepared only by special methods. Emulsion are colloids of liquid dispersed phase and liquid dispersion medium.

Micelles formation occurs when temperature is above a particular temperature called Kraft temperature T_K and concentration above a particular value know as critical micelle concentration (CMC).

Tyndall effect can be observed when the refractive indices of the dispersed phase and the dispersion medium differ greatly in magnitude.

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3. In the following reactions, P, Q, R and S are the major products.

CI
$$\frac{\text{(i) Mg, dry ether}}{\text{(ii) H}_2\text{O}} \quad \text{P}$$

$$\frac{\text{(i) Mg, dry ether}}{\text{(ii) CO}_2, \text{dry ether}} \quad \text{Q}$$

$$\frac{\text{(ii) Mg, dry ether}}{\text{(iv) NaOH}} \quad \text{Q}$$

$$\frac{\text{(i) Mg, dry ether}}{\text{(iv) NaOH}} \quad \text{R}$$

$$\frac{\text{(i) Mg, dry ether}}{\text{(ii) CH}_3\text{CHO, then H}_2\text{O}} \quad \text{R}$$

$$\frac{\text{(ii) CH}_3\text{CHO, then H}_2\text{O}}{\text{(iii) H}_2/\text{Ni}} \quad \text{S}$$

$$\frac{\text{(i) ethanolic NaCN}}{\text{(iii) CHCI}_3/\text{KOH, }\Delta} \quad \text{S}$$

$$\frac{\text{(iv) LiAIH}_4, \text{then H}_2\text{O}}{\text{(iv) LiAIH}_4, \text{then H}_2\text{O}}$$

The correct statement about P, Q, R and S is

- (A) ${f P}$ is a primary alcohol with four carbons.
- (B) **Q** undergoes Kolbe's electrolysis to give an eight-carbon product.
- (C) R has six carbons and it undergoes Cannizzaro reaction.
- (D) **S** is a primary amine with six carbons.

Answer (B)

Sol.
$$(i)$$
 Mg, dry ether (P) (i) Mg, dry ether (P) (i) MgCl (P) $(P$

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Kolbe electrolysis:

$$\begin{array}{c|c}
2 & & \\
\hline
C & ONa \\
(Q) & O
\end{array}$$
+ 2e⁻ + 2CO₂

$$2 \longrightarrow \bigcirc$$

8 carbon containing product

$$CI \xrightarrow{\text{(i) Mg, dry Ether}} CH_3$$

$$CI \xrightarrow{\text{(ii) CH}_3\text{CHO, H}_2\text{O}} CH_3$$

$$CH_3$$

$$CH_3$$

It undergoes aldol reaction not Cannizzaro.

CI (i) Ethanolic NaCN
$$C = N$$
 H_2/Ni

Secondary amine

Since only Q fulfilled all the given conditions.

Hence, the correct option is (B).

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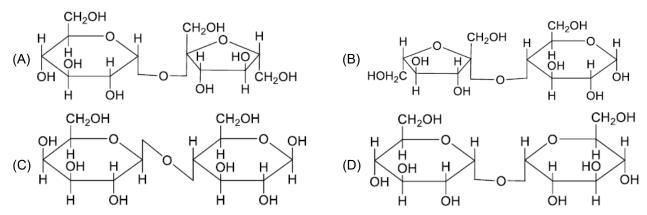


2020

JEE (Advanced)-2023 (Paper-2)



4. A disaccharide **X** cannot be oxidised by bromine water. The acid hydrolysis of **X** leads to a laevorotatory solution. The disaccharide **X** is



Answer (A)

Sol. A and D cannot be oxidised by bromine water as they do not have hemiacetal linkage.

The acid hydrolysis of A leads to a laevorotatory solution.

A is sucrose which is dextrorotatory, on acid hydrolysis gives mixture of α -D-glucose and β -D-fructose, the mixture is laevorotatory.

SECTION 2 (Maximum Marks: 12)

- This section contains THREE (03) questions.
- Each question has **FOUR** options (A), (B), (C) and (D). **ONE OR MORE THAN ONE** of these four option(s) is(are) correct answer(s).
- For each question, choose the option(s) corresponding to (all) the correct answer(s).
- Answer to each question will be evaluated according to the following marking scheme:

Full Marks : +4 ONLY if (all) the correct option(s) is(are) chosen;

Partial Marks : +3 If all the four options are correct but **ONLY** three options are chosen;

Partial Marks : +2 If three or more options are correct but **ONLY** two options are chosen, both of which

are correct;

Partial Marks : +1 If two or more options are correct but ONLY one option is chosen and it is a correct

option;

Zero Marks : 0 If unanswered;

Negative Marks : -2 In all other cases.

5. The complex(es), which can exhibit the type of isomerism shown by $[Pt(NH_3)_2Br_2]$, is(are) $[en = H_2NCH_2CH_2NH_2]$

(A) $[Pt(en)(SCN)_2]$ (B) $[Zn(NH_3)_2Cl_2]$

(C) $[Pt(NH_3)_2Cl_4]$ (D) $[Cr(en)_2(H_2O)(SO_4)]^+$

Answer (C, D)

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Sol. [Pt(NH₃)₂Cl₂] can exhibit G.I.

G.I. can be shown by

 $[Pt(NH_3)_2Cl_4], [Cr(en)_2(H_2O)(SO_4)]^+$

6. Atoms of metals x, y and z form face-centred cubic (fcc) unit cell of edge length L_x, body-centred cubic (bcc) unit cell of edge length L_y, and simple cubic unit cell of edge length L_z, respectively.

If
$$r_z = \frac{\sqrt{3}}{2} r_y$$
; $r_y = \frac{8}{\sqrt{3}} r_x$; $M_z = \frac{3}{2} M_y$ and $M_z = 3M_x$, then the correct statement(s) is(are)

[Given: M_x , M_y , and M_z are molar masses of metals x, y, and z, respectively.

rx, ry, and rz are atomic radii of metals x, y, and z, respectively.]

- (A) Packing efficiency of unit cell of x > Packing efficiency of unit cell of y > Packing efficiency of unit cell of z
- (B) $L_y > L_z$
- (C) $L_x > L_y$
- (D) Density of x > Density of y

Answer (A, B, D)

Sol. Metal x forms FCC (edge length L_x)

Metal y forms BCC (edge length L_y)

Metal z forms SC (edge length Lz)

Given
$$r_z = \frac{\sqrt{3}}{2} r_y$$
 and $r_y = \frac{8}{\sqrt{3}} r_x$

$$\therefore r_z = \frac{\sqrt{3}}{2} \times \frac{8}{\sqrt{3}} r_x = 4r_x$$

$$M_z = \frac{3}{2}M_y \qquad M_z = 3M_x$$

$$M_V = 2M_X$$

Packing efficiency FCC > BCC > SC

Packing efficiency unit cell x > y > z

In FCC unit cell:- atoms along the face diagonals are in contact.

$$\therefore \quad \sqrt{2} \, \mathsf{L}_{\mathsf{x}} = 4 \mathsf{r}_{\mathsf{x}} \Rightarrow \mathsf{L}_{\mathsf{x}} = 2 \sqrt{2} \, \mathsf{r}_{\mathsf{x}}$$

In BCC unit cell: atoms along the body diagonal are

$$\therefore \quad \sqrt{3}\,L_y = 4r_y \ \Rightarrow L_y = \frac{4}{\sqrt{3}}r_y \ \equiv \ \frac{4}{\sqrt{3}} \times \frac{8}{\sqrt{3}}r_x = \frac{32}{3}r_x$$

$$L_y = \frac{32}{3} r_x$$

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In SC unit cell, atoms along the edge are in contact

$$\therefore L_z = 2r_z$$

$$= 2 \times 4r_x = 8r_x$$

$$L_x = 2\sqrt{2}r_x$$

$$L_y = \frac{32}{3}r_x$$

$$L_z = 8r_x$$

$$\therefore L_v > L_z > L_x$$

Density of x (Number of atoms of x per unit cell (z) = 4)

$$\begin{split} d_{x} &= \frac{zM_{x}}{(L_{x})^{3}N_{A}} = \frac{4 \times M_{x}}{(2\sqrt{2}\,r_{x})^{3} \times N_{A}} \\ &= \frac{4M_{x}}{16\sqrt{2}r_{x}^{3}N_{A}} = \frac{M_{x}}{4\sqrt{2}r_{x}^{3}N_{A}} \end{split}$$

Density of y: (Number of atoms of y per unit cell (z) = 2)

$$d_{y} = \frac{zM_{y}}{(L_{y})^{3}N_{A}} = \frac{2 \times 2M_{x}}{\left(\frac{32}{3}r_{x}\right)^{3}N_{A}} = \frac{108M_{x}}{32768r_{x}^{3}N_{A}}$$

- \therefore Density of x > density of y.
- 7. In the following reactions, P, Q, R, and S are the major products.

$$\begin{array}{c|c} & & \underbrace{\begin{array}{c} \text{(i) KMnO}_{4}, \text{ KOH, } \Delta}_{\text{(ii) } \text{ H}_{3}\text{O}^{\oplus}} \text{ P} \end{array}}_{\text{MeOOC}} \\ & & \underbrace{\begin{array}{c} \text{(i) NaOH, H}_{2}\text{O} \\ \text{(ii) H}_{3}\text{O}^{\oplus} \end{array}}_{\text{COOMe}} \text{ Q} \\ \\ & \underbrace{\begin{array}{c} \text{(i) NaOH, H}_{2}\text{O} \\ \text{(ii) H}_{3}\text{O}^{\oplus}, \Delta \\ \hline \text{(ii) H}_{2}\text{CrO}_{4} \end{array}}_{\text{R}} \text{ R} \\ \\ & \underbrace{\begin{array}{c} \text{(i) Mg, dry ether} \\ \text{(iii) CO}_{2}, \text{ then H}_{3}\text{O}^{\oplus} \\ \hline \text{(iii) Ammoniacal AgNO}_{3}, \text{H}_{3}\text{O}^{\oplus} \end{array}}_{\text{R}} \\ \end{array}$$

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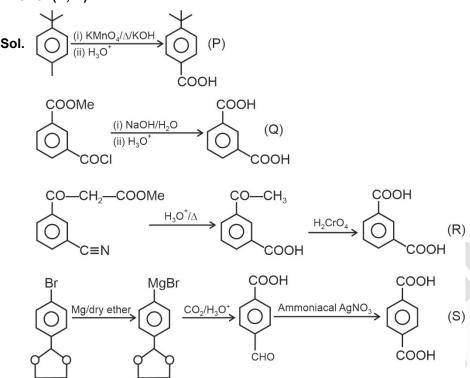
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The correct statement(s) about P, Q, R, and S is(are)

- (A) P and Q are monomers of polymers dacron and glyptal, respectively.
- (B) P, Q, and R are dicarboxylic acids.
- (C) Compounds Q and R are the same.
- (D) R does not undergo aldol condensation and S does not undergo Cannizzaro reaction.

Answer (C, D)



SECTION 3 (Maximum Marks: 24)

- This section contains SIX (06) questions.
- The answer to each question is a NON-NEGATIVE INTEGER.
- For each question, enter the correct integer corresponding to the answer using the mouse and the on-screen virtual numeric keypad in the place designated to enter the answer.
- Answer to each question will be evaluated according to the following marking scheme:

Full Marks : +4 If **ONLY** the correct integer is entered;

Zero Marks : 0 In all other cases.

8. H₂S (5 moles) reacts completely with acidified aqueous potassium permanganate solution. In this reaction, the number of moles of water produced is **x**, and the number of moles of electrons involved is **y**. The value of (**x + y**) is _____.

Answer (18)

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Sol. $KMnO_4 \xrightarrow{H^+} Mn^{+2}$

$$S^{-2} \longrightarrow \overset{0}{S}$$

 \therefore n_{factor} of KMnO₄ = 5

$$n_{factor}$$
 of $S^{-2}(H_2S) = 2$

$$(n_{KMnO_4} \times 5) = (5 \times 2)_{H_2S}$$

$$\left[(\mathsf{GEN})_{\mathsf{KMnO}_4} = (\mathsf{GEP})_{\mathsf{H}_2\mathsf{S}} \right]$$

$$\therefore \quad n_{\mathsf{KMnO_4}} = 2$$

$$\therefore$$
 2 KMnO₄ + 3H₂SO₄ + 5H₂S \rightarrow K₂SO₄ + 2MnSO₄ + 5S + 8H₂O

Number of moles of water produced = '8'

Number of moles of electrons involved = 10

$$\therefore$$
 x = 8, y = 10 \Rightarrow (x + y) = 18

9. Among [I₃]⁺, [SiO₄]⁴⁻, SO₂Cl₂, XeF₂, SF₄, CIF₃, Ni(CO)₄, XeO₂F₂, [PtCl₄]²⁻, XeF₄, and SOCl₂, the total number of species having *sp*³ hybridised central atom is _____.

Answer (5)

Sol.

$$\begin{bmatrix} \bigcirc \bigcirc \bigcirc \\ | \bigcirc | \bigcirc | \end{bmatrix}^{\dagger}$$





$$\begin{array}{c}
CO \\
\downarrow \\
OC \rightarrow Ni \leftarrow C \\
\uparrow \\
CO \\
SO^{3}
\end{array}$$



10. Consider the following molecules : Br₃O₈, F₂O, H₂S₄O₆, H₂S₅O₆, and C₃O₂. Count the number of atoms existing in their zero oxidation state in each molecule. Their sum is ______.

Answer (6)

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2020



Total atom with zero oxidation number state are 6.

11. For He⁺, a transition takes place from the orbit of radius 105.8 pm to the orbit of radius 26.45 pm. The wavelength (in nm) of the emitted photon during the transition is _____,

[Use:

Bohr radius, a = 52.9 pm

Rydberg constant, $R_H = 2.2 \times 10^{-18} J$

Planck's constant, $h = 6.6 \times 10^{-34} Js$

Speed of light, $c = 3 \times 10^8 \text{ ms}^{-1}$]

Answer (30)

Sol.
$$r = 52.9 \times \frac{n^2}{z} pm$$

∴
$$105.8 = \frac{52.9 \times n^2}{2}$$
 ∴ $n_2 = 2$

$$\therefore$$
 $n_2 = 2$

and 26.45 =
$$52.9 \times \frac{n^2}{2}$$
 $\therefore n_1 = 1$

$$\therefore \quad \Delta E = R_H hC \times z^2 \left[\frac{1}{n_1^2} - \frac{1}{n_2^2} \right]$$

$$\frac{hc}{\lambda} = R_H hC \times z^2 \left[\frac{1}{n_1^2} - \frac{1}{n_2^2} \right]$$

$$\frac{6.6 \times 10^{-34} \times 3 \times 10^8}{\lambda} = 2.2 \times 10^{-18} \times 4 \times \left[\frac{1}{1} - \frac{1}{4}\right]$$

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$$\frac{6.6 \times 10^{-34} \times 3 \times 10^8}{\lambda} = 2.2 \times 10^{-18} \times 4 \times \frac{3}{4}$$

- $\lambda = 300 \text{ Å}$
- $\therefore \quad \lambda = 30 \text{ nm}$
- **12.** 50 mL of 0.2 molal urea solution (density = 1.012 g mL⁻¹ at 300 K) is mixed with 250 mL of a solution containing 0.06 g of urea. Both the solutions were prepared in the same solvent. The osmotic pressure (in torr) of the resulting at 300 K is

[Use: Molar mass of urea = 60 g mol⁻¹; gas constant, R = 62 L torr K⁻¹ mol⁻¹;

Assume, $\Delta_{mix} H = 0$, $\Delta_{mix} V = 0$]

Answer (682)

Sol. Mole of urea = 0.2

Weight of urea = $0.2 \times 60 = 12 g$

Weight of solvent = 1000 g

Weight of solution = 1012 g

$$\therefore \text{ Volume of solution} = \frac{1012}{1.012} = 1000 \text{ ml}$$

: 1000 ml solution contain 0.2 mole

$$\therefore \quad 50 \text{ ml solution contain} = \frac{0.2 \times 50}{1000} = 0.01$$

Mole of urea in other solution = $\frac{0.06}{60}$ = 0.001

$$\therefore \quad \text{Concentration of solution} = \frac{0.01 + 0.001}{\frac{300}{1000}}$$
$$= 0.0366$$

$$\pi$$
 = CRT
= 0.0366 × 62 × 300
= 682

13. The reaction of 4-methyloct-1-ene (P, 2.52 g) with HBr in the presence of (C₆H₅CO)₂O₂ gives two isomeric bromides in a 9 : 1 ratio, with a combined yield of 50%. Of these, the entire amount of the primary alkyl bromide was reacted with an appropriate amount of diethylamine followed by treatment with aq. K₂CO₃ to give a non-ionic product S in 100% yield.

The mass (in mg) of **S** obtained is _____.

[Use molar mass (in g mol $^{-1}$): H = 1, C = 12, N = 14, Br = 80]

Answer (1791)

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2020



Sol.
$$CH_3 - CH_2 - CH_2 - CH_2 - CH - CH_2 - CH = CH_2$$

$$CH_3$$
4-methyloct-1-ene
$$W = 2.52 \text{ g}$$

$$CH_3 - CH_2 -$$

SECTION 4 (Maximum Marks: 12)

- This section contains TWO (02) paragraphs.
- Based on each paragraph, there are TWO (02) questions.
- The answer to each question is a NUMERICAL VALUE.
- For each question, enter the correct numerical value of the answer using the mouse and the on-screen virtual numeric keypad in the place designated to enter the answer.
- If the numerical value has more than two decimal places, truncate/round-off the value to TWO decimal places.
- Answer to each question will be evaluated according to the following marking scheme:

Full Marks : +3 If **ONLY** the correct numerical value is entered in the designated place;

Zero Marks : 0 In all other cases.

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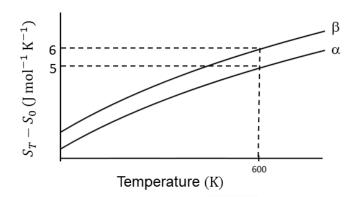




PARAGRAPH I

The entropy versus temperature plot for phases α and β at 1 bar pressure is given.

S_T and S₀ are entropies of the phases at temperatures T and 0 K, respectively.



The transition temperature for α to β phase change is 600 K and $C_{p,\beta}-C_{p,\alpha}=1$ J mol⁻¹ K⁻¹. Assume $(C_{p,\beta}-C_{p,\alpha})$ is independent of temperature in the range of 200 to 700 K. $C_{p,\alpha}$ and $C_{p,\beta}$ are heat capacities of α and β phases, respectively.

14. The value of entropy change, $S_{\beta} - S_{\alpha}$ (in J mol⁻¹ K⁻¹), at 300 K is _

[Use: ln 2 = 0.69

Given: $S_{\beta} - S_{\alpha} = 0$ at 0 K]

Answer (0.31)

Sol.
$$\Delta S_{600} - \Delta S_{300} = \int_{300}^{600} \frac{1 \times (C_{p,\beta} - C_{p,\alpha}) dT}{T}$$

$$= 1 \times 1 \times \left(In \frac{T_2}{T_1} \right) \qquad \begin{pmatrix} T_2 = 600 \text{ K} \\ T_1 = 300 \text{ K} \end{pmatrix}$$

$$1 - \Delta S_{300} = 1 \times 1 \times ln2$$

$$\Delta S_{300}=1-0.69$$

$$\Delta S_{300} = 0.31 \, J \, mol^{-1} K^{-1}$$

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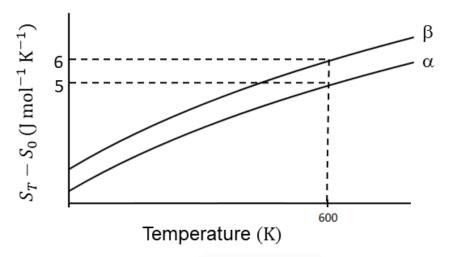




PARAGRAPH I

The entropy versus temperature plot for phases α and β at 1 bar pressure is given.

S_T and S₀ are entropies of the phases at temperatures T and 0 K, respectively.



The transition temperature for α to β phase change is 600 K and $C_{p,\beta}-C_{p,\alpha}$ = 1 J mol⁻¹ K⁻¹. Assume $(C_{p,\beta}-C_{p,\alpha})$ is independent of temperature in the range of 200 to 700 K. $C_{p,\alpha}$ and $C_{p,\beta}$ are heat capacities of α and β phases, respectively.

15. The value of enthalpy change, $H_{\beta} - H_{\alpha}$ (in J mol⁻¹), at 300 K is _____

Answer (300)

Sol.
$$\Delta H_{600} - \Delta H_{300} = 1 \times (C_{p,\beta} - C_{p,\alpha}) (600 - 300)$$

Now, at transition temperature,

$$\Delta H_{600} = T\Delta S_{600}$$

$$= 600 \times (6 - 5)$$

$$= 600 \text{ J mol}^{-1}$$

$$600 - \Delta H_{300} = 1 \times 1 \times 300$$

$$\Delta H_{300} = 600 - 300$$

$$= 300 \text{ J mol}^{-1}$$

Paragraph II

A trinitro compound, 1,3,5-tris-(4-nitrophenyl) benzene, on complete reaction with an excess of Sn/HCl gives a major product, which on treatment with an excess of NaNO₂/HCl at 0°C provides **P** as the product. **P**, upon treatment with excess of H₂O at room temperature, gives the product **Q**. Bromination of **Q** in aqueous medium furnishes the product **R**. The compound **P** upon treatment with an excess of phenol under basic conditions gives the product **S**.

The molar mass difference between compounds $\bf Q$ and $\bf R$ is 474 g mol⁻¹ and between compounds $\bf P$ and $\bf S$ is 172.5 g mol⁻¹.





16. The number of heteroatoms present in one molecule of **R** is ______.

[Use: Molar mass (in g mol⁻¹): H = 1, C = 12, N = 14, O = 16, Br = 80, Cl = 35.5

Atoms other than C and H are considered as heteroatoms]

Answer (9)

Sol.

Number of Heteroatoms in R is 9.

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Paragraph II

A trinitro compound, 1,3,5-tris-(4-nitrophenyl)benzene, on complete reaction with an excess of Sn/HCl gives a major product, which on treatment with an excess of NaNO₂/HCl at 0°C provides **P** as the product. **P**, upon treatment with excess of H₂O at room temperature, gives the product **Q**. Bromination of **Q** in aqueous medium furnishes the product **R**. The compound **P** upon treatment with an excess of phenol under basic conditions gives the product **S**.

The molar mass difference between compounds **Q** and **R** is 474 g mol⁻¹ and between compounds **P** and **S** is 172.5 g mol⁻¹.

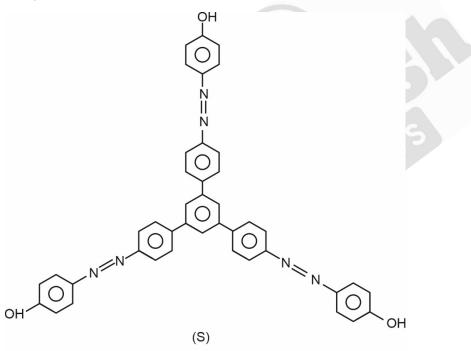
17. The total number of carbon atoms and heteroatoms present in one molecule of S is _____.

[Use: Molar mass (in g mol⁻¹): H = 1, C = 12, N = 14, O = 16, Br = 80, Cl = 35.5

Atoms other than C and H are considered as heteroatoms]

Answer (51)

Sol. Compound S is



Number of Carbon atoms + Number of Heteroatoms = 51

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PART-III: MATHEMATICS

SECTION 1 (Maximum Marks: 12)

- This section contains FOUR (04) questions.
- Each question has FOUR options (A), (B), (C) and (D). ONLY ONE of these four options is the correct answer.
- For each question, choose the option corresponding to the correct answer.
- Answer to each question will be evaluated according to the following marking scheme:

Full Marks : +3 If **ONLY** the correct option is chosen;

Zero Marks : 0 If none of the options is chosen (i.e. the question is unanswered);

Negative Marks : -1 In all other cases.

- 1. Let $f:[1,\infty)\to\mathbb{R}$ be a differentiable function such that $f(1)=\frac{1}{3}$ and $3\int_1^x f(t)dt=x\ f(x)-\frac{x^3}{3}$, $x\in[1,\infty)$. Let e denote the base of the natural logarithm. Then the value of f(e) is
 - (A) $\frac{e^2 + 4}{3}$

(B) $\frac{\log_e 4 + e}{3}$

(C) $\frac{4e^2}{3}$

(D) $\frac{e^2-4}{3}$

Answer (C)

Sol.
$$3\int_{1}^{x} f(t)dt = x \ f(x) - \frac{x^3}{3}$$

$$\Rightarrow$$
 3 $f(x) = f(x) + xf'(x) - x^2$

$$\Rightarrow xf'(x) - 2f(x) = x^2$$

$$\Rightarrow f'(x) - \frac{2}{x}f(x) = x \Rightarrow$$
 (Linear differential equation)

$$\Rightarrow$$
 I.F. = $e^{-\frac{2}{x}dx} = \frac{1}{x^2}$

$$\Rightarrow y\left(\frac{1}{x^2}\right) = \int x \times \frac{1}{x^2} dx = \ln x + C$$

$$\Rightarrow y = x^2(\ln x + C)$$

$$\Rightarrow f(x) = x^2 (\ln x + C)$$

$$\Rightarrow$$
 $f(1) = 1(0 + C) \Rightarrow C = \frac{1}{3}$

$$\Rightarrow f(e) = e^2 \left(\ln e + \frac{1}{3} \right)$$

$$\Rightarrow f(e) = \frac{4e^2}{3}$$

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- Consider an experiment of tossing a coin repeatedly until the outcomes of two consecutive tosses are same. If the probability of a random toss resulting in head is $\frac{1}{3}$, then the probability that the experiment stops with head is
 - (A) $\frac{1}{3}$
 - (B) $\frac{5}{21}$
 - (C) $\frac{4}{21}$
 - (D) $\frac{2}{7}$

Answer (B)

Sol. $P(H) = \frac{1}{3}$

$$P(T)=\frac{2}{3}$$

 $P(E) = P(HH) + P(THH) + P(HTHH) + P(THTHH) + P(HTHTHH) + P(THTHTHH) + \dots$

$$=\frac{1}{3^2}+\frac{2}{3^3}+\frac{2}{3^4}+\frac{4}{3^5}+\frac{4}{3^6}+\frac{8}{3^7}+\frac{8}{3^8}+....$$

$$= \left(\frac{1}{3^2} + \frac{2}{3^4} + \frac{4}{3^6} + \dots \right) + \left(\frac{2}{3^3} + \frac{4}{3^5} + \frac{8}{3^7} + \dots \right)$$

$$P(E) = \frac{1}{7} + \frac{2}{21} = \frac{5}{21}$$

3. For any $y \in \mathbb{R}$, let $\cot^{-1}(y) \in (0, \pi)$ and $\tan^{-1}(y) \in \left(-\frac{\pi}{2}, \frac{\pi}{2}\right)$. Then the sum of all the solutions of the equation

$$\tan^{-1} \left(\frac{6y}{9 - y^2} \right) + \cot^{-1} \left(\frac{9 - y^2}{6y} \right) = \frac{2\pi}{3}$$
 for $0 < |y| < 3$, is equal to

- (A) $2\sqrt{3} 3$
- (B) $3-2\sqrt{3}$
- (C) $4\sqrt{3}-6$
- (D) $6-4\sqrt{3}$

Answer (C)

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Sol.
$$tan^{-1} \left(\frac{6y}{9 - y^2} \right) + \cot^{-1} \left(\frac{9 - y^2}{6y} \right) = \frac{2\pi}{3}$$
 $0 < |y| < 3$ $\Rightarrow y \in (-3, 3) - \{0\}$

Case-I:
$$\frac{6y}{9-y^2} > 0 \Rightarrow y > 0$$

$$\tan^{-1}\left(\frac{6y}{9-y^2}\right) + \tan^{-1}\left(\frac{6y}{9-y^2}\right) = \frac{2\pi}{3}$$

$$\Rightarrow 2\tan^{-1}\left(\frac{6y}{9-y^2}\right) = \frac{2\pi}{3}$$

$$\Rightarrow \tan^{-1}\left(\frac{6y}{9-y^2}\right) = \frac{\pi}{3} \Rightarrow \frac{6y}{9-y^2} = \sqrt{3}$$

$$\Rightarrow 6y = 9\sqrt{3} - \sqrt{3}y^2$$

$$\Rightarrow \sqrt{3}y^2 + 6y - 9\sqrt{3} = 0$$

$$\Rightarrow \sqrt{3}y^2 + 9y - 3y - 9\sqrt{3} = 0$$

$$\Rightarrow \sqrt{3}y(y + 3\sqrt{3}) - 3(y + 3\sqrt{3}) = 0$$

$$\Rightarrow (y + 3\sqrt{3}) - (\sqrt{3}y - 3) = 0$$

$$y \neq -3\sqrt{3} \qquad \therefore y = \sqrt{3} \text{ as } y \in (0, 3)$$

Case-II:
$$\frac{6y}{9-y^2} < 0 \Rightarrow y < 0$$
$$\tan^{-1}\left(\frac{6y}{9-y^2}\right) + \pi + \tan^{-1}\left(\frac{6y}{9-y^2}\right) = \frac{2\pi}{3}$$
$$\Rightarrow 2\tan^{-1}\left(\frac{6y}{9-y^2}\right) = -\frac{\pi}{3} \Rightarrow \tan^{-1}\left(\frac{6y}{9-y^2}\right) = -\frac{\pi}{6}$$

$$\Rightarrow \frac{6y}{9-y^2} = -\frac{1}{\sqrt{3}}$$

$$\Rightarrow 6\sqrt{3}y = -9 + y^2$$

$$\Rightarrow y^2 - 6\sqrt{3}y - 9 = 0$$

$$\Rightarrow y = \frac{6\sqrt{3} \pm \sqrt{108 + 36}}{2} = \frac{6\sqrt{3} \pm 12}{2} = 3\sqrt{3} \pm 6$$

as
$$y \in (-3.0)$$
 : $y = 3\sqrt{3} - 6$

 \therefore Sum of solutions = $\sqrt{3} + (3\sqrt{3} - 6) = 4\sqrt{3} - 6$

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4. Let the position vectors of points *P*, *Q*, *R* and *S* be $\vec{a} = \hat{i} + 2\hat{j} - 5\hat{k}$, $\vec{b} = 3\hat{i} + 6\hat{j} + 3\hat{k}$, $\vec{c} = \frac{17}{5}\hat{i} + \frac{16}{5}\hat{j} + 7\hat{k}$ and

 $\vec{d} = 2\hat{i} + \hat{j} + \hat{k}$, respectively. Then which of the following statements is true?

- (A) The points P, Q, R and S are NOT coplanar
- (B) $\frac{\vec{b} + 2\vec{d}}{3}$ is the position vector of a point which divides *PR* internally in the ratio 5 : 4
- (C) $\frac{\vec{b} + 2\vec{d}}{3}$ is the position vector of a point which divides *PR* externally in the ratio 5 : 4
- (D) The square of magnitude of the vector $\vec{b} \times \vec{d}$ is 95

Answer (B)

Sol.
$$P(1, 2, -5)$$
, $Q(3, 6, 3)$, $R\left(\frac{17}{5}, \frac{16}{5}, 7\right)$, $S(2, 1, 1)$

$$\frac{\vec{b}+2\vec{d}}{3}=\frac{7i+8j+5k}{3}$$

$$\begin{array}{c|cccc}
\lambda & & & 1 \\
\hline
P(1, 2, -5) & & & & & \\
\hline
\left(\frac{7}{3}, \frac{8}{3}, \frac{5}{3}\right) & & & & & \\
R\left(\frac{17}{5}, \frac{16}{5}, 7\right)
\end{array}$$

$$\Rightarrow \frac{17\lambda}{5} + 1 = \frac{7}{3}(\lambda + 1)$$

$$\Rightarrow$$
 51 λ +15 = 35 λ + 35

$$\Rightarrow 16\lambda = 20 \qquad \Rightarrow \lambda = \frac{5}{4}$$

SECTION 2 (Maximum Marks: 12)

- This section contains THREE (03) questions.
- Each question has **FOUR** options (A), (B), (C) and (D). **ONE OR MORE THAN ONE** of these four option(s) is(are) correct answer(s).
- For each question, choose the option(s) corresponding to (all) the correct answer(s).
- Answer to each question will be evaluated according to the following marking scheme:

Full Marks : +4 ONLY if (all) the correct option(s) is(are) chosen;

Partial Marks : +3 If all the four options are correct but **ONLY** three options are chosen;

Partial Marks : +2 If three or more options are correct but **ONLY** two options are chosen, both of which

are correct;

Partial Marks : +1 If two or more options are correct but **ONLY** one option is chosen and it is a correct

option;

Zero Marks : 0 If unanswered; Negative Marks : -2 In all other cases.









JEE (Advanced)-2023 (Paper-2)



- Let $M = (a_{ij}), i, j \in \{1, 2, 3\}$, be the 3 \times 3 matrix such that $a_{ij} = 1$ if j + 1 is divisible by i, otherwise $a_{ij} = 0$. Then which of the following statements is(are) true?
 - (A) M is invertible
 - (B) There exists a nonzero column matrix $\begin{pmatrix} a_1 \\ a_2 \\ a_3 \end{pmatrix}$ such that $M \begin{pmatrix} a_1 \\ a_2 \\ a_3 \end{pmatrix} = \begin{pmatrix} -a_1 \\ -a_2 \\ -a_3 \end{pmatrix}$
 - (C) The set $\{X \in \mathbb{R}^3 : MX = 0\} \neq \{0\}$, where 0 = 0
 - (D) The matrix (M-2I) is invertible, where I is the 3×3 identity matrix

Answer (B, C)

Sol. $M = (a_{ij}), i, j \in \{1, 2, 3\},$

 $a_{ii} = 1$ if j + 1 is divisible by i

otherwise
$$a_{ij} = 0$$

$$M = \begin{bmatrix} 1 & 1 & 1 \\ 1 & 0 & 1 \\ 0 & 1 & 0 \end{bmatrix}$$

$$|M| = 1(-1) - 1(-1) = -1 + 1 = 0$$

M is not invertible

$$\begin{bmatrix} 1 & 1 & 1 \\ 1 & 0 & 1 \\ 0 & 1 & 0 \end{bmatrix} \begin{bmatrix} a_1 \\ a_2 \\ a_3 \end{bmatrix} = \begin{bmatrix} -a_1 \\ -a_2 \\ -a_3 \end{bmatrix}$$

$$\begin{bmatrix} a_1 + a_2 + a_3 \\ a_1 + a_3 \\ a_2 \end{bmatrix} = \begin{bmatrix} -a_1 \\ -a_2 \\ -a_3 \end{bmatrix}$$

$$a_1 + a_2 + a_3 = -a_1$$
 $a_1 + a_3 = -a_2$ $a_2 = -a_3$
 $a_1 + a_2 + a_3 = 0$ $a_2 + a_3 = 0$
 $a_1 = 0$ & $a_2 + a_3 = 0$

⇒ There exist a column matrix (infinite possibilities)

$$\begin{bmatrix} 1 & 1 & 1 \\ 1 & 0 & 1 \\ 0 & 1 & 0 \end{bmatrix} \begin{bmatrix} x \\ y \\ z \end{bmatrix} = \begin{bmatrix} 0 \\ 0 \\ 0 \end{bmatrix} \Rightarrow \begin{aligned} x + y + z &= 0 \\ x + z &= 0 \\ y &= 0 \end{aligned}$$
 Yes it is possible

$$|M-2I| = \begin{bmatrix} -1 & 1 & 1 \\ 1 & -2 & 1 \\ 0 & 1 & -2 \end{bmatrix} = -1(3) - 1(-2-1) = -3 + 3 = 0$$









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Let $f:(0, 1) \to \mathbb{R}$ be the function defined as $f(x) = [4x] \left(x - \frac{1}{4}\right)^2 \left(x - \frac{1}{2}\right)$, where [x] denotes the greatest integer

less than or equal to x. Then which of the following is(are) true?

- (A) The function f is discontinuous exactly at one point in (0,1)
- (B) There is exactly one point in (0, 1) at which the function f is continuous but **NOT** differentiable
- (C) The function f is **NOT** differentiable at more than three points in (0, 1)
- (D) The minimum value of the function f is $-\frac{1}{512}$

Answer (A, B)

Sol. $f:(0,1) \rightarrow \mathbb{R}$

$$f(x) = [4x]\left(x - \frac{1}{4}\right)^2 \left(x - \frac{1}{2}\right) \Rightarrow \text{Critical point} = \frac{1}{4}, \frac{1}{2}, \frac{3}{4}$$

Discontinuity at $x = \frac{3}{4}$

Continuous and differentiable at $x = \frac{1}{4}$

Continuous but non-differentiable at $x = \frac{1}{2}$

LHD
$$\left(\text{at } x = \frac{1}{4}\right)$$

RHD
$$\left(\text{at } x = \frac{1}{4} \right)$$

$$\lim_{h \to 0^+} \frac{0 - 0}{-h} = 0$$

$$\lim_{h \to 0^+} \frac{h^2 \left(-\frac{1}{2} + h \right)}{h} = 0$$

$$LHD\left(\text{at } x = \frac{1}{2}\right)$$

$$RHD\left(\text{at } x = \frac{1}{2}\right)$$

$$\lim_{h \to 0^{+}} \frac{\left(\frac{1}{4} - h\right)^{2} (-h) - 0}{-h} = \frac{1}{16} \qquad \lim_{h \to 0^{+}} \frac{2\left(\frac{1}{4} + h\right)^{2} h - 0}{h} = \frac{1}{8}$$

$$\lim_{h \to 0^{+}} \frac{2\left(\frac{1}{4} + h\right) h - 0}{h} = \frac{1}{8}$$

Minimum –ve value will exist between $\frac{1}{4}$ & $\frac{1}{2}$

$$f(x) = \left(x - \frac{1}{4}\right)^2 \left(x - \frac{1}{2}\right) \qquad \frac{1}{4} \le x \le \frac{1}{2}$$

$$\frac{1}{4} \le x \le \frac{1}{2}$$

$$f'(x) = \left(x - \frac{1}{4}\right)\left(3x - \frac{5}{4}\right)$$
 \Rightarrow minima at $x = \frac{5}{12}$

$$\Rightarrow$$
 minima at $x = \frac{5}{12}$

$$f\left(\frac{5}{12}\right) = \frac{1}{36} \times \frac{-1}{12} = \frac{-1}{432}$$









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Let S be the set of all twice differentiable functions f from \mathbb{R} to \mathbb{R} such that $\frac{d^2f}{dx^2}(x) > 0$ for all $x \in (-1,1)$. For 7.

 $f \in S$, let X_f be the number of points $x \in (-1,1)$ for which f(x) = x. Then which of the following statements is(are) true?

- (A) There exists a function $f \in S$ such that $X_f = 0$
- (B) For every function $f \in S$, we have $X_f \leq 2$
- (C) There exists a function $f \in S$, such that $X_f = 2$
- (D) There does **NOT** exist any function f in S such that $X_f = 1$

Answer (A, B, C)

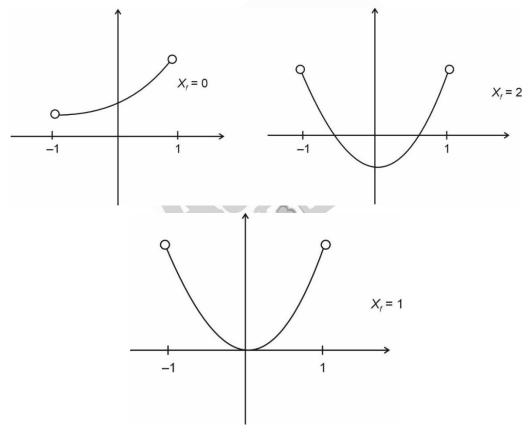
Sol.
$$f''(x) > 0$$
; $f(x) - x = 0$

Number of solutions = ?

Let
$$g(x) = f(x) - x \Rightarrow g'(x) = f'(x) - 1$$

$$g''(x) = f'(x) > 0 \Rightarrow \text{Concave}$$

Possibilities













SECTION 3 (Maximum Marks: 24)

- This section contains SIX (06) questions.
- The answer to each question is a **NON-NEGATIVE INTEGER**.
- For each question, enter the correct integer corresponding to the answer using the mouse and the on-screen virtual numeric keypad in the place designated to enter the answer.
- Answer to each question will be evaluated according to the following marking scheme:

Full Marks : +4 If **ONLY** the correct integer is entered;

Zero Marks : 0 In all other cases.

8. For $x \in \mathbb{R}$, let $\tan^{-1}(x) \in \left(-\frac{\pi}{2}, \frac{\pi}{2}\right)$. Then the minimum value of the function

$$f: \mathbb{R} \to \mathbb{R}$$
 defined by $f(x) = \int_0^{x \tan^{-1} x} \frac{e^{(t-\cos t)}}{1+t^{2023}} dt$ is

Answer (0)

Sol.
$$f'(x) = \frac{e^{\left[x \tan^{-1} x - \cos\left(x \tan^{-1} x\right)\right]}}{1 + \left(x \tan^{-1} x\right)^{2023}} \times \left(\frac{x}{1 + x^2} + \tan^{-1} x\right)$$

$$f'(x) = g(x).h(x)$$

where
$$g(x) = \frac{e^{\left[x \tan^{-1} x - \cos(x \tan^{-1} x)\right]}}{1 + \left(x \tan^{-1} x\right)^{2023}} > 0 \,\forall x$$

and
$$h(x) = \frac{x}{1+x^2} + \tan^{-1} x < 0 \text{ for } x < 0$$

$$= 0$$
 $x = 0$

$$> 0$$
 $x > 0$

 \therefore f(x) has minimum at x = 0

And
$$f(x)_{min} = f(0) = 0$$

9. For $x \in \mathbb{R}$, let y(x) be a solution of the differential equation

$$(x^2-5)\frac{dy}{dx}-2xy = -2x(x^2-5)^2$$
 such that $y(2) = 7$.

Then the maximum value of the function y(x) is

Answer (16)









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Sol.
$$(x^2-5)\frac{dy}{dx} - 2xy = -2x(x^2-5)^2$$

$$\frac{dy}{dx} + \left(\frac{-2x}{x^2 - 5}\right)y = -2x\left(x^2 - 5\right)$$

$$I.F. = \frac{1}{\left|x^2 - 5\right|}$$

Solution of D.E. is
$$y \cdot \frac{1}{|x^2 - 5|} = \int -2x \cdot \frac{x^2 - 5}{|x^2 - 5|} dx \Rightarrow \frac{y}{|x^2 - 5|} = \frac{x^2 - 5}{|x^2 - 5|} (-x^2) + C$$

$$\therefore y(2) = 7 \Rightarrow C = 3$$

$$\Rightarrow$$
 $y = -x^2(x^2 - 5) + 3|x^2 - 5| \Rightarrow y = f(x)$ is even function

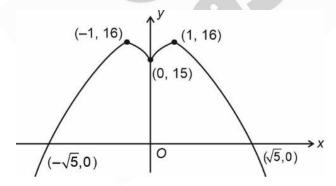
• If
$$0 < x < \sqrt{5}$$
, $y = -x^4 + 5x^2 - 3x^2 + 15 = -x^4 + 2x^2 + 15$

For increasing function $\frac{dy}{dx} > 0 \Rightarrow x < 1$

• If
$$x > \sqrt{5}$$
, $y = -x^4 + 5x^2 + 3x^2 - 15$

For increasing function $\frac{dy}{dx} > 0 \Rightarrow x = \phi$

 \Rightarrow y(x) is increasing over (0, 1)



$$\Rightarrow f(x)_{\text{max}} = 16$$

10. Let *X* be the set of all five digit numbers formed using 1,2,2,2,4,4,0. For example, 22240 is in *X* while 02244 and 44422 are not in *X*. Suppose that each element of *X* has an equal chance of being chosen. Let *p* be the conditional probability that an element chosen at random is a multiple of 20 given that it is a multiple of 5. Then the value of 38*p* is equal to

Answer (31)

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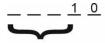
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Sol. Number of five-digit numbers divisible by 5 (____0)

$$2441 \longrightarrow \frac{12}{38}$$

Number of five-digit numbers divisible by 5 but 'not' by 20



$$244 \longrightarrow \underline{3}$$

Number of five-digit numbers divisible by $5 \frac{\text{and}}{20} = 38 - 7 = 31$

$$p=\frac{31}{38}$$

$$38p = 31$$

11. Let A_1 , A_2 , A_3 ,..., A_8 be the vertices of a regular octagon that lie on a circle of radius 2. Let P be a point on the circle and let PA_i denote the distance between the points P and A_i for i = 1, 2, ..., 8. If P varies over the circle, then the maximum value of the product $PA_1 \cdot PA_2 \cdot ... \cdot PA_8$, is

Answer (512)

Sol. A_1 , A_2 , A_3 ,..., A_8 vertices of a regular octagon lying on a circle of radius 2.

Let say,
$$Z = (2) (1)^{1/8}$$

$$\Rightarrow$$
 Z⁸ = 2⁸ × 1

$$\Rightarrow$$
 Z⁸ - 2⁸ = 0

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⇒
$$Z = 2, 2\alpha, 2\alpha^{2}, 2\alpha^{3}, ..., 2\alpha^{7}; \alpha = e^{\frac{i^{2\pi}}{8}}$$

⇒ $Z^{8} - 2^{8} = (Z - 2) (Z - 2\alpha) (Z - 2\alpha^{2}) (Z - 2\alpha^{3}) ... (Z - 2\alpha^{7})$
⇒ $|Z^{6} - 2^{8}| = |Z - 2||Z - 2\alpha|...||Z - 2\alpha^{7}||$
But $|Z^{6} + (-2^{8})| \le |Z|^{8} + 2^{8}$
⇒ $|Z - 2||Z - 2\alpha|...|Z - 2\alpha^{7}| \le |Z|^{8} + 2^{8}$
≤ $2^{8} + 2^{8}$
≤ 2^{9}
⇒ $|Max(PA_{1}.PA_{2}....PA_{8}) = 2^{9}|$

12. Let
$$R = \left\{ \begin{pmatrix} a & 3 & b \\ c & 2 & d \\ 0 & 5 & 0 \end{pmatrix} : a, b, c, d \in \{0, 3, 5, 7, 11, 13, 17, 19\} \right\}$$
. Then the number of invertible matrices in R is

Answer (3780)

Sol.
$$|R| = -5 \begin{vmatrix} a & b \\ c & d \end{vmatrix}$$

|R| can be zero in following cases:

- (i) Two of a, b, c, d are zeroes which can be (a and b), (b and d), (d and c) or (c and a)
- \rightarrow 4 x7² ways = 196
- (ii) Any three of a, b, c, d are zeroes
- \rightarrow $^4C_3 \times 7 = 28$
- (iii) All four of a, b, c, d are zeroes
- $\rightarrow 1$
- (iv) All four of a, b, c, d are non-zero but same number
- $\rightarrow 7$
- (v) When two are alike and 2 other are alike (non-zero) \rightarrow 7C₂ × 2 × 2 = 84

Number of invertible matrices = $8^4 - 196 - 28 - 1 - 7 - 84 = 3780$

13. Let C_1 be the circle of radius 1 with center at the origin. Let C_2 be the circle of radius r with center at the point A = (4,1), where 1 < r < 3. Two distinct common tangents PQ and ST of C_1 and C_2 are drawn. The tangent PQ touches C_1 at P and C_2 at Q. The tangent Q and Q at Q and Q at Q are joined to form a line which meets the Q axis at a point Q at Q at Q and Q at Q and Q are joined to form a line which meets the Q axis at a point Q at Q at Q at Q at Q at Q and Q at Q and Q at Q at Q and Q at Q and Q at Q at

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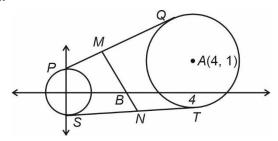


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Answer (2)

Sol.



Let *M* and *N* be midpoints of *PQ* and *ST* respectively.

 \Rightarrow MN is a radical axis of two circles

$$C_1: x^2 + y^2 = 1$$
 ...(i)

$$C_2: (x-4)^2 + (y-1)^2 = r^2$$

$$\Rightarrow x^2 + y^2 - 8x - 2y + 17 - r^2 = 0$$
 ...(ii)

From (i) and (ii);

Equation of MN: $8x + 2y - 18 + r^2 = 0$

$$\Rightarrow$$
 B is on x-axis $\Rightarrow B\left(\frac{18-r^2}{8},0\right)$

$$AB = \sqrt{5}$$

$$\sqrt{\left(\frac{18-r^2}{8}-4\right)^2+1}=\sqrt{5}$$

 \Rightarrow On solving $r^2 = 2$

(By distance formed a)

SECTION 4 (Maximum Marks: 12)

- This section contains TWO (02) paragraphs.
- Based on each paragraph, there are **TWO (02)** questions.
- The answer to each question is a **NUMERICAL VALUE**.
- For each question, enter the correct numerical value of the answer using the mouse and the on-screen virtual numeric keypad in the place designated to enter the answer.
- If the numerical value has more than two decimal places, truncate/round-off the value to TWO decimal places.
- Answer to each question will be evaluated according to the following marking scheme:

Full Marks : +3 If **ONLY** the correct numerical value is entered in the designated place;

Zero Marks : 0 In all other cases.











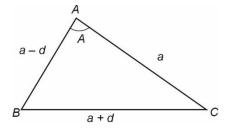
PARAGRAPH "I"

Consider an obtuse angled triangle *ABC* in which the difference between the largest and the smallest angle is $\frac{\pi}{2}$ and whose sides are in arithmetic progression. Suppose that the vertices of this triangle lie on a circle of radius 1.

(There are two questions based on PARAGRAPH "I", the question given below is one of them)

14. Let a be the area of the triangle ABC. Then the value of $(64a)^2$ is

Answer (1008) Sol.



Let sides be a - d, a, a + d

$$A-C=\frac{\pi}{2}$$

$$R = 1$$

Now

$$\frac{a+d}{\sin A} = \frac{a}{\sin B} = \frac{a-d}{\sin C} = 2$$

$$\therefore$$
 A = $\frac{\pi}{2}$ + C

$$\sin A = \sin \left(\frac{\pi}{2} + C \right)$$

$$\sin A = \cos C$$
.

$$\frac{a+d}{2} = \sqrt{1-\sin^2 C}$$

$$\left(\frac{a+d}{2}\right)^2 = 1 - \left(\frac{a-d}{2}\right)^2$$

$$\frac{2\left(a^2+d^2\right)}{4}=1$$

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$$a^2 + d^2 = 2$$
 (1)

Now
$$\cos B = \frac{(a-d)^2 + (a+d)^2 - a^2}{2(a^2 - d^2)}$$

$$\sqrt{1-\sin^2 B} = \frac{2(a^2+d^2)-a^2}{2(a^2-d^2)}$$

$$\sqrt{1 - \frac{a^2}{4}} = \frac{4 - a^2}{2(a^2 - d^2)}$$
 (:: $a^2 + d^2 = 2$)

$$(a^2 - d^2)^2 = 4 - a^2 \qquad(2)$$

From (1) & (2)

$$a^2 = \frac{7}{4}$$
, $d^2 = \frac{1}{4}$

Area of triangle

$$\Delta = \frac{a(a^2 - d^2)}{4}$$

$$\alpha = \frac{\sqrt{7}}{2} \times \frac{6}{4 \times 4}$$

$$(64 \ \alpha)^2 = 1008$$

PARAGRAPH "I"

Consider an obtuse angled triangle ABC in which the difference between the largest and the smallest angle is $\frac{\pi}{2}$ and whose sides are in arithmetic progression. Suppose that the vertices of this triangle lie on a circle of radius 1.

(There are two questions based on PARAGRAPH "I", the question given below is one of them)

15. Then the inradius of the triangle *ABC* is

Answer (0.25)

Sol.
$$r = \frac{\Delta}{S} = \frac{\frac{\sqrt{7}}{2} \times \frac{6}{16}}{\frac{3}{2} \times \frac{\sqrt{7}}{2}} = \frac{4}{16} = \frac{1}{4} = 0.25$$

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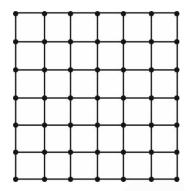


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PARAGRAPH "II"

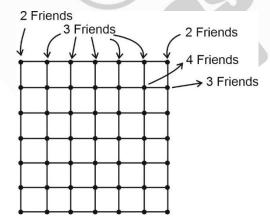
Consider the 6 \times 6 square in the figure. Let A_1 , A_2 ,..., A_{49} be the points of intersections (dots in the picture) in some order. We say that A_i and A_j are friends if they are adjacent along a row or along a column. Assume that each point A_i has an equal chance of being chosen.



(There are two questions based on PARAGRAPH "II", the question given below is one of them)

16. Let p_i be the probability that a randomly chosen point has i many friends, i = 0,1, 2, 3, 4. Let X be a random variable such that for i = 0,1, 2, 3, 4, the probability $P(X = i) = p_i$. Then the value of F(X) is

Answer (24) Sol.



Number of points having 0 friend = 0

Number of points having 1 friend = 0

Number of points having 2 friends = 4

Number of points having 3 friends = $5 \times 4 = 20$









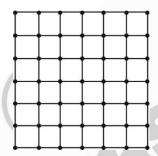


Number of points having 4 friends = 49-24 = 25

$$7 (E(X)) = 7 \left(0 + 0 + \frac{4}{49} \times 2 + \frac{20}{49} \times 3 + \frac{25}{49} \times 4 \right)$$
$$= \left(\frac{100 + 60 + 8}{49} \right)$$
$$= 24$$

PARAGRAPH "II"

Consider the 6x6 square in the figure. Let A_1 , A_2 ,..., A_{49} be the points of intersections (dots in the picture) in some order. We say that A_i and A_j are friends if they are adjacent along a row or along a column. Assume that each point A_i has an equal chance of being chosen.



(There are two questions based on PARAGRAPH "II", the question given below is one of them)

17. Two distinct points are chosen randomly out of the points A_1, A_2, \dots, A_{49} . Let p be the probability that they are friends. Then the value of 7p is

Answer (0.50)

Sol. Number of ways of selecting 2 adjacent dots in I row = 6

Similarly, number of ways of selecting 2 adjacent dots in I column = 6

 \therefore Number of ways of selecting 2 adjacent dots from the matrix = $6 \times 7 + 6 \times 7 = 84$

$$p = \frac{84}{49} = \frac{84 \times 2}{49 \times 48}$$

$$7\rho = \frac{7 \times 84 \times 2}{49 \times 48} = \frac{1}{2}$$

0.50 is answer

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